PROGRAM OF STUDIES

FACULTY: CHEMISTRY
MAIN FIELD OF STUDY: BIOSCIENCES
BRANCH OF SCIENCE: natural sciences

DISCIPLINES: D1 chemical sciences (major discipline)

EDUCATION LEVEL: **second-level studies** (3-semester)

FORM OF STUDIES: full-time studies general academic

LANGUAGE OF STUDY: English

Content:

1. Assumed learning outcomes – attachment no. 1 to the program of studies

2. Program of studies description – attachment no. 2 to the program of studies

Resolution no. ... of the Senate of Wroclaw University of Science and Technology

In effect since **2024/2025**

ASSUMED LEARNING OUTCOMES

FACULTY: Chemistry

MAIN FIELD OF STUDY: BIOSCIENCES

EDUCATION LEVEL: second-level studies

PROFILE: general academic

Location of the main-field-of study:

Branch of science: **natural sciences**Discipline: **chemical sciences**

Explanation of the markings:

Reference to PRK characteristics:

P7U – universal first-degree characteristics corresponding to education at the second-level studies - 7 PRK level

P7S – second degree characteristics corresponding to education at the second-level studies - 7 PRK level

after the underscore:

W – category "knowledge" (extension: G = depth and scope, K = context),

U – category "skills" (extension: W = use of knowledge, K = communication, O = work organization, U = learning),

 \mathbf{K} – category "social competences" (extension: \mathbf{K} = critical assessment, \mathbf{O} = responsibility, \mathbf{R} = professional role),

INŻ – learning outcomes leading to obtaining engineering competences.

Symbols of main field of study learning outcomes at the second cycle of studies for BIOSCIENCES (bs)

before the underscore:

K – directional learning outcomes,

2 – second cycle of studies

A – general academic profile

bs – direction code,

after the underscore:

 \mathbf{W} – knowledge category, \mathbf{U} – skills category, \mathbf{K} – social competence category

		Reference to PRK characteristics								
Main field of study	Description of learning outcomes for the main-field-of study	Universal first	Second degree characteristics typical for qualifications obtained in higher education (S)							
learning outcomes	BIOSCIENCES After completion of studies, the graduate:	degree characteristics (U)	Characteristics for qualifications on 6 / 7* levels of PRK	Characteristics for qualifications on 6 and 7 levels of PRK, enabling acquiring engineering competences						
	KNOWLEDG	E (W)	1							
K2Abs_W01	Possesses in-depth knowledge of mathematics enabling understanding, quantitative description and/or modeling of chemical and/or biotechnological processes	P7U_W	P7S_WG	P7S_WG_INŻ						
K2Abs_W02	Has structured and theoretically based the knowledge of advanced methods used in the identification and characterization of biomolecules and the organization of a research laboratory	P7U_W	P7S_WG							
K2Abs_W03	Knows the principles of formulating hypotheses, building models and formulating theories in the context of the concepts of development of biotechnology and chemistry	P7U_W	P7S_WG							
K2Abs_W04	Knows the concepts and principles of intellectual property protection, patent protection and copyright in the context of the preparation of master thesis	P7U_W	P7S_WK							
K2Abs_W05	Has in-depth knowledge of chemistry necessary to perform chemical analyses, illustrating them with chemical reactions. Recognizes and explains the accompanying physicochemical phenomena	P7U_W	P7S_WG							
K2Abs_W06	Possesses based knowledge enabling the description and characterization of modern instrumental analytical and/or computational methods.	P7U_W	P7S_WG							
K2Abs_W07	Has extended knowledge of the structure of matter and its mathematical description. Explains the laws important in structure identification	P7U_W	P7S_WG							
K2Abs_W08	Knows the postulates of quantum mechanics and the mathematical bases of computational methods of quantum chemistry and molecular mechanics	P7U_W	P7S_WG							
K2Abs_W09	Knows the concepts of molecular mechanics and dynamics.	P7U_W	P7S_WG							

	·			
K2Abs_W10	Has knowledge of mathematics to the extent necessary to design and drug analysis. Knows the theoretical basis of computational methods and computer techniques used in drug design	P7U_W	P7S_WG	P7S_WG_INŻ
K2Abs_W11	Has knowledge of mathematics, numerical and computational methods on the level of molecular modeling and correlating the obtained results with experimental and observational data	P7U_W	P7S_WG	
K2Abs_W12	Knows the theoretical foundations of the functioning of appropriate scientific equipment in the field of drug analysis	P7_UW	P7S_WG	
K2Abs_W13	Knows the physicochemical basis of techniques used in designing of new materials for biotechnology, nanomedicine and pharmacy	P7U_W	P7S_WG	
K2Abs_W14	Knows informatic tools useful in biological research.	P7U_W	P7S_WG	
K2Abs_W15	Has structured, theoretically based knowledge in area biopharmaceuticals	P7U_W	P7S_WG	
K2Abs_W16	Understands the economic, legal and ethical conditions of professional activity	P7U_W	P7S_WK	P7S_WK_INŻ
K2Abs_W17	Has in-depth knowledge of mathematical and IT tools enabling understanding, quantitative description, modeling and design of materials or engineering objects or chemical/biotechnological processes.	P7U_W	P7S_WG	P7S_WG_INŻ
K2Abs_W18	Has in-depth knowledge of exact and natural sciences as well as engineering and technology, allowing the use of methods and concepts necessary to describe materials, chemical or biotechnological processes.	P7U_W	P7S_WG	P7S_WG_INŻ
	SKILLS (U	J)		
K2Abs_U01	Selects and applies mathematical methods in planning and analysis of the experiments	P7U_UW	P7S_UW	P7S_UW_INŻ
K2Abs_U02	Selects and is able to use appropriate methods, techniques and research tools within the appropriate field of study necessary to explain the given problem	P7U_UW	P7S_UW	
K2Abs_U03	Uses computer software to prepare results and statistically analyze the experimental data	P7U_UW	P7S_UW	P7S_UW_INŻ
K2Abs_U04	Uses acquired knowledge in chemistry to related fields of science and scientific disciplines. Demonstrates the ability to work in interdisciplinary teams	P7U_UW	P7S_UO	
K2Abs_U05	Is able to develop research results, critically analyze them and formulate the conclusions	P7U_UW	P7S_UW	
K2Abs_U06	Is able to present the results of own research in the form of a written study	P7U_UW	P7S_UW P7S_UK	
K2Abs_U07	Is able to present the goals and results of scientific work in the form of an oral presentation using modern techniques of communication	P7U_UW	P7S_UW P7S_UK	

K2Abs_U08	Is able to plan experiments and perform basic analyzes using appropriate instrumental equipment and evaluate the results of the experiments. Possesses ability to make theoretical calculations and use available software to simulate the experiment	P7U_UW	P7S_UW	
K2Abs_U09	Is able to use selected the programs that implement quantum chemical computational methods	P7U_UW	P7S_UW	
K2Abs_U10	Is able to apply methods of molecular mechanics and dynamics to solve chemical problems and use algorithms of differentiation, integration and trajectory analysis	P7U_UW	P7S_UW	
K2Abs_U11	Is able to efficiently use modern informatic tools for solving problems in the field of biological and chemical sciences	P7U_UW	P7S_UW	
K2Abs_U12	Is able to use basic methods of quantum chemistry to describe the structure and physicochemical properties of molecules	P7U_UW	P7S_UW	
K2Abs_U13	Is able to write programs or scripts solving numerical problems in the area of computational chemistry and engineering sciences	P7U_UW	P7S_UW	P7S_UW_INŻ
K2Abs_U14	can use a foreign language at level B2+ of the Common European Framework of Reference for Languages	P7U_U	P7U_UK	
K2Abs_U15	is able to independently plan and implement continuous own education in the field of chemical engineering and related sciences; is able to pass on his knowledge to others	P7U_U	P7U_UU	
	SOCIAL COMPETE	NCES (K)		
K2Abs_K01	is ready to critically evaluate his knowledge and received content.	P7U_K	P7S_KK	
K2Abs_K02	understands the need for entrepreneurial thinking and action and is aware of the need to act in the public interest.	P7U_K	P7S_KO	
K2Abs_K03	understands the need to take initiatives, inspire and organize activities for the benefit of the socio-economic environment.	P7U_K	P7S_KO	
K2Abs_K04	is ready to cooperate responsibly in a group, performing role taking into account the needs of the team (and/or social needs)	P7U_K	P7S_KR	
K2Abs_K05	is ready to comply with the principles of professional ethics and respect the law, including copyright.	P7U_K	P7S_KR	
K2Abs_K06	recognizes the importance and understands the non-technical aspects and consequences of scientific and engineering activities, including their impact on the environment, as well as the associated responsibilities.	P7U_K	P7S_KK P7S_KO	
K2Abs_K07	is aware of the social role of a technical university graduate and the need to maintain the ethos of the engineering profession.	P7U_K	P7S_KR	
K2Abs_K08	is ready to recognize the importance of knowledge in solving problems in the field of study and related sciences; recognizes the need to seek expert opinion when difficulties arise in solving problems.	P7U_K	P7S_KK	

DESCRIPTION OF THE PROGRAM OF STUDIES

Main field of study: Biosciences	Profile: general academic
Level of studies: 2 nd level studies (3 sem. magisterskie)	Form of studies: full-time

1. General description

1.1 Number of semesters: 3	1.2 Total number of ECTS points necessary to complete studies at a given level: 90
1.3 Total number of hours: 1095 MDC 1110 BII	1.4 Prerequisites (particularly for second-level studies): are set out in the Order-''The conditions and procedures for recruitment'' in the Technical University of Wroclaw
1.5 Upon completion of studies graduate obtains professional degree of: magister inżynier	1.6 Graduate profile, employability:

The graduate has mastered theory and practice in the use of modern methods of medicinal chemistry, theoretical and computational chemistry and bioinformatics tools enabling: drug design, structural and spectroscopic analysis and insight into the dynamics of processes occurring at the molecular level in macromolecules. The graduate knows the basics of bioinformatics data analysis, machine learning, data mining and big data science, and also has in-depth programming skills in Python and is prepared to work in IT companies, the pharmaceutical industry and in research laboratories.

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned ⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

1.7 Possibility of continuing studies:

Eligibility to apply for admission to a doctoral school, non-degree postgraduate programmes

1.8 Indicate connection with University's mission and its development strategy:

The mission and strategy of Wrocław University of Science and Technology were defined in the document entitled: "Strategy of Wrocław University of Science and Technology 2023-2030". The second-cycle study program in **Biosciences** fits into the key areas of the strategy and the overarching strategic goals both in the area of education, scientific research and cooperation with the environment. It is also consistent with the mission of "creating and transmitting knowledge that responds to new challenges and opportunities

emerging before society, economy and civilization.

The study program is consistent with the strategic goals by:
(1) developing creative skills in the nature of scientific work through an increased number of classes related to the completion of a diploma thesis, (2) a large share (over 50%) of active classes, such as laboratories, exercises, seminars and projects, (3) ensuring a balance between the general and specialized knowledge, (4) providing students with knowledge and skills covering the latest achievements of science and technology in the field of bioinformatics and medicinal chemistry, (5) developing social competences, with particular emphasis on the development of teamwork skills, (6) developing the ability to work using the project method in the IT and (bio-)chemical laboratories.

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- 2. Detailed description
 - 2.1 Total number of learning outcomes in the program of study: W (knowledge) =18, U (skills) = 15, K (competences) = 8 W + U + K = 41
 - 2.2 For the main field of study assigned to more than one discipline the number of learning outcomes assigned to the discipline:

D1 (major) (this number must be greater than half the total number of learning outcomes)

2.3 For the main field of study assigned to more than one discipline - percentage share of the number of ECTS points for each discipline:

D1 100% ECTS points

2.4a. For the general academic profile of the main field of study – the number of ECTS points assigned to the classes related to the University's academic activity in the discipline or disciplines to which the main field of study is assigned – DN (must be greater than 50% of the total number of ECTS points from 1.2

Specialization	Total number of ECTS points
Bioinformatics	73
Medicinal Chemistry	73

- 2.4b. For the practical profile of the main field of study the number of ECTS points assigned to the classes shaping practical skills (must be greater than 50% of the total number of ECTS points from 1.2)
- 2.5 Concise analysis of compliance of the assumed learning outcomes with the needs of the labor market

The needs of the labor market in the field of Biosciences are indirectly presented in this Study Program under the heading Profile of the graduate, employment opportunities. The preparation of graduates listed there is reflected, among others, in the following learning outcomes:

- (1) Is able to conduct scientific experiments, develop and interpret their results and relate them to appropriate theories or scientific hypotheses,
- (2) Using techniques appropriate to the field studied, I can characterize in terms of properties physicochemical and biological biological systems using both quantum chemistry and molecular modeling tools as well as selected experimental methods, (3) Has in-depth knowledge of quantum chemistry, molecular dynamics and molecular modeling, (4) Has in-depth programming skills in Python and knows the operating system Linux,

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(5) Has practical knowledge of various methods of bioinformatic data analysis in contemporary research in the field of exact and natural sciences, (6) Is aware of the importance and understanding of non-technical aspects and effects of scientific and engineering activities, including their impact on the environment, and as well as the associated responsibilities.

The expected learning outcomes meet the current needs of the biotechnology industry, including companies and workplaces dealing with the design, synthesis and development of biologically active substances technology, the pharmaceutical industry, as well as the IT sector.

2.6 The total number of ECTS points that a student must obtain in classes requiring direct participation of academic teachers or other persons conducting classes and students (enter the sum of ECTS points for subjects/ groups of classes marked with the BU¹ code)

Specialization	Total number of ECTS points (BU)
Bioinformatics	50,5
Medicinal Chemistry	49,5

2.7. Total number of ECTS points, which student has to obtain from basic sciences classes

	BII	MDC
Number of ECTS points for obligatory subjects	3	3
Number of ECTS points for optional subjects	4	2
Total number of ECTS points	7	5

2.8. Total number of ECTS points, which student has to obtain from practical classes, including project and laboratory classes (enter total number of ECTS points for subjects/group of classes denoted with code P)

	BII	MDC
Number of ECTS points for obligatory subjects	10	10
Number of ECTS points for optional subjects	53	50
Total number of ECTS points	63	60

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⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes - enter P. For the group of classes - in brackets enter the number of ECTS points assigned to practical courses

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2.9. Minimum number of ECTS points, which student has to obtain doing education blocks offered as part of University-wide classes or other main field of study (enter number of ECTS points for subjects/group of classes denoted with code O)

8 ECTS points

2.10. Total number of ECTS points, which student may obtain doing optional blocks (min. 30% of total number of ECTS points) 70 ECTS points

3. Description of the process leading to learning outcomes acquisition:

Verification and assessment of learning outcomes with reference to courses or groups of courses throughout the entire education cycle takes place in relation to the information contained in the subject cards (syllabuses).

The student acquires knowledge and skills by participating in theoretical and practical classes, which are largely based on the results of scientific research conducted by academic teachers - course tutors conducting classes with students. The basis of teaching and learning process are laboratory, seminar and project courses. Education in the field of studies is conducted in accordance with the principle of increasing the complexity of theoretical and practical tasks set for students. Modern teaching methods are implemented in the teaching practice, thanks to which the students' activity during the classes increases. Theoretical courses in the form of lectures and seminars are supplemented with project and laboratory classes, which include, among others: computer modelling and design, as well as conducting scientific research. The program is complemented by humanities and foreign language courses. The course (study programme) ends with a master thesis preparation and its defence checking the student's theoretical knowledge.

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³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

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4. List of education blocks:

4.1. List of obligatory blocks:

4.1.1 List of general education blocks

4.1.1.1 *Liberal-managerial subjects* block (min. ECTS points):

No.	Subject group of classes code	Name of Subject/group of classes (denote group of courses with symbol GK)	Weekly number of hours					Learning	Number of hours		Number of ECTS points			Form ² of		Subject/group of classes			
			lec	cl	lab	pr	sem	effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
		Total																	

4.1.1.2 Foreign languages block (min. ECTS points):

1	No.	Subject	Name of Subject/group of classes		Weekly number of hours					Number of hours		Number of ECTS points			Form ² of		Subject/group of classes			
		group of classes code	(denote group of courses with symbol GK)	lec	cl	lab	pr	sem	effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU¹ classes	Subject/g roup of crediting courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
			Total																	

4.1.1.3 *Sporting classes* block (0 ECTS points):

N	о.	Subject	Name of Subject/group of classes	Weekly number of hours					Learning	Number of hours		Number of ECTS points		Form ² of		Subject/group of classes				
		group of classesco de	(denote group of courses with symbol GK)	lec	cl	lab	pr	sem	effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
			Total																	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

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4.1.1.4 *Information technologies* block (min. ... ECTS points):

No.	Subject	Name of Subject/group of classes	V	Veekly	numbe	er of ho	ours	Learning		ber of urs	Numbe	er of ECTS	points	Form ² of	2 -	Sı	ıbject/grou	p of classes	
	group of classes code	(denote group of courses with symbol GK)	lec	cl	lab	pr	sem	effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
		Total					•												

Altogether for general education blocks

						0	0		
	Total	number	of hours		Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
lec	cl	lab	pr	sem					

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³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

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4.1.2 List of basic sciences blocks

4.1.2.1 Mathematics block

No.	Subject	Name of Subject/group of classes	V	Veekly	numbe	er of ho	ours	Learning		per of urs	Numbe	er of ECTS	S points	Form ² of		Sı	ıbject/grouj	of classes	
	group of classesco de	(denote group of courses with symbol GK)	lec	cl	lab	pr	sem	effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
		Total																	

4.1.2.2 Physics block

No.	Subject	Name of Subject/group of classes	V	Veekly	numbe	er of ho	ours	Learning		ber of urs	Numbe	er of ECTS	points	Form ² of		Sı	ubject/grou	p of classes	3
	group of classes code	(denote group of courses with symbol GK)	lec	cl	lab	pr	sem	effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
		Total																	

4.1.2.3 Chemistry block BII i MDC

No.	Subject group of classes code	Name of Subject/group		ekly n	umber	of hou	rs			ber of urs	Numbe	er of ECTS	points	Form ² of	_	Sı	ıbject/grou	p of classes	,
		of classess (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1.	W03BSS-SM2001W	Theoretical chemistry	2					K2Abs_W01 K2Abs_W02 K2Abs_W07 K2Abs_W08 K2Abs_W11 K2Abs_W17 K2Abs_K01	30	75	3	3	1,3	T/Z	Е		DN		PD
		Total	2						30	75	3	3	1,3		1				

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

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Altogether for basic sciences blocks:

	Т	otal nu	ımber	of hou	rs	Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
	w	ć	1	p	s					
BII	2					30	120	3	3	1,3
MDC	2					30	120	3	3	1,3

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem) ⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

4.1.3 List of the main field of study blocks

4.1.3.1 Obligatory main field of study blocks

No.	Subject group of classes code	Name of Subject group of			numbe	er of ho	ours		Numl		Numbe	er of ECTS	points	Form ² of		Sı	ıbject/grouj	of classes	
		classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2001L	Theoretical chemistry			2			K2Abs_U03 K2Abs_U04 K2Abs_U05 K2Abs_U09 K2Abs_U12 K2Abs_K08	30	50	2	2	1,4	Т	Z		DN	P	K
2	W03BSS-SM2001C	Theoretical chemistry		1				K2Abs_U02 K2Abs_U12	15	50	2	2	0,7	T/Z	Z		DN	P	K
3	W03BSS-SM2006W	Rational drug design	2					K2Abs_W02 K2Abs_W05 K2Abs_W06 K2Abs_W10 K2Abs_W12 K2Abs_W13 K2Abs_W14 K2Abs_W15 K2Abs_W18 K2Abs_K01 K2Abs_K01	30	75	3	3	1,3	T/Z	Z		DN		K
4	W03BSS-SM2007W	Molecular modeling	1					K2Abs_W09 K2Abs_W11 K2Abs_W14 K2Abs_W17	15	50	2	2	0,65	T/Z	E		DN		K
5	W03BSS-SM2007L	Molecular modeling.			2			K2Abs_U13 K2Abs_U10 K2Abs_U05	30	50	2	2	1,4	Т	Z		DN	Р	K
6	W03BSS-SM2007S	Molecular modeling					1	K2Abs_U07	15	25	1	1	0,7	T/Z	Z		DN	Р	K
7	W03BSS-SM2008L	Retrieval of scientific and technical information			1			K2Abs_U05 K2Abs_K05	15	25	1		0,7	Т	Z			Р	K

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

8	W03BSS-SM2013W	Machine learning for chemistry	2				K2Abs_W01	30	50	2		1,3	T/Z	E			K
		and biology					K2Abs_W14										
							K2Abs_W08										
							K2Abs_W09										
							K2Abs_K01										
9	W03BSS-SM2013L	Machine learning for chemistry			2		K2Abs_U02	30	50	2		1,4	T	Z		P	K
		and biology					K2Abs_U11										
							K2Abs_K02										
							K2Abs_K04										
							K2Abs_K05										
		Total	5	1	7	1		210	425	17	12	9,55		2		10	

Altogether (for main field of study blocks):

	Т	otal nu	ımber	of hou	rs	Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
	w	ć	1	p	S					
BII MDC	5	1	7		1	210	425	17	12	9,55

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

 $^{^3}$ Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned ⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

4.2 List of optional blocks

4.2.1 List of general education blocks

4.2.1.1 Liberal-managerial subjects blocks (min. 5 ECTS points):

No.	Subject group of classes	Name of Subject group of	We	ekly n	umber	of hou	rs		Num ho	ber of urs	Numbe	er of ECTS	points	Form ² of		Sı	ıbject/grouj	of classes	
	code	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03-SM2002BH	Managerial course I	1					K2Abt_W16 K2Abs_K02 K2Abs_K03 K2Abs_K07	15	60	2		0,65	T/Z	Z	0			КО
2	W03-SM2001BH	Managerial course II	2					K2Abt_W16 K2Abs_K02 K2Abs_K03 K2Abs_K07	30	90	3		1,3	T/Z	Z	0			КО
	•	Total	3						45	150	5		1,95						

4.2.1.2 Foreign languages block (min. 3 ECTS points):

No.	Subject group of classes	Name of Subject group of classes			numbei			,		ber of urs	Numb	er of ECT	S points	Form ² of	Way ³	5	Subject/gro	up of classe	es
	code	(denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU¹ classes	Subject/g roup of courses	of crediti ng	Univers ity- wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	SJO-SM0004	Foreign language I		1				K2Abs_U14 K2Abs_U15 K2Abs_K01 K2Abs_K04	15	30	1		0,6	T/Z	Z	O		P	КО
2	SJO-SM0003	Foreign language II		3				K2Abs_U14 K2Abs_U15 K2Abs_K01 K2Abs_K04	45	60	2		1,8	T/Z	Z	O		P	КО
		Total		4					60	90	3		2,4					3	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Altogether for general education blocks:

			1 1	wsc	UIICI	TOI SCIN	ciui cuu	ation or	ociso.	
	Т	otal nu	umber	of hou	rs	Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
	w	ć	1	p	s					
BII	3	4				105	240	8		4,35
MDC	3	4				105	240	8		4,35

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses ⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

4.2.2 List of basic sciences blocks

4.2.2.1 *Mathematics* block

BII

No.	Subject group of classes code	Name of Subject group of	W	Weekly number of hours					per of urs	Numbe	er of ECTS	points	Form ² of		Sı	ıbject/grou	of classes		
		classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2002W	Molecular dynamics	2					K2Abs_W03 K2Abs_W07 K2Abs_W09 K2Abs_W11 K2Abs_W18 K2Abs_K01	30	100	4	4	1,3	T/Z	Z		DN		PD
		Total	2						30	100	4	4	1,3						

MDC

No.	Subject group of classes	Name of Subject group of	W	eekly 1	numbe	r of ho	ırs		Num ho	ber of urs	Numbe	er of ECTS	points	Form ² of		Sı	ubject/group	p of classes	
	code	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2017C	Introductory statistics		1				K2Abs_U03	15	50	2		0,7	T/Z	Z			P	PD
		Total		1					15	50	2		0,7					2	

4.2.2.2 *Physics* block (min. ECTS points):

No	. Subject	Name of Subject group of classes	V	Veekly	numb	er of ho	ours	Learning		ber of urs	Numbe	er of ECTS	points	Form ² of		Sı	ubject/grou	p of classes	
	group of classes code	(denote group of courses with symbol GK)	lec	cl	lab	pr	sem	effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
		Total			·														

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

4.2.2.3 *Chemistry* block (min. ECTS points):

No.	Subject	Name of Subject group of classes	V	Weekly number of hours				Learning		oer of urs	Numbe	er of ECTS	S points	Form ² of		Sı	ıbject/grou	p of classes	
	group of classes code	(denote group of courses with symbol $\mathbf{G}\mathbf{K}$)	lec	cl	lab	pr	sem	effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
		Total	·				•												

Altogether for basic sciences blocks:

	То	otal nu	mber o	f hours	3	Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
	lec	cl	lab	pr	sem					
BII	2	cl lab pr sem			30	100	4	4	1,3	
MDC		1				15	50	2	0	0,7

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem) ⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses ⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

4.2.3 List of main-field of study blocks

4.2.3.1 *Diploma profile* block (min. 29. ECTS points):

No.	Subject							.5).	Num	ber of	Nun	nber of E	CTS			,	7.1: //	C 1	
	group of classes code	Name of Subject/group of classes	W	eekly i	numbei	of ho	ırs		ho	urs		points		Form ² of			Subject/gro	up of classe	es
		(denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	Subject/g roup of courses	Way ³ of crediting	Univers ity- wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03W03-SM2053S	Graduation proseminar					1	K2Abs_U08 K2Abs_U14 K2Abs_K01 K2Abs_K07	15	25	1	1	0,7	T/Z	Z		DN	P	K
2	W03W03-SM2054D	Graduate laboratory I			4			K2Abs_U02 K2Abs_U05 K2Abs_U14 K2Abs_K01 K2Abs_K05 K2Abs_K07	60	150	6	6	3	Т	Z		DN	P	К
3	W03W03-SM2055D	Graduate laboratory II			14			K2Abs_U02 K2Abs_U05 K2Abs_U14 K2Abs_U06 K2Abs_K01 K2Abs_K05 K2Abs_K07	210	500	20	20	9,5	T	Z		DN	P	К
4	W03W03-SM2056S	Graduation seminar					1	K2Abs_U05 K2Abs_U07 K2Abs_K01 K2Abs_K06 K2Abs_K07 K2Abs_K08	15	50	2	2	0,7	T/Z	Z		DN	P	К
•	•	Total			18		2		300	725	29	29	13,9					29	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

4.2.3.2 Optional courses block

No.	Subject group of classes	Name of Subject/group of classes	W	eekly 1	numbe	r of ho	urs	Learning effect		ber of urs	Numl	oer of ECT	S points	Form ² of Subjec	Wav ³ of	S	ubject/group	of classes	
	code	(denote group of courses with symbol GK)	lec	cl	lab	pr	sem	symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	t/group of course s	crediting	Universi ty-wide ⁴	Concerning scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM20BW	Elective course*	2					K2Abt_W02 K2Abt_K01	30	50	2		1,3	T/Z	Z				K
		Total																	

List of elective course

No.	Subject group of classes	Name of Subject group of	We	Weekly number of hours				Numl ho	ber of urs	Numbe	er of ECTS	points	Form ² of		Sı	ıbject/grou	p of classes	1	
	code	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU¹ classes	Subject/g roup of courses	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2101w	Medicinal and biological chemistry	2						30	50	2		1,3	T/Z	Z				K
2	W03BSS-SM2102w	Methodology of experimental research	2						30	50	2		1,3	T/Z	Z				K
3	W03BSS-SM2103w	Bioprocess project	2						30	50	2		1,3	T/Z	Z				K
4	W03BSS-SM2104w	Advanced polymers for chemical and medical applications	2						30	50	2		1,3	T/Z	Z				K

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Altogether for blocks:

		1.	11108	CLIIC	1 101	DIOCIS	•			
	То	otal nu	mber o	f hours		Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
	lec	cl	lab	pr	sem					
BII	2		18		2	330	775	31	29	15,2
MDC	2		18		2	330	775	31	29	15,2

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

 $^{^3}$ Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

4.2.4 List of specialization blocks

4.2.4.1 Specialization subjects blocks

BII Bioinformatics (min. 27 ECTS points):

No.	Subject group of classes code	Name of Subject/group of	W	eekly r	numbei	of ho	urs		Numl ho		Numbe	er of ECT	S points	Form ² of Subjec		S	Subject/gro	up of classe	es
		courses (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	t/group of course s	Way ³ of crediting	Univers ity- wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2002L	Molecular dynamics			2			K2Abs_U02 K2Abs_U03 K2Abs_U06 K2Abs_U10 K2Abs_K08	30	50	2	2	1,4	Т	Z		DN	P	S
2	W03BSS-SM2004W	Bioinformatics	2					K2Abs_W14	30	75	3	3	1,3	T/Z	E		DN		S
3	W03BSS-SM2004L	Bioinformatics.			2			K2Abs_U11	30	50	2	2	1,4	Т	Z		DN	P	S
4	W03BSS-SM2003L	Networks and workstations with UNIX system			2			K2Abs_U13	30	50	2		1,4	Т	Z			P	S
5	W03BSS-SM2005L	Applied informatics			4			K2Abs_U13 K2Abs_U11 K2Abs_K08	60	100	4	4	2,8	T	Z		DN	Р	S
6	W03BSS-SM2010P	Advanced bioinformatics				3		K2Abs_K04 K2Abs_U11 K2Abs_U13	45	75	3	3	2,25		Z		DN	Р	S
7	W03BSS-SM2011W	Bionanotechnology	2					K2Abs_W07 K2Abs_W09 K2Abs_W11 K2Abs_W13 K2Abs_W18	30	50	2	2	1,3	T/Z	Е		DN		S
8	W03BSS-SM2011S	Bionanotechnology.					1	K2Abs_U07 K2Abs_K07	15	25	1	1	0,7	T/Z	Z		DN	P	S
9	W03BSS-SM2012L	Advanced programming and numerical methods			3			K2Abs_W17 K2Abs_K04	45	75	3	3	2,1	Т	Z		DN	P	S

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

							K2Abs_U11 K2Abs_U13										
10	W03BSS-SM2009L	Data mining		1			K2Abs_U02	15	25	1	1	0,7	Т	Z	DN	P	S
							K2Abs_K05										
11	W03BSS-SM2014W	Computational genomics	1				K2Abs_W02	15	25	1	1	0,65	T/Z	E	DN		S
							K2Abs_W14										
							K2Abs_W17										
12	W03BSS-SM2014L	Computational genomics.		1			K2Abs_U01	15	25	1	1	0,7	T	Z	DN	P	S
							K2Abs_K07										
13	W03BSS-SM2015L	Molecular engineering in		3			K2Abs_W06	45	50	2	2	2,1	T	Z	DN	P	S
		genomic analyses					K2Abs_U02										
							K2Abs_U03										
							K2Abs_U05										
							K2Abs_U08										
		Total	5	18	3	1		405	675	27	25	18,8		3		21	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned ⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

MDC Medicinal chemistry (min 29 ECTS points)

No.	Subject group of classes code	Name of Subject/group of courses (denote group of courses with symbol GK)					Learning effect symbol		ber of ours	Nun	nber of E points	CTS	Form ² of Subject/g	Way ³ of		, c	oup of class	es	
			lec	cl	lab	pr	sem		ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	roup of courses	crediting	Unive rsity- wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1.	W03BSS-SM2020W	Spectroscopic methods in medicinal chemistry	2					K2Abs_W02 K2Abs_W06 K2Abs_W13	30	50	2	2	1,3	T/Z	Е		DN		S
2.	W03BSS-SM2020L	Spectroscopic methods in medicinal chemistry			2			K2Abs_U02 K2Abs_U08 K2Abs_K08	30	50	2	2	1,4	Т	Z		DN	P	S
3.	W03BSS-SM2021W	Metabolomics	1					K2Abs_W06 K2Abs_W07 K2Abs_K01	15	50	2	2	0,65	T/Z	Z		DN		S
4.	W03BSS-SM2021L	Metabolomics			2			K2Abs_W14 K2Abs_U03 K2Abs_K06	30	50	2	2	1,4	Т	Z		DN	Р	S
5.	W03BSS-SM2018W	Crystallography and structure of solids	2					K2Abs_W02 K2Abs_W06 K2Abs_W07 K2Abs_W18 K2Abs_K01	30	50	2	2	1,3	T/Z	Z		DN		S
6.	W03BSS-SM2018C	Crystallography and structure of solids		1				K2Abs_U01 K2Abs_U02	15	25	1	1	0,7	T/Z	Z		DN	P	S
7.	W03BSS-SM2019W	Analytical methods in drug design and technology	1					K2Abs_W02 K2Abs_W06	15	50	2	2	0,65	T/Z	Z		DN		S
8.	W03BSS-SM2019L	Analytical methods in drug design and technology.			2			K2Abs_U01 K2Abs_U03 K2Abs_U08	30	50	2	2	1,4	Т	Z		DN	Р	S
9.	W03BSS-SM2016L	Isolation and identification of bioproducts			2			K2Abs_U02 K2Abs_U03 K2Abs_W05 K2Abs_K08	30	50	2	2	1,4	T	Z		DN	P	S
10.	W03BSS-SM2022W	Medicinal natural products	1					K2Abs_W03 K2Abs_W12 K2Abs_W15	15	50	2	2	0,65	T/Z	E		DN		S

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

11.	W03BSS-SM2022L	Medicinal natural products.			2		K2Abs_U02	30	50	2	2	1,4	T	Z	DN	P	S
							K2Abs_K04										
12.	W03BSS-SM2023W	Modern pharmaceuticals and	2				K2Abs_W02	30	50	2	2	1,3	T/Z	E	DN		S
		biopharmaceuticals					K2Abs_W07										
							K2Abs_W12										
							K2Abs_W06										
							K2Abs_K01										
13.	W03BSS-SM2023L	Modern pharmaceuticals and			2		K2Abs_U02	30	50	2	2	1,4	T	Z	DN	P	S
		biopharmaceuticals					K2Abs_U05										
							K2Abs_K05										
							K2Abs_K08										
14.	W03BSS-SM2024L	Multistep organic synthesis			4		K2Abs_U02	60	75	3	3	2,8	T	Z	DN	P	S
							K2Abs_U04										
							K2Abs_U05										
							K2Abs_K08										
15.	W03BSS-SM2025W	Inorganic drugs	1			•	K2Abs_W02	15	25	1	1	0,65	T/Z	Z	DN		S
							K2Abs_W03										
	·	Total	10	1	16			405	725	29	29	18,4		3		16	

Altogether for specialization blocks:

			7 1110	Scui	CI 101	specializat.	ion blocks.				
		Total	number	of hou	rs	Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹	
	lec	cl	lab	pr	sem						
BII	5		18	3	1	405	675	27	25	18,8	
MDC	10	1	16			405	725	29	29	18,4	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

4.3 Training block - concerning principles of training crediting - attachment no.

Not applicable

Opinion of the Advisory Faculty Council concerning the rules of crediting training block

Name of training						
Number of ECTS points	Number of ECTS	S points for BU¹ classes	Training crediting mode	Code		
Training duration	Training objective					

4.4 "Diploma dissertation" block

problem.

Type of diploma dissertation	Licencjat / inżynier / magister / magister inżynier*								
Number of diploma dissertation semesters	Number of ECTS points	Code							
3	29	W03W03-SM2053S							
	·	W03W03-SM2054D							
		W03W03-SM2055D							
		W03W03-SM2056S							
Character	of diploma dissertation								
Thesis of the second cycle (master) should have traits of scientific results of original research or technical and technological solutions the knowledge and skills of the author, including but not limited literature and other sources of knowledge; (3)The ability to plan	a, and its presentation in the form of written work st to:(1)The ability to formulate objectives and res	hould include the results and show earch questions; (2) Ability to use							

(4)Ability to correctly interpret the results; (5)Ability to use precise and clear language and the proper matching of the images presented to illustrate the

Number of BU¹ ECTS points 13,9

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem) ⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

5. Ways of verifying assumed learning outcomes

Form of classes	Ways of verifying assumed learning outcomes
lecture	e.g. examination, progress/final test
class	e.g. progress/final test
laboratory	e.g. pretest, report from laboratory
project	e.g. project defence
seminar	e.g. participation in discussion, topic presentation, essay
diploma dissertation	prepared diploma dissertation

6. Range of diploma examination

BII

- 1. General aspects of biotechnology.
- 2. Drug design methods
- 3. Bioinformatics selected issues

MDC

- 1. General aspects of medicinal chemistry.
- 2. Methods of drug design and synthesis
- 3. Biological chemistry selected issues

7. Requirements concerning deadlines for crediting subject/groups of subject for all courses in particular blocks

Each course should be passed in accordance with the study plan. If it is necessary to repeat a course, it should be completed in the next semester in which it is offered.

*T/Z Remote form of classes requires Dean's approved, but cannot exceed 75 % of ECTS points.

T/Z option is accepted only for lectures, exercises and seminars

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²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

8. Plan of studies (attachment no. 4) Approved by faculty student government legislative body: Date name and surname, signature of student representative Date Dean's signature

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes ²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

PLAN OF STUDIES

FACULTY: Chemistry

MAIN FIELD OF STUDY: BIOSCIENCES

EDUCATION LEVEL: second-level studies (3-semester)

FORM OF STUDIES: full-time studies

PROFILE: general academic

SPECIALIZATION: Bioinformatics

LANGUAGE OF STUDY: English

In effect since **2024/2025**

Plan of studies structure (optionally)

1) in ECTS point layout (space for scheme of plan)

2) in hourly layout (space for scheme of plan)

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

2nd LEVEL STUDIES (MASTER LEVEL STUDIES) (3 sem)

Field of study: BIOSCIENCES

Specialization: Bioinformatics

Specialization subjects
Obligatory subjects
Optional subjects

Sem.	I	П	III
Godz.	24h / 30ECTS / 2E	26h / 30ECTS / 2E	24h/ 30 ECTS /2E
26		Rational drug design	
25		2w (3ECTS)	
24	Elective course	Molecular modeling E	Machine learning for chemistry and biology
23	2w(2ECTS)	1w+2l+1s (2+2+1) ECTS	2w+2l E (2+2) ECTS
22	Theoretical chemistry E 2w +1c +2l	(2-2-1) Be 15	(2+2) E016
21	(3+2+2) ECTS		
20		Retrieval of scientific and technical information 11 (1 ECTS)	Computational genomics E 1w+11
19		Data mining 11 (1ECTS)	(1+1) ECTS
18		Advanced bioinformatics	Molecular engineering in genomic analyses
	Molecular dynamics	3p (3 ECTS)	31 (2 ECTS)
17	2w +2l (4 + 2) ECTS		
16	(4 + 2) EC1S	Bionanotechnology E	Graduate laboratory II
15		Bionanotechnology E 2w + 1s	141 (20 ECTS)
14	Networks and workstations with UNIX system	(2+1) ECTS	
13	21 (2 ECTS)	Advanced programming and numerical methods	
11	Bioinformatics E	31 (3 ECTS)	
10	2w + 2l		
9	(3+2) ECTS	Managerial course II	
8		2w (3 ECTS)	
7	Applied informatics	Foreign language II	
6	41 (4 ECTS)	3c (2 ECTS)	
5			
4		Graduate laboratory I	
3	Managerial course I 1w (2 ECTS) Foreign language I 1c (1 ECTS)	41 (6 ECTS)	
1	Graduation proseminar 1s (1 ECTS)		Graduation seminar 1s (2 ECTS)
Sem.	I	II	III

1. Set of obligatory and optional subjects and groups of classes in semestral arrangement

Semester 1

Obligatory subjects / groups of classes

Number of ECTS points 7

No.	Subject / groups of classescode	Name of subject / groups of classes (denote group of courses with symbol GK)	W	eekly 1	numbei	r of hou	ırs	Learning effect symbol	Number of hours		Number of ECTS points			Form ² of	W. 2 a	Subject / groups of classes			
			lec	cl	lab	pr	sem		ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2001W	Theoretical chemistry	2					K2Abs_W01 K2Abs_W02 K2Abs_W07 K2Abs_W08 K2Abs_W11 K2Abs_W17 K2Abs_K01	30	75	3	3	1,3	T/Z	E		DN		PD
2	W03BSS-SM2001L	Theoretical chemistry			2			K2Abs_U03 K2Abs_U04 K2Abs_U05 K2Abs_U09 K2Abs_U12 K2Abs_K08	30	50	2	2	1,4	Т	Z		DN	Р	K
3	W03BSS-SM2001C	Theoretical chemistry		1				K2Abs_U02 K2Abs_U12	15	50	2	2	0,7	T/Z	Z	_	DN	P	K
	_	Total	2	1	2				75	175	7	7	3,4	·	1			4	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

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Specialization subjects: *Bioinformatics* Number of ECTS points 17

	Specialization	subjects. Divinjormanes		T	uiii	CI U	1 11 0	To pomis.	<u> </u>										
No.	Subject / groups of	Name of subject / groups of	W	eekly 1	numbei	r of hou	ırs		Number of hours		Number of ECTS points			Form ² of	Way ³ of	Subject / groups of classes			
	classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem		ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2002W	Molecular dynamics	2					K2Abs_W03 K2Abs_W07 K2Abs_W09 K2Abs_W11 K2Abs_W18 K2Abs_K01	30	100	4	4	1,3	T/Z	Z		DN		PD
2	W03BSS-SM2002L	Molecular dynamics			2			K2Abs_U02 K2Abs_U03 K2Abs_U06 K2Abs_U10 K2Abs_K08	30	50	2	2	1,4	Т	Z		DN	P	S
3	W03BSS-SM2004W	Bioinformatics	2					K2Abs_W14	30	75	3	3	1,3	T/Z	E		DN		S
4	W03BSS-SM2004L	Bioinformatics.			2			K2Abs_U11	30	50	2	2	1,4	T	Z		DN	P	S
5	W03BSS-SM2003L	Networks and workstations with UNIX system			2			K2Abs_U13	30	50	2		1,4	Т	Z			P	S
6	W03BSS-SM2005L	Applied informatics			4			K2Abs_U13 K2Abs_U11 K2Abs_K08	60	100	4	4	2,8	Т	Z		DN	P	S
		Total	4		10				210	425	17	15	9,6		1			10	

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²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

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Optional subjects / groups of classes 6 ECTS points

No.		Name of subject / groups of classes (denote group of courses	W	eekly 1	numbei	of ho	urs			ber of urs	Numb	er of ECTS	points	Form ² of		Su	bject / grou	ps of classe	es
	Subject / groups of classescode	with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	SJO-SM0004	Foreign language I		1				K2Abs_U14 K2Abs_U15 K2Abs_K01 K2Abs_K04	15	30	1		0,6	T/Z	Z	0		Р	КО
2	W03-SM2002BH	Managerial course I	1					K2Abt_W16 K2Abs_K02 K2Abs_K03 K2Abs_K07	15	60	2		0,65	T/Z	Z	О			КО
3	W03W03-SM2053S	Graduation proseminar					1	K2Abs_U08 K2Abs_U14 K2Abs_K01 K2Abs_K07	15	25	1	1	0,7	T/Z	Z		DN	P	K
4	W03BSS-SM20BW	Elective course*	2					K2Abt_W02 K2Abt_K01	30	50	2		1,3	T/Z	Z				K
		Total	3	1			1		75	165	6	1	3,25					2	

Altogether in semester

	Total 1	number o	f hours		Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
lec	cl	lab	pr	sem					
9	2	12	0	1	360	765	30	23	16,25

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²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

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Semester 2

Obligatory subjects / groups of classes

Number of ECTS points 9

	O DIES	jects / groups of classes			1 10211		<u> </u>	2C15 point	· ·										
No.	Subject / anoung of	Name of subject / groups of	W	eekly r	numbei	r of ho	urs			ber of urs	Numl	ber of ECT	S points	Form ² of subjec		Sul	bject / grou	ps of classe	s
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	t / group s of classe s	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2008L	Retrieval of scientific and technical information			1			K2Abs_U05 K2Abs_K05	15	25	1		0,7	T	Z			P	K
2	W03BSS-SM2007W	Molecular modeling	1					K2Abs_W09 K2Abs_W11 K2Abs_W14 K2Abs_W17	15	50	2	2	0,65	T/Z	E		DN		K
3	W03BSS-SM2007L	Molecular modeling.			2			K2Abs_U13 K2Abs_U10 K2Abs_U05	30	50	2	2	1,4	Т	Z		DN	P	K
4	W03BSS-SM2007S	Molecular modeling					1	K2Abs_U07	15	25	1	1	0,7	T/Z	Z		DN	P	K
5	W03BSS-SM2006W	Rational drug design	2					K2Abs_W02 K2Abs_W05 K2Abs_W06 K2Abs_W10 K2Abs_W12 K2Abs_W13 K2Abs_W14 K2Abs_W15 K2Abs_W18 K2Abs_K01 K2Abs_K01 K2Abs_K06	30	75	3	3	1,3	T/Z	Z		DN		К
	•	Total	3		3		1		105	225	9	8	4,75		1			4	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Specialization subjects: *Bioinformatics* Number of ECTS points 10

No.	Subject / groups of	Name of subject / groups of	W	eekly	numbe	r of ho	urs		Num ho	ber of urs	Numl	ber of ECT	S points	Form ² of subjec		Sul	oject / grou	ps of classe	es
	classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	group s of classe s	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2012L	Advanced programming and numerical methods			3			K2Abs_W17 K2Abs_U11 K2Abs_U13 K2Abs_K04	45	75	3	3	2,1	Т	Z		DN	P	S
2	W03BSS-SM2011W	Bionanotechnology	2					K2Abs_W07 K2Abs_W09 K2Abs_W11 K2Abs_W13 K2Abs_W18	30	50	2	2	1,3	T/Z	E		DN		S
3	W03BSS-SM2011S	Bionanotechnology.					1	K2Abs_U07 K2Abs_K07	15	25	1	1	0,7	T/Z	Z		DN	P	S
4	W03BSS-SM2010P	Advanced bioinformatics				3		K2Abs_U11 K2Abs_U13 K2Abs_K04	45	75	3	3	2,25		Z		DN	P	S
5	W03BSS-SM2009L	Data mining			1			K2Abs_U02 K2Abs_K05	15	25	1	1	0,7	T	Z		DN	P	S
		Total	2		4	3	1		150	250	10	10	7,05		1			8	

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²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Optional subjects / groups of classes 11 ECTS points

No.	Cl.: / f	Name of subject / groups of	w	eekly 1	numbei	r of ho	urs			ber of urs	Numbe	er of ECTS	points	Form ² of		Sul	oject / grou	ps of classe	S
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	SJO-SM0003	Foreign language II		3				K2Abs_U14 K2Abs_U15 K2Abs_K01 K2Abs_K04	45	60	2		1,8	T/Z	Z	О		P	КО
2	W03-SM2001BH	Managerial course II	2					K2Abt_W16 K2Abs_K02 K2Abs_K03 K2Abs_K07	30	90	3		1,3	T/Z	Z	0			КО
3	W03W03-SM2054D	Graduate laboratory I			4			K2Abs_U02 K2Abs_U05 K2Abs_U14 K2Abs_K01 K2Abs_K05 K2Abs_K07	60	150	6	6	3	T	Z		DN	P	K
	•	Total	2	3	4				135	300	11	6	6,1					8	

Altogether in semester

	Total	number o	of hours		Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
lec	cl	lab	pr	sem					
7	3	11	3	2	390	775	30	24	17,9

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

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Semester 3

Obligatory subjects / groups of classes

Number of ECTS points 4

No.		Name of subject / groups of	W	eekly 1	number o	f hour	s			ber of urs	Numbe	er of ECTS	points	Form ² of		Sul	oject / grou	ps of classe	s
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2013W	Machine learning for chemistry and biology	2					K2Abs_W01 K2Abs_W14 K2Abs_W08 K2Abs_W09 K2Abs_K01	30	50	2		1,3	T/Z	Е				K
2	W03BSS-SM2013L	Machine learning for chemistry and biology			2			K2Abs_U01 K2Abs_U02 K2Abs_U04 K2Abs_U11 K2Abs_K02 K2Abs_K04 K2Abs_K05	30	50	2		1,4	Т	Z			P	K
		Total	2		2				60	100	4		2,7		1			2	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Specialization subjects: *Bioinformatics* Number of ECTS points 4

No.	Subject / groups of	Name of subject / groups of	W	eekly 1	numbei	of ho	urs		Num ho	per of urs	Numbe	er of ECTS	points	Form ² of	_	Sul	oject / grou	ps of classe	s
	classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2014W	Computational genomics	1					K2Abs_W02 K2Abs_W14 K2Abs_W17	15	25	1	1	0,65	T/Z	E		DN		S
2	W03BSS-SM2014L	Computational genomics.			1			K2Abs_U01 K2Abs_K07	15	25	1	1	0,7	Т	Z		DN	Р	S
3	W03BSS-SM2015L	Molecular engineering in genomic analyses			3			K2Abs_W06 K2Abs_U02 K2Abs_U03 K2Abs_U05 K2Abs_U08	45	50	2	2	2,1	T	Z		DN	P	S
		Total	1		4				75	100	4	4	3,45		1			3	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Optional subjects / groups of classes 22 ECTS points

No.	Subject / arround of	Name of subject / groups of	W	eekly r	numbei	r of hou	ırs			ber of urs	Numbe	er of ECTS	points	Form ² of		Sut	oject / grouj	ps of classe	s
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03W03-SM2055D	Graduate laboratory II			14			K2Abs_U02 K2Abs_U05 K2Abs_U14 K2Abs_U06 K2Abs_K01 K2Abs_K05 K2Abs_K07	210	500	20	20	9,5	Т	Z		DN	P	K
2	W03W03-SM2056S	Graduation seminar					1	K2Abs_U05 K2Abs_U07 K2Abs_K01 K2Abs_K06 K2Abs_K07 K2Abs_K08	15	50	2	2	0,7	T/Z	Z		DN	P	K
		Total			14		1		225	550	22	22	10,2					22	

Altogether in semester

	Total 1	number o	of hours		Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
lec	cl	lab	pr	sem					
3		20		1	360	750	30	26	16,35

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⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

2. Set of examinations in semestral arrangement

Subject / groups of classes code	Names of subjects / groups of classesending with examination	Semester
W03BSS-SM2001W	Theoretical chemistry	1
W03BSS-SM2004W	Bioinformatics	1
W03BSS-SM2007W	Molecular modeling	2
W03BSS-SM2011W	Bionanotechnology	2
W03BSS-SM2014W	Computational genomics	2
W03BSS-SM2013W	Machine learning for chemistry and biology	3

3. Numbers of allowable deficit of ECTS points after particular semesters

Semester	Allowable deficit of ECTS points after semester
1	15
2	15
3	0

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³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Opinion of student government legi	slative body
Date	Name and surname, signature of student representative
Date	Dean's signature

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses ⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

PLAN OF STUDIES

FACULTY: Chemistry

MAIN FIELD OF STUDY: BIOSCIENCES

EDUCATION LEVEL: second-level studies (3-semester)

FORM OF STUDIES: full-time studies

PROFILE: general academic

SPECIALIZATION: Medicinal Chemistry

LANGUAGE OF STUDY: English

In effect since 2024/2025

Plan of studies structure (optionally)

1) in ECTS point layout (space for scheme of plan)

2) in hourly layout (space for scheme of plan)

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

2nd LEVEL STUDIES (MASTER LEVEL STUDIES) (3 sem)

Field of study: BIOSCIENCES

Specialization: Medicinal Chemistry

Specialization subjects
Obligatory subjects
Optional subjects

Sem.	I	П	III
Godz.	24h / 30 ECTS / 2E	25h / 30 ECTS / 3E	24h / 30 ECTS /1E
25		Rational drug design	
24	Elective course	2w (3ECTS)	Multistep organic synthesis
23	2w (2 ECTS)	Molecular modeling E	41 (3 ECTS)
22	Theoretical chemistry E	1w+2l+1s	
21	2w + 1c + 2l	(2+2+1) ECTS	
	(3+2+2) ECTS		Inorganic drugs
20			1w (1 ECTS)
10		Retrieval of scientific and technical information	Machine learning for chemistry and biology
19		11 (1 ECTS) Metabolomics	2w+21 E
18	Isolation and identification of bioproducts	21 (2 ECTS)	(2+2) ECTS
17	21 (2 ECTS)		
16		Medicinal natural products E	
	Introductory statistics	1w +2l	Graduate laboratory II
15	1c (2 ECTS)	(2 +2) ECTS	14l (20 ECTS)
14	Crystallography and structure of solids $2w+1c$		
13	2w +1c (2+1) ECTS	Modern pharmaceuticals and biopharmaceuticals E	
12		2w+2l (2+2) ECTS	
11	Analytical methods in drug design and technology	(2+2) EC13	
10	1w +21		
9	(2+2) ECTS	Managerial course II	
8	Spectroscopic methods in medicinal chemistry E	2w (3 ECTS)	
7	2w + 2l	Foreign language II	
	(2+2) ECTS	3c (2 ECTS)	
5			
5	Metabolomics	Graduate laboratory I	
4	1w (2 ECTS)	41 (6 ECTS)	
	Managerial course I		
3	1w (2 ECTS)		
	Foreign language I 1c (1 ECTS)		
1	Graduation proseminar 1s (1 ECTS)		Graduation seminar 1s (2 ECTS)
	T (1 De 10)	TT.	
Sem.	l	II	III

1. Set of obligatory and optional subjects and groups of classes in semestral arrangement

Semester 1

Obligatory subjects / groups of classes

Number of ECTS points 7

	Obligatory su	bjects / groups of classes			1 (uli	IDCI	OI I	2C15 point	3 1										
No.	Subject / groups of	Name of subject / groups of	W	eekly 1	numbe	of ho	urs			ber of urs	Num	ber of EC	ΓS points	Form ² of		Su	bject / grou	ps of classe	s
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2001W	Theoretical chemistry	2					K2Abs_W01 K2Abs_W02 K2Abs_W07 K2Abs_W08 K2Abs_W17 K2Abs_W11 K2Abs_K01	30	75	3	3	1,3	T/Z	E		DN		PD
2	W03BSS-SM2001L	Theoretical chemistry			2			K2Abs_U03 K2Abs_U04 K2Abs_U05 K2Abs_U09 K2Abs_U12 K2Abs_K08	30	50	2	2	1,4	Т	Z		DN	Р	K
3	W03BSS-SM2001C	Theoretical chemistry		1				K2Abs_U02 K2Abs_U12	15	50	2	2	0,7	T/Z	Z		DN	P	K
		Total	2	1	2				75	175	7	7	3,4		1	_		4	

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²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

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Specialization subjects: *Medicinal Chemistry* Number of ECTS points 17

No.	G 1: . /	Name of subject / groups of	W	eekly 1	number	r of hou	ırs		Num		Num	ber of ECT	S points	Form ² of		Su	bject / grou	ps of classe	es
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2021W	Metabolomics	1					K2Abs_W02 K2Abs_W06 K2Abs_W13 K2Abs_K01	15	50	2	2	0,65	T/Z	Z		DN		S
2	W03BSS-SM2020W	Spectroscopic methods in medicinal chemistry	2					K2Abs_U02 K2Abs_U08	30	50	2	2	1,3	T/Z	E		DN		S
3	W03BSS-SM2020L	Spectroscopic methods in medicinal chemistry			2			K2Abs_W02 K2Abs_W06 K2Abs_W13 K2Abs_K08	30	50	2	2	1,4	Т	Z		DN	P	S
4	W03BSS-SM2019W	Analytical methods in drug design and technology	1					K2Abs_W02 K2Abs_W06	15	50	2	2	0,65	T/Z	Z		DN		S
5	W03BSS-SM2019L	Analytical methods in drug design and technology.			2			K2Abs_U01 K2Abs_U03 K2Abs_U08	30	50	2	2	1,4	Т	Z		DN	Р	S
6	W03BSS-SM2018W	Crystallography and structure of solids	2					K2Abs_W02 K2Abs_W06 K2Abs_W07 K2Abs_W18 K2Abs_K01	30	50	2	2	1,3	T/Z	Z		DN		S
7	W03BSS-SM2018C	Crystallography and structure of solids		1				K2Abs_U01 K2Abs_U02	15	25	1	1	0,7	T/Z	Z		DN	P	S
8	W03BSS-SM2017C	Introductory statistics		1				K2Abs_U03	15	50	2		0,7	T/Z	Z			P	PD
9	W03BSS-SM2016L	Isolation and identification of bioproducts			2			K2Abs_U02 K2Abs_U03 K2Abs_W05 K2Abs_K08	30	50	2	2	1,4	Т	Z		DN	P	S
		Total	6	2	6				210	425	17	15	9,5		1			9	

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⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

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Optional subjects / groups of classes 6 ECTS points

No.		Name of subject / groups of classes (denote group of courses	W	Weekly number of hours					ber of ours	Numb	er of ECTS	S points	Form ² of		Sul	oject / grou	ps of classe	s	
	Subject / groups of classescode	with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	SJO-SM0004	Foreign language I		1				K2Abs_U14 K2Abs_U15 K2Abs_K01 K2Abs_K04	15	30	1		0,6	T/Z	Z	0		Р	КО
2	W03-SM2002BH	Managerial course I	1					K2Abt_W16 K2Abs_K02 K2Abs_K03 K2Abs_K07	15	60	2		0,65	T/Z	Z	О			КО
3	W03W03-SM2053S	Graduation proseminar					1	K2Abs_U08 K2Abs_U14 K2Abs_K01 K2Abs_K07	15	25	1	1	0,7	T/Z	Z		DN	Р	К
4	W03BSS-SM20BW	Elective course*	2					K2Abt_W02 K2Abt_K01	30	50	2		1,3	T/Z	Z				K
		Total	3	1			1		75	165	6	1	3,25					2	

Altogether in semester

	Total 1	number o	f hours		Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
lec	cl	lab	pr	sem					
11	4	8		1	360	765	30	23	16,15

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²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

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Semester 2

Obligatory subjects / groups of classes Num

Number of ECTS points 9

	Obligatory sub	jects / groups of classes	IDCI	OI I	C 15 point	5 /													
No.	Cl.:	Name of subject / groups of	W	eekly ı	numbei	r of ho	urs			ber of urs	Numl	ber of ECT	S points	Form ² of subjec		Sul	oject / grou	ps of classe	s
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	group s of classe s	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2008L	Retrieval of scientific and technical			1			K2Abs_W09	15	25	1		0,7	T	Z			P	K
		information						K2Abs_W11											
								K2Abs_W14											
								K2Abs_K05											
2	W03BSS-SM2007W	Molecular modeling	1					K2Abs_W17	15	50	2	2	0,65	T/Z	E		DN		K
								K2Abs_U13											
								K2Abs_U10											
	**************************************	261						K2Abs_U05	20	7 0	2						517		**
3	W03BSS-SM2007L	Molecular modeling.			2			K2Abs_U07	30	50	2	2	1,4	T	Z		DN	P	K
4	W03BSS-SM2007S	Molecular modeling					1	K2Abs_W09	15	25	1	1	0,7	T/Z	Z		DN	P	K
								K2Abs_W11											
								K2Abs_W14											
5	W03BSS-SM2006W	Rational drug design	2					K2Abs_W02	30	75	3	3	1,3	T/Z	Z		DN		K
								K2Abs_W05											
								K2Abs_W06											
								K2Abs_W10											
								K2Abs_W12											
								K2Abs_W13											
								K2Abs_W14											
								K2Abs_W15 K2Abs_W18											
								K2Abs_K01											
								K2Abs_K06											
	I	Total	3		3		1		105	225	9	8	4,75		1			4	
			1	1	1	1	1					1	/	1		l		ı	

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³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Specialization subjects: Medicinal Chemistry

Number of ECTS points 10

	~ P • • • • • • • • • • • • • • • • • •	Bubjects. Medicinal Chem	120. 3			1 102	11201	or ECTO	3011100										
No.	Subject / groups of	Name of subject / groups of	W	eekly	numbe	r of ho	urs		Numl ho	ber of urs	Num	ber of ECT	S points	Form ² of subjec		Su	bject / grou	ps of classe	es .
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	group s of classe s	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2023W	Modern pharmaceuticals and biopharmaceuticals	2					K2Abs_W02 K2Abs_W07 K2Abs_W12 K2Abs_W18 K2Abs_W06 K2Abs_K01	30	50	2	2	1,3	T/Z	E		DN		S
2	W03BSS-SM2023L	Modern pharmaceuticals and biopharmaceuticals			2			K2Abs_U02 K2Abs_U05 K2Abs_K05 K2Abs_K08	30	50	2	2	1,4	Т	Z		DN	P	S
3	W03BSS-SM2022W	Medicinal natural products	1					K2Abs_W03 K2Abs_W12 K2Abs_W15 K2Abs_W18	15	50	2	2	0,65	T/Z	Е		DN		S
4	W03BSS-SM2022L	Medicinal natural products.		_	2			K2Abs_U02 K2Abs_K04	30	50	2	2	1,4	T	Z		DN	P	S
5	W03BSS-SM2021L	Metabolomics			2			K2Abs_W14 K2Abs_U03 K2Abs_K06	30	50	2	2	1,4	T	Z		DN	P	S
		Total	3		6				135	250	10	10	6,15		2			6	

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²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Optional subjects / groups of classes 11 ECTS points

No.	Codings / amount of	Name of subject / groups of	W	eekly	numbe	r of ho	urs			ber of ours	Numb	er of ECTS	S points	Form ² of		Sul	bject / grou	ps of classe	es
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	SJO-SM0003	Foreign language II		3				K2Abs_U14 K2Abs_U15 K2Abs_K01 K2Abs_K04	45	60	2		1,8	T/Z	Z	О		Р	КО
2	W03-SM2001BH	Managerial course II	2					K2Abt_W16 K2Abs_K02 K2Abs_K03 K2Abs_K07	30	90	3		1,3	T/Z	Z	О			КО
3	W03W03-SM2054D	Graduate laboratory I			4			K2Abs_U02 K2Abs_U05 K2Abs_U14 K2Abs_K01 K2Abs_K05 K2Abs_K07	60	150	6	6	3	T	Z		DN	P	K
	•	Total	2	3	4				135	300	11	6	6,1					8	

Altogether in semester

	Total 1	number o	of hours		Total number of ZZU hours	Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹
lec	cl	lab	pr	sem					
8	3	13		1	375	775	30	24	17

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²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Semester 3

Obligatory subjects / groups of classes

Number of ECTS points 4

No.	Subject / groups of	Name of subject / groups of	W	eekly 1	number	r of ho	urs			ber of urs	Numbe	er of ECTS	points	Form ² of		Sul	oject / grou	ps of classe	s
	classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2013W	Machine learning for chemistry and biology	2					K2Abs_W01 K2Abs_W14 K2Abs_W08 K2Abs_W09 K2Abs_K01	30	50	2		1,3	T/Z	Е				K
2	W03BSS-SM2013L	Machine learning for chemistry and biology			2			K2Abs_U01 K2Abs_U02 K2Abs_U04 K2Abs_U11 K2Abs_K02 K2Abs_K04 K2Abs_K05	30	50	2		1,4	Т	Z			P	K
		Total	2		2				60	100	4		2,7		1			2	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Specialization subjects: *Medicinal chemistry* Number of ECTS points 4

No.	C1:	Name of subject / groups of	W	eekly 1	numbe	r of ho	urs			ber of urs	Numbe	er of ECTS	S points	Form ² of		Sul	oject / grou	ps of classe	s
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03BSS-SM2025W	Inorganic drugs	1					K2Abs_W02 K2Abs_W03	15	25	1	1	0,65	T/Z	Z		DN		S
2	W03BSS-SM2024L	Multistep organic synthesis			4			K2Abs_U02 K2Abs_U04 K2Abs_U05 K2Abs_K08	60	75	3	3	2,8	T	Z		DN	Р	S
		Total	1		4				75	100	4	4	3,45		1			3	

Optional subjects / groups of classes 22 ECTS points

_	<u> </u>						1 .					Optional Subjects (Groups of classes							
No.	Subject / arroying of	Name of subject / groups of		Number of Number of ECTS points Number of Subject / groups of Number of Subject / groups of Number of Hours Number of ECTS points Form ² of			Sul	Subject / groups of classes											
	Subject / groups of classescode	classes (denote group of courses with symbol GK)	lec	cl	lab	pr	sem	Learning effect symbol	ZZU	CNPS	Total	DN ⁵ classes	BU ¹ classes	subject / groups of classes	Way ³ of crediting	University -wide ⁴	Concerni ng scientific activities ⁵	Practical ⁶	Type ⁷
1	W03W03-SM2055D	Graduate laboratory II			14			K2Abs_U02 K2Abs_U05 K2Abs_U14 K2Abs_U06 K2Abs_K01 K2Abs_K05 K2Abs_K07	210	500	20	20	9,5	Т	Z		DN	P	K
2	W03W03-SM2056S	Graduation seminar					1	K2Abs_U05 K2Abs_U07 K2Abs_K01 K2Abs_K06 K2Abs_K07 K2Abs_K08	15	50	2	2	0,7	T/Z	Z		DN	P	K
		Total			14		1		225	550	22	22	10,2					22	

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Altogether in semester

	Total 1	number of hours Total number of ZZU hours		Total number of CNPS hours	Total number of ECTS points	Total number of ECTS points for DN classes ⁵	Number of ECTS points for BU classes ¹		
lec	cl	lab	pr	sem					
3		20		1	360	750	30	26	16,35

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

2. Set of examinations in semestral arrangement

Subject / groups of classescode	Names of subjects / groups of classesending with examination	Semester
W03BSS-SM2001W	Theoretical chemistry	1
W03BSS-SM2020W	Spectroscopic methods in medicinal chemistry	1
W03BSS-SM2007W	W03BSS-SM2007W Molecular modeling	
W03BSS-SM2022W	Medicinal natural products	2
W03BSS-SM2023W	Modern pharmaceuticals and biopharmaceuticals	
W03BSS-SM2013W	Machine learning for chemistry and biology	3

3. Numbers of allowable deficit of ECTS points after particular semesters

Semester	Allowable deficit of ECTS points after semester
1	15
2	15
3	0

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses

⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

Opinion of student government legislative body						
Date	Name and surname, signature of student representative					
Date	Dean's signature					

¹BU – number of ECTS points assigned to hours of classes requiring direct participation of academic teachers and other persons conducting classes

²Traditional – enter T, remote – enter Z

³Exam – enter E, crediting – enter Z. For the group of classes – after the letter E or Z - enter in brackets the final subject form (lec, cl, lab, pr, sem)

⁴University-wide subject /group of classes – enter O

⁵DN - number of ECTS points assigned to the classes related to the University's academic activity in the discipline/disciplines to which the main field of study is assigned

⁶Practical subject / group of classes – enter P. For the group of classes – in brackets enter the number of ECTS points assigned to practical courses ⁷KO – general education courses, PD – basic sciences courses, K – main field of study courses, S – specialization courses

KARTY PRZEDMIOTÓW

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Zaawansowana Bioinformatyka

Name of subject in English Advanced bioinformatics

Main field of study (if applicable): Biosciences Specialization (if applicable): Bioinformatics

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03BSS-SM2010P

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)				30	
Number of hours of total student workload (CNPS)				60	
Form of crediting (Examination / crediting with grade)				crediting with grade	
For group of courses mark (X) final course					
Number of ECTS points				2	
including number of ECTS points for practical classes (P)				2	
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)				1	

*delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Knowledge of biological basis related to bioinformatics
- 2. Proficiency in use of online bioinformatics databases of sequences, genes, genomes rtc.
- 3. Proficiency with use of Unix systems, ability to work in the command line
- 4. Ability to solve algorithmic problems and automating computing tasks with use of a programming language like Python

SUBJECT OBJECTIVES

- C1 Gain of the skills necessary to use remote computing facilities ("cloud computing" services), including preparation and configuration of system images packaged with necessary sotware to be run on such services;
- C2 Gain of the skills necessary to document and archive the team work on the process of creation, testing and development of specialized software and workflows to process bioinformatical data sets, with use of interactive Jupyter notebooks, version control systems and online repositories;
- C3 Gain of the skills necessary to perform data processing, analysis and interpretation of the results of new generation sequencing experiments, including quality control and filtering (trimming), mapping on reference genomes, analysis and visualization;
- C4 Gain of the skill necessary to use GNU R package with selected Bioconductor modules to perform typical bioinformatics analyses and visualization of experimental results.

SUBJECT EDUCATIONAL EFFECTS

Relating to skills:

- PEU_U01 Students can prepare and configure system images with software packages and services required to run bioinformatics related computation, data processing and analysis on remote computing facilities ("cloud computing").
- PEU_U02 Students can document and archive the performed analysis workflows and results and developed software codes using version control systems, online repositories and interactive notebooks.
- PEU_U03 Students can perform full analysis of NGS sequencing results, from the initial data processing to mapping on the reference genomes, visualization and interpretation
- PEU_U04 Students can use the GNU R system and Bioconductor libraries to perform typical analyses of experimental data sets available online, and to visualize the results.

Relating to social competences:

PEU_K01 Students can work in teams to solve problems and improve proposed solutions PEU_K02 Students can communicate and describe the workflow and results of typical data analyses performed in bioinformatics.

Project Project Introductory classes: the form and organization of the classes, didactic 2

2

4

10

developed software:
Version control system git and online repositories (github, gitlab etc.). Team working practices.

Proj 2 Documentation and archiving of the performed analyses, workflows and

- Documentation of the work using interactive Jupyter notebooks
- Markdown syntax

materials, requirements for the final grade

- Proj 3 Project 1: Cloud computing

 Preparation of system images for remote execution using Docker.

 Configuring network services and permanent storage. Simple servers of Jupyter, ssh, sshfs.
 - Preparation and configuration of Docker system image with all services needed to run the GALAXY environment.
 - Preparation and configuration of Docker system image with Jupyter or JupyterLab server, Jupyter R kernel, R system and selected Bioconductor modules.
 - Depositing of the results (Dockerfiles, notebooks) in a git repository. Working in teams to test solutions and solve problems.
- Proj 4 Project 2: Analysis of NGS sequencing data using the GALAXY platform
 - Different sequencing platforms and their relation to the results and their analysis
 - Characteristics of NGS data: read quality Q, FASTQ format variants, sequencing depth, filtering of low quality results.
 - Initial data processing (trimming).
 - Mapping results on the reference genome; analysis and visualization of results.

Proj 5	Project 3: Analysis of experimental data sets using GNU R and Bioconductor					
	 Introduction to use of the GNU R system: data types, plotting and visualization Selected Bioconductor modules and functions Characteristics of experimental data depending on the platform, required processing Analysis of example data sets 					
Proj 6	6 Students presentations of their projects					
	Total hours					

TEACHING TOOLS USED

- N1. Instructions and video recording for self-study prior to the relevant classes (the "reverse classroom" approach).
- N2. Multimedia presentations and live demonstration how to use software.
- N3. Problem solving individual and in teams with the help of the tutor and using online resources.
- N4. Use of specialized software for performing typical analyses of bioinformatics data sets.
- N5. Students presentations of their work, solved problems and results.

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

valuation (F – forming uring semester), P – oncluding (at semester nd)	Learning outcomes code	Way of evaluachievement	ating learning outcomes
	PEU_U01, PEU_U02, PEU_K01	Project 1 scor	re
2	PEU_U03, PEU_K02	Project 2 scor	re
3	PEU_U04, PEU_K02	Project 3 scor	re
= F1 + F2 + F3		Score 50-59,99% 60-69,99% 70-79,99% 80-89,99% 90-100%	Grade 3,0 3,5 4,0 4,5 5,0
PRI	MARY AND SECON	80-89,99% 90-100%	4,5 5,0

PRIMARY LITERATURE:

Due to rapid progress of the relevant technologies, the best sources of information are the online learning resources and software documentation:

- [1] https://git-scm.com/doc
- [2] https://docs.docker.com
- [3] https://docs.jupyter.org
- [4] <u>https://training.galaxyproject.org</u>
- [5] https://cran.r-project.org/doc/manuals
- [6] https://bioconductor.org/help
- [7] Relevant articles on https://wikipedia.org

SECONDARY LITERATURE:

- [1] "Next-generation sequencing: current technologies and applications", ed. Xu, Jianping; Caister Academic Press, Norfolk 2014. ISBN 978-1-908230-33-1
 - https://omnis-
 - pwr.primo.exlibrisgroup.com/permalink/48OMNIS_TUR/d7ok8p/alma9960747679207668
 - (mostly of historical interest due to rapid progress of NGS technology and software, but it does introduce the basic background and concepts)
- [2] https://socviz.co/gettingstarted.html (Introduction to RMarkdown as a tool to document an analysis workflow and results using the R system)

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Paweł Kędzierski, pawel.kedzierski@pwr.edu.pl

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Zaawansowane programowanie i metody numeryczne Name of subject in English Advanced programming and numerical methods

Main field of study (if applicable): Biosciences **Specialization** (if applicable): Bioinformatics

Profile: academic / practical*

Level and form of studies: 1st/ 2nd level, uniform magister studies*, full-time / part-time*

Kind of subject: obligatory / optional / university-wide*

Subject code W03BSS-SM2012L Group of courses YES/NO*

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)			45		
Number of hours of total student workload (CNPS)			75		
Form of crediting (Examination / crediting with grade)			Crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points			3		
including number of ECTS points for practical classes (P)			3		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			2,1		

*delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Knowledge of basics of any programming language
- 2.
- 3.

SUBJECT OBJECTIVES

- C1 Familiarizing students with good practices in programming
- C2 Teaching students the construction of algorithms
- C3 Familiarizing students with numerical recipes

SUBJECT EDUCATIONAL EFFECTS

relating to skills:

PEU_U01 Student is able to apply a random number generator in Monte Carlo algorithms

PEU_U02 Student is able to design and implement an algorithm for different sorting algorithms

PEU_U03 Student is able to develop the code for numerical integration of Newton equations of motion

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relating to social competences:

PEU	K01	Student i	s able to	work in team
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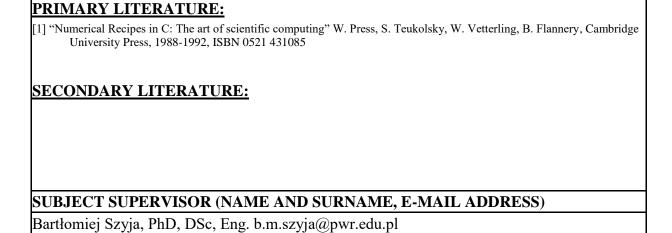
	PROGRAMME CONTENT					
	Laboratory					
Lab 1	Organization of course and conditions for passing the course. Programming environment.	3				
Lab 2	Random number generators.	3				
Lab 3	Numerical integration of functions.	3				
Lab 4	Interpolation and extrapolation.	9				
Lab 5	Numerical analysis of functions.	12				
Lab 6	Monte Carlo methods.	12				
Lab 7	End credit	3				
	Total hours	45				

TEACHING TOOLS USED

- N1.Multimedia presentation N2.Specialized computer software N3.Gamification

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	Learning outcomes code	Way of evaluating learning outcomes achievement					
P	PEU_W01-W03 PEU_U01-U03, PEU_K01	Final assignment (max 100 pts)					
3.0, when (F1+F2) = 51-5 3.5, when (F1+F2) = 60-6 4.0, when (F1+F2) = 70-7 4.5, when (F1+F2) = 80-8	P 2.0, when (F1+F2) < 50% points 3.0, when (F1+F2) = 51-59% points 3.5, when (F1+F2) = 60-69% points 4.0, when (F1+F2) = 70-79% points 4.5, when (F1+F2) = 80-89% points 5.0, when (F1+F2) = 90-99% points						
PRIMARY AND SECONDARY LITERATURE							



Attachment no. 4. to the Program of Studies

FACULTY of CHEMISTRY

SUBJECT CARD

Name of subject in Polish Metody Analityczne w Projektowaniu i Technologii

Wytwarzania Leku

Name of subject in English Analytical Methods in Drug Design and Technology

Main field of study (if applicable): Biosciences

Specialization (if applicable): Medicinal Chemistry

Profile: academic / practical*

Level and form of studies: 1st/ 2nd level, uniform magister studies*, full-time / part-time*

Kind of subject: obligatory / optional / university-wide* Subject code W03BSS-SM2019W, W03BSS-SM2019L

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	15		30		
Number of hours of total student workload (CNPS)	50		50		
Form of crediting (Examination / crediting with grade)	crediting with grade		crediting with grade		
For group of courses mark (X) final course	X				
Number of ECTS points	2		2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)	0.65		1,4		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Principles of organic chemistry, theoretical and practical.
- 2. Basic knowledge on chromatographic and spectroscopic methods.
- 3. Knowledge in the field of basis of analytical chemistry is recommended.

SUBJECT OBJECTIVES

- C1 To acquaint student with the theoretical and practical aspects of good laboratory practice (GLP) and good manufacture practice (GMP).
- C2 Gaining of the knowledge on the modern chromatographic techniques and their applications in drug design and technological process of drugs production.
- C3 Acquaintance with the different technological concepts of application of spectroscopic methods in drugs design and quality control in the production system.
- C4 Expanding the knowledge in the field of electrochemical methods applications in the design

of biologically active compounds and the production procedures of them.

C5 Acquaintance with the different concepts in the field of mixed analytical methods.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

Student, who has completed the course:

- PEU_W01 has knowledge on good laboratory practice (GLP) rules, good manufacture practice (GMP) rules, and validation procedures necessary to be used in analytical methods,
- PEU_W02 has knowledge about the modern chromatographic, spectroscopic, electrochemical and mixed analytical techniques and their applications in drug design and technological process of drugs production,
- PEU_W03 can define the advantages and disadvantages of the analytical techniques, the sensitivity level of each of them.

relating to skills:

Student, who has completed the course:

- PEU_U01 has skills of use chromatographic techniques for separation of a mixture of different compounds, to detect them, do interpretation of the results and prepare the report according to GLP,
- PEU_U02 has knowledge about using different types of spectrometric instruments, and about the parameters of the sample ready to analyze,
- PEU_U03 has skills to do the analysis of the biologically active compounds using electrochemical methods, do interpretation of the results and prepare the report according to GLP,
- PEU_U04 has skills to detect the biologically active compounds in a drug formulation using physical and physicochemical methods.

PROGRAMME CONTENT Number Lecture of hours Introduction to analytical techniques as tools for drug design and Lec 1 production. Good practice rules in analytical chemistry. Error estimation in 2 analytical methods used in drugs design and technology. Validation techniques. Pharmacopoeias. GLP, GMP and drugs production Lec 2 2 normalization rules. Chromatographic techniques in drugs design and control of production 2 Lec 3 process. Solving of popular troubles. Lec 4 Spectroscopic techniques in drugs design and control of production process. 2 Mixed advanced analytical techniques as a tool in drugs design and control Lec 5 2 of their activity. 2 Lec 6 The electrochemical methods in drug design and technology. Methods of the analysis of solid state drug formulation ingredients -Lec 7 2 powders and granules. Novel advanced applications in quality control systems in the Lec 8 1 pharmaceutical industry. Total hours 15

	Laboratory	Number of hours
Lab 1	Safety rules in the laboratory of organic chemistry, good laboratory practice and the rules of the reports preparation.	2
Lab 2	HPLC technique – a scheme of the procedure of a sample preparation. Preparation of a sample to HPLC analysis.	2
Lab 3	HPLC – the equipment scheme. The analysis of biologically active components of a pharmaceutical formulation. Gas chromatography equipment and the procedure of analysis. Detection techniques.	2
Lab 4	GC analysis - diagram of API separation procedure. Sample preparation for GC analysis.	2
Lab 5	GC-MS — the equipment diagram. Chromatographic analysis and interpretation of the results.	2
Lab 6	Turbidimetry – the analytical method useful to drug design and quality control of it using microplates reader.	4
Lab 7	Comparison of thermostability and photostability of the active substance in solid, semi-solid and liquid pharmaceutical formulations.	4
	Potentiometry – the method used for potentiometric titration of the biologically active molecules possessing positive or negative charge. Application of potentiometric titration to pH-metric analysis.	4
Lab 9	UV-Vis spectrophotometry – principles of the method and procedure of measurement. The quality analysis of a pharmaceutical formulation.	4
Lab 10	Infrared spectroscopy (FT-IR) of a biologically active compound. Sample preparation and spectrum collection.	4
	Total hours	30

TEACHING TOOLS USED

- N1 Multimedial presentation.
- N2 Performing experiments with different laboratory equipment and instruments.
 N3 Preparation of report including analysis and interpretation of obtained results.

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F –	Learning	Way of evaluating learning outcomes achievement			
forming during	outcomes code				
semester), P –					
concluding (at					
semester end)					
F1	PEU_W01 -	grades for the short queries in the topics of the laboratory			
r ₁	PEU_W03	experiments.			
F2	PEU_U01 –	grades for reports on the experiments conducted.			
	PEU_U4				
		Average from N grades for the queries (F1) and N for the			
P1 (laboratory)		reports on the experiments conducted (F2)			
		$P1 = \Sigma (F1+F2)/N$			
P2 (lecture)	PEU_W01-	Final test.			
	PEU_W03				

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] J. Ermer, J.H.McB. Miller, Method Validation in Pharmaceutical Analysis. A Guide to Best Practice. Wiley-VCH, Weincheim. 2005.
- [2] Farmakopea Polska, Urząd Rejestracji Leków, Wyrobów Medycznych i Produktów Biobójczych, Warszawa.
- [3] W. Jennings, E. Mittlefehldt, P. Stremple, Analytical Gas Chromatography. 2nd Ed. Academic Press, 1997.
- [4] R.P.W. Scott, Tandem Techniques. John Wiley & Sons, 1997.
- [5] M.S. Lee, Integrated Strategies in Drug Discovery Using Mass Spectrometry. John Wiley & Sons, 2005.
- [6] A.J. Bard, R.L. Faulkner, Electrochemical Methods. Fundamental and Applications. John Wiley & Sons, 2001.

SECONDARY LITERATURE:

- [1] D.M. Bliesner, Validating Chromatographic Methods. A Practical Guide. John Wiley & Sons, 2006.
- [2] P.A. Christensenand A. Hamnett, Techniques and Mechanisms in Electrochemistry. Kluver Academic Press, 1994.
- [3] AC Moffat, MD Osselton, B Widdop, Clarke's analysis of drugs and poisons. Pharmaceutical Press, 2005.
- [4] F.A. Settle, Handbook of Instrumental Techniques for Analytical Chemistry. Prentice-Hall Inc., 1997.

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

dr hab. inż. Izabela Pawlaczyk-Graja, prof. uczelni izabela.pawlaczyk@pwr.edu.pl

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Informatyka stosowana Name of subject in English Applied informatics Main field of study (if applicable): Biosciences Specialization (if applicable): Bioinformatics

Profile: academic / practical*

Level and form of studies: 1st/ 2nd level, uniform magister studies*, full-time / part-time*

Kind of subject: obligatory / optional / university-wide*

Subject code W03BSS-SM2005L Group of courses YES-/ NO*

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)			60		
Number of hours of total student workload (CNPS)			100		
Form of crediting (Examination / crediting with grade)			Crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points			4		
including number of ECTS points for practical classes (P)			4		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			2,8		

*delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

1. None

SUBJECT OBJECTIVES

- C1 Familiarizing students with basics of Unix environment
- C2 Teaching students the basic algorithms and numerical methods
- C3 Familiarizing students with concepts of procedural and object-oriented programming

SUBJECT EDUCATIONAL EFFECTS

• • •

relating to skills:

PEU_U01 Student is able to use a programming environment to develop a program

PEU_U02 Student is able to design and implement an algorithm for the common numerical methods

PEU_U03 Student is able to effectively use the procedural and object-oriented methods in programming

. . .

relating to social competences:

PEU_K01 – uznaje znaczenie wiedzy w rozwiązywaniu problemów informatycznych

	PROGRAMME CONTENT	
	Laboratory	Number of hours
Lab 1	Organization of course and conditions for passing the course. Basic Unix commands.	4
Lab 2	BASH scripting. Resource management. Queueing systems	8
Lab 3	Using the programming environment. Writing and executing programs. Conditional expressions. Loops.	8
Lab 4	Simple and complex types of data. Objects. Functions and methods.	8
Lab 5	Commonly used numerical algorithms.	16
Lab 6	Applications of programming in biochemistry and biotechnology	12
Lab 7	End credit	4
	Total hours	60

- N1.Multimedia presentation N2.Specialized computer software N3.Gamification

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	Learning outcomes code	Way of evaluating learning outcomes achievement				
F1	PEU_W01-W03, PEU_K01	Partial tests (max 4 pts)				
P1	PEU_U01-U03, PEU_K01	Final assignment (max 6 pts)				
P (F1+P1)		•				
2.0, if $P < 50\%$ pts						
3.0, if $P = 51-59%$ pts	3.0, if $P = 51-59%$ pts					
3.5, if $P = 60-69%$ pts	6.5, if $P = 60-69%$ pts					
4.0, if P = 70-79% pts	.0, if P = 70-79% pts					
4.5, if P = 80-89% pts	.5, if P = 80-89% pts					
5.0, if P = 90-99% pts	0.0, if $P = 90-99%$ pts					
5.5, if $P = 100%$ pts	5.5, if $P = 100%$ pts					
PR	PRIMARY AND SECONDARY LITERATURE					

PRIMARY LITERATURE: [1] "Python Programming for Biology: Bioinformatics and Beyond", Tim J. Stevens, Wayne Boucher, Cambridge University Press; 1 edition (April 6, 2015) ISBN-13: 978-0521720090 [2] [3] [4] SECONDARY LITERATURE: [1] [2] [3]

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Bartłomiej Szyja, PhD, DSc, Eng. b.m.szyja@pwr.edu.pl

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish
Name of subject in English
Main field of study (if applicable): Biosciences
Specialization (if applicable): Bioinformatics
Profile: Bioinformatics
academic

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03BSS-SM2004W, W03BSS-SM2004L

Group of courses NO

<u>.</u>					
	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30		30		
Number of hours of total student workload (CNPS)	75		50		
Form of crediting (Examination / crediting with grade)	Examination		crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points	3		2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)	1,3		1,4		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Knowledge of basics concepts and vocabulary of biology, biochemistry and genetics
- 2. Ability to use computer, Internet and command line (shell) interface
- 3. Fluent command of English language

SUBJECT OBJECTIVES

- C1 Teaching the retrieval of specific information from databases of biosequences, genes, genomes, structures, protein families and other biochemistry and medicine related databases.
- C2 Understanding of various sequence similarity measures ant their interpretation required to perform comparative analysis of multiple sequences.
- C3 Ability to search for homologous sequences, creation and use of sequence similarity profiles ant to analyze relations between sequences.
- C4 Ability to build and evaluate protein models using contemporary structure prediction methods.
- C5 Ability to automate typical bioinformatics analyzes and searches using self programmed scripts using specialized libraries.

SUBJECT EDUCATIONAL EFFECTS

Relating to knowledge:

- PEU_W01 Knowledge on the availability and scope of information available in databases of sequences, sequence families, genes, genomes, biochemical and medicinal databases.
- PEU_W02 Knowledge on the theoretical foundations of methods employed to compare sequences and analyze their features and function, necessary to evaluate the statistical significance of the results.
- PEU_W03 Knowledge on the theoretical foundations of methods used for comparative analyzes, their advantages, disadvantages and scope
- PEU_W04 Knowledge of the contemporary methods of structure prediction

Relating to skills:

- PEU_U01 Ability to construct complex queries against the databases of sequence, gene, genome, biochemical and medicinal information to retrieve the specific data or sequences
- PEU_U02 Ability to search sequence databases for similar sequences, including searches with similarity profiles, to identify homologs
- PEU_U03 Ability to calculate, edit and employ multiple sequence alignments to recognize features, functions, structure and phylogenesis and other comparative analyzes of biosequence database
- PEU_U04 Ability to automate common bioinformatics tasks and analyses using scripting programming languages and specialized libraries

PROGRAMME CONTENT Number of Lecture hours Bioinformatics databases 2 Lec 1 Lec 2 Efficient use of databases: annotations, organization of information, complex 2 queries against specified record fields. Similarity and homology. Evaluation and interpretation of sequence similarity and 2 Lec 3 of its statistical significance. Methods of sequence alignments. Theoretical basis of methods of similarity-based 2 Lec 4 database searches. Approaches to multiple sequence alignment problem. 2 Lec 5 Similarity profiles as representation of sequence similarity and features. Families 2 Lec 6 of sequences and databases of families of homologs. Database queries using sumilarity profiles. Introduction to Bayesian statistics and interpretation of information encoded in 2 Lec 7 bilogical sequences Hidden Markov Models, machine learning methods and stochastic optimization 2 Lec 8 approaches – applications in bioinformatics. Theoretical models and calculation of evolutionary distances. 2 Lec 9 Lec 10 Methods of molecular phylogenetic analysis: inferring relations and mutation 2 history among related sequences Structure prediction methods, model evaluation and optimization 2 Lec 12 Automation of common bioinformatics tasks and analyses: bioinformatics 2 programming APIs and libraries Lec 13 Automation of sequence analysis, structure prediction and other tasks

Lec 14	Contemporary research, analytic and diagnostic techniques.	4
	Total hours	30
	Laboratory	Number of hours
Lab 1	Introduction to course topics, organization of the lab, required software. Introduction to NCBI databases and the Entrez System	2
Lab 2	Complex queries. Available sequence, gene, genome and secondary databases. Different search engines.	2
Lab 3	Uniprot KB, Protein Data Bank, Brenda Enzymes and a selection of other databases related to biochemistry and medicine.	2
Lab 4	Individual task #1	2
Lab 5	Searching for similar sequences using BLAST variants. Interpretation of results.	2
Lab 6	Searching for remote homology using similarity profiles	2
Lab 7	Individual task #2	2
Lab 8	Calculation, analysis, verification and visualization of multiple sequence alignments	2
	Use of Python scripting language and Biopython library for automation of database queries and calculations	2
Lab 10	Phylogenetic analysis	2
Lab 11	Statistical evaluation of results using bootstrap analysis	2
Lab 12	Individual task #3	2
Lab 13	Protein structure prediction based on templates.	2
	Ab initio protein structure prediction. Evaluation of models.	2
Lab 15	Individual task #4	2
	Total hours	30
	TEACHING TOOLS USED	

N1. Lecture with multimedia presentation

- N2. Problem solving
 N3. Use of specialized software
- N4. Preparation of reports of individial tasks, with analysis of results

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	Learning outcomes code	Way of evalu	nating learning outcomes achievement			
F1_Lab	PEU_U01	Individual ta	ask #1			
F2_Lab	PEU_U02	Individual ta	ask #2			
F3_Lab	PEU_U03, PEU_U05	Individual ta	ask #3			
F4_Lab	PEU_U04, PEU_U05	Individual ta	ask #4			
P_Lecture: grade based on exam score		Score	Grade			
P_Lab: grade based on total score		50-59,99%	3,0			
F1_Lab+F2_Lab+F3_Lab	+F4_Lab	60-69,99%	3,5			
			4,0			
		80-89,99%	4,5			
		90-100%	5,0			
PR	PRIMARY AND SECONDARY LITERATURE					

PRIMARY LITERATURE:

- [1] S.Q. Ye, Bioinformatics. A practical approach, Chapman & Hall/CRC, 2008
- [2] I. Eidhammer, I. Johanssen, W.R. Taylor, Protein Bioinformatics an algorythmic approach to sequence and structure analysis, Wiley, 2004
- [3] P.E. Bourne & H. Weissig (ed.), Structural Bioinformatics, Wiley, 2003
- [4] A.D. Baxevanis, B.F.F. Oullette, Bioinformatics, Wiley, 2001

SECONDARY LITERATURE:

- [1] The National Center for Biotechnology Information (NCBI) Handbook: https://www.ncbi.nlm.nih.gov/books/NBK21101/
- [2] Documentation of used WWW services (available online)
- [3] http://www.ncbi.nlm.nih.gov/guide/training-tutorials/

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Paweł Kędzierski, Pawel. Kedzierski@pwr.edu.pl

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish
Name of subject in English
Main field of study (if applicable):
Bionanotechnology
Bionanotechnology
Bioscences
Bioinformatics

Profile: academic

Level and form of studies: 2nd level

Kind of subject: obligatory

Subject code W03BSS-SM2011W, W03BSS-SM2011S

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30				15
Number of hours of total student workload (CNPS)	50				25
Form of crediting	Examination				crediting with grade
For group of courses mark (X) final course					
Number of ECTS points	2				1
including number of ECTS points for practical (P) classes					1
including number of ECTS points for direct teacher-student contact (BK) classes	1,3				0,7

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- Basic knowledge of physical chemistry (1st level)
 Basic knowledge of biochemistry (1st level)
 Basic knowledge of molecular dynamics (2nd level) 1.
- 2.3.

	SUBJECT OBJECTIVES				
C1	C1 Principles underlying the functioning of molecular machines in biology				
C2	Basic knowledge about methods utilized in bionanotechnology to design, synthesize and analyze bionanomachines				
С3	Practical knowledge on how to perform basic molecular dynamics (MD) simulations to solve problems in bionanotechnology				
C4	Basic knowledge on the recent achievements in bionanotechnology				

SUBJECT EDUCATIONAL EFFECTS

Relating to knowledge:

- PEU_W01 Basic concepts of nanobiotechnology and bionanotechnology
- PEU W02 Principles of functioning of molecular machines in biology
- PEU_W03 Basic knowledge on experimental methods used in design, synthesis and analysis in bionanotechnology
- PEU_W04 Basic knowledge on experimental methods used in structural investigation in bionanotechnology
- PEU_W05 Basic techniques in designing synthetic bionanomachines
- PEU_W06 Basic principles of functioning of bionanomachines
- PEU_W07 Basic knowledge on molecular modelling tools used in designing bionanomachines
- PEU_W08 Basic knowledge on the recent achievements in modern bionanotechnology

Relating to skills:

- PEU_U01 Practical knowledge on how to prepare input files and how to perform minimization and MD of nanopore
- PEU_U02 Practical knowledge of performing basic MD simulations of DNA within the nanopore
- PEU_U03 Practical knowledge on how to prepare and present a seminar on the last achievements in bionanotechnology

	PROGRAMME CONTENT				
	Form of classes - lecture	Nu			
Lec1	Basic concepts. Nonotechnology, biotechnology, bionanotechnology, nanobiotechnology. Feynman's idea. Top-down and bottom-up appoaches. Milestone achievements in bionanotechnology. Nanobiotechnology/bionanotechnology in electronics, informatics, energetics, army, agriculture and food technology – examples.	2			
Lec2	How do molecular machines work in biology?:Properties of particles at macro- and nano-levels. Bionanomolecules in water environment – hydrophobic effect. Proteins as a structural material in bionanotechnology. Limitations of natural bionanomolecules.	2			
Lec3	Methods in bionanotechnology: to design, synthesize and analyze. Rekombinant DNA technology. DNA clonning. PCR method. Protein synthesis in vitro. Directed mutagenesis. Fusion and chimeric proteins. Monoclonal antibodies.	2			
Lec4	Methods in bionanotechnology: to design, synthesize and analyze – part 2. X-ray and NMR methods to investigate structure of biomolecules. Electron spectroscopy methods: TEM, SEM, tomography. AFM method. Molecular modelling as a tool to obtain information on structure and dynamics of biomolecule.	2			
Lec5	Design of nanomachines. Methods used in bionanomachines design: sequential covalent bond formation, polimeryzation, self-organizatoin and aggregation. Protein folding. Role of chaperones in folding. Proteins stable in high temperatures. How to make a protein more rigid? How to introduce a disorder in a protein? Symmetric and quasi-symmetric complexes.	2			
Lec6	Functional aspects of biomoleculs . Energy transfer in natural bionanomachines. Electron transfer in natural bionanomachines. Light-driven molecular bionanomachines. Charge transfer in biosystems. How do enzymes work? Methods to control bionanomachines – allosteric regulation and covalent modyfication.	2			

Lec7	Design of bionanomachines. De novo protein design. Enzyme design based on molecular modelling methods. Design of biosystems having specific spectral properties. PNA (Peptide Nucleic Acid) vs. DNA.	2	
Lec8	Exam	2	
Lab1	DNA sequencing using MD – part 1 . Construction of a cystal membrane of Si ₃ N ₄ . Synthetic nanopore in Si ₃ N ₄ .membrane.	2	
Lab2	DNA sequencing using MD – part 2. Calibration of force field to reproduce experimental value of dielectric constant.	2	
Lab3	DNA sequencing using MD – part 3 . Solvation of a nanopore.	2	
Lab4	DNA sequencing using MD – part 4. Energy minimization. Molecular dynamics under constant pressure. Measuring ionic current in nanopores.	2	
Lab5	5 DNA sequencing using MD – part 5 . Simulating the process of DNA transport through a nonopore.		
Lab6	DNA sequencing using MD – part 6. Ionic current in nanopores in the presence of DNA. Comparison of ionic current with/without DNA in the system.	2	
Lab7	DNA sequencing using MD – part 7. Transporting DNA through nanopore – MD simulation. Transporting ubiquitin through nanopore – MD simulation.	2	
	Total hours	30	

	Form of classes - seminar	Nu
Se1-15	Students in the form of oral contribution present and discuss the late achievements and trends in bionanotechnology based on the most recent scientific literature. The list of possible topics is upgraded every year due to the very rapid progress in this field.	15
	Total hours	15

	TEACHING TOOLS USED			
N1	N1 Lecture with multimedia presentation			
N2	Practical usage of software			
N3	Preparation of reports			
N4	Seminar presentation			

EVALUATION	OF SUBJECT EDUCA	TIONAL EFFECTS ACHIEVEMENT
Evaluation F – forming (during semester), C – concluding (at semester end)	Educational effect number	Way of evaluating educational effect achievement
F1 (lecture)	PEU_W01 – PEU_W07	Written exam
F2 (lecture)	PEU_U01 – PEU_U02	Report
P (seminar)	PEU_U03, PEU_K01	Seminar presentation
P (lecti	3.5 if (F1 + F 4.0 if (F1 + F 4.5 if (F1 + F 5.0 if (F1 + F	= 50-60% max. no of poins. (2) = 61-70% max. no of poins. (2) = 71-80% max. no of poins. (2) = 81-90% max. no of poins. (2) = 91-99% max. no of poins. (2) = 100% max. no of poins.

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

[1] D.S. Goodsell "*Bionanotechnology: Lessons from nature*" Plenty of room for biology at the bottom: An introduction to bionanotechnology", Wiley-Liss, 2004.

SECONDARY LITERATURE:

- [1] *Bionanotechnology: Proteins to Nanodevices*, Eds. V. Renugopalakrishnan, R.V.Lewis, Springer, 2006.
- [2] *Nanobiotechnology: Concepts, Applications and Perspectives*, Eds. C.M.Niemeyer, C.A.Mirkin, Wiley-VCH, 2004.
- [3] *Nanobiotechnology II: More Concepts and Applications*, Eds. C.M.Niemeyer, C.A.Mirkin, Wiley-VCH, 2007.
- [4] E. Gazit "Plenty of room for biology at the bottom: An introduction to bionanotechnology", Imperial College Press, 2007.

SUBJECT SUPERVISOR

(NAME AND SURNAME, E-MAIL ADDRESS)

prof. dr hab. Tadeusz Andruniów, tadeusz.andruniów@pwr.wroc.pl

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Genomika obliczeniowa Name of subject in English Computational genomics Main field of study (if applicable): Biosciences

Specialization (if applicable): Bioinformatics

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03BSS-SM2014W, W03BSS-SM2014L

Group of courses NO

	Lecture	Classe s	Laboratory	Projec t	Semina r
Number of hours of organized classes in University (ZZU)	15		15		
Number of hours of total student workload (CNPS)	25		25		
Form of crediting (Examination / crediting with grade)	Examinati on		crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points	1		1		
including number of ECTS points for practical classes (P)			1		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)	0,65		0,7		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Basic bioinformatics, genetics and molecular biology knowledge
- 2. Basic knowledge of computer science
- 3. Specialized English

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SUBJECT OBJECTIVES

- C1 Teaching elementary topics in genomics.
- C2 Introducing main genomics databases.
- C3 Familiarizing students with methods of sequencing, assembling and description of genomes.
- C4 Familiarizing students with comparative genomics methods and applications.
- C5 Introducing main concepts and methods used in transcriptomic research.
- C6 Teaching about practical applications of genomics research and genomic information.
- C7 Acquainting students with the ethical aspects of genomics research and the use and safety of genomic information.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

PEU_W01 – knowledge of the basic concepts in genomics;

PEU_W02 – knowledge of the content and organization of genomic databases;

PEU_W03 – knowledge of the genome mapping, sequencing, assembly and description methods;

PEU_W04 – knowledge of the tools used to analyze and compare genomic sequences;

PEU_W05 – knowledge of the methods used in transcriptomic and their applications;

PEU_W06 – knowledge of the possible use of genomic information.

relating to skills:

PEU_U01 – ability to search the genomic databases and retrieve information from such sources;

PEU_U02 – ability to select appropriate methods and tools for the studied problem;

PEU_U03 – ability to conduct basic manipulations, comparisons and analysis on genomic information;

PEU_U04 – ability to perform the quality control and genome assembly using sequencing data;

PEU_U05 – ability to conduct the basic analysis and visualization of transcriptomic data:

PEU_U06 – ability to analyze the obtained results.

relating to social competences:

PEU_K01 - awareness of the ethical aspects of genomics research and challenges associated with data protection.

PROGRAMME CONTENT	
Lecture	Number of hours

		1
Lec 1	Introductory lecture : the plan and content of the course and crediting rules. Introduction of the basic concepts in genomics, historical background of genomic research, applications and perspectives.	2h
Lec 2	Genomic information organization : sources of genomic information and principles of data collection and access. Introduction of genomic databases and data structure.	2h
Lec 3	Assembly of genome sequences : presentation of the gene assembly process based on homology approach and <i>de novo</i> . Introduction to the procedures and methods used for quality control and assembly of genome sequences.	
Lec 4	Structural genomics and description of genomes : overview of principles and methods of genome mapping including types of genomic maps. Presentation of main rules and methods of genes prediction and genome annotations.	
Lec 5	Functional and comparative genomics : the types of data gained from transcriptomic experiments, approach to the transcriptomic data analysis, presentation and applications. Overview of the comparative genomics methods together with applications examples.	2h
Lec 6	Experimental techniques : presentation of main experimental techniques used for the exploration of genomes including new generation techniques. Discussion of the application possibilities and the future of these field.	
Lec 7	Ethical aspects of genomic research: the ethical aspects of genomic research, the use of genomic information in science and other fields and challenges of data safety. Law regulations regarding the genomic information.	
Lec 8	3 Written exam	
	Total hours	15h

	Laboratory				
Lab 1	Lab 1 Introductory classes : the program of laboratory classes, organization and rules of the computer lab. Overview of basic tools and software used during the course. Introduction to the Ensembl genome browser.				
Lab 2	Lab 2 Genomic databases : introduction to the main genomic databases, data organization and visualization. Overview of related 'omics' databases.				
Lab 3	Lab 3 Genomic databases; genome description : Practical examples reflecting the genome annotation, including analysis of known transcript or variants. The use of genomic databases as a source of information including basic comparative analysis.				
Lab 4	Lab 4 Project I: Practical individual tasks for the first report.				
Lab 5	Lab 5 Genome information analysis: Practical examples of large-scale genomic data retrieving, handling, sorting, comparing, etc., using genomic databases and online tools.				

Lab 6	Genome assembly: Introduction to genome sequencing data (reads) quality control and genome assembly. Practical examples.	2h
Lab 7	Transcriptomics: Practical examples of transcriptomic data analysis and methods of visualization of the results.	2h
Lab 8	Project II: Practical individual tasks for the second report.	1h
	Total hours	15h

TEACHING TOOLS USED

- N1. Lecture
- N2. Multimedia presentation
- N3. Practical usage of databases
- N4. Practical usage of software
- N5. Tutorials with examples for analyzed problems
- N5. Solving individual tasks
- N6. Preparation of reports

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	Learning outcomes code	Way of evaluating learning outcomes achievement
P (Lecture)	PEU_W01- PEU_W06, PEU_U01, PEU_U02, PEU_K01	Final exam
F1 (Laboratory)	PEU_W01-PEU_W02, PEU_U01- PEU_U03	Report from the Individual Project I
F2 (Laboratory)	PEU_W01- PEU_W05, PEU_U01- PEU_U06	Report from the Individual Project II
P (Laboratory) = F1 -	+ F2	•

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] Fundamentals of Bioinformatics and Computational Biology, G.B. Singh, Springer-Verlag London, 2015.
- [3] Introduction to Genomics, Lesk A. Oxford University Press, Oxford, 2017.
- [2] Big Data Analytics in Genomics, Wong, Ka-Chun, Springer-Verlag London, 2016.

SECONDARY LITERATURE:

- [1] Comparative Gene Finding, Models, Algorithms and Implementation, M. Axelson-Fisk, Springer-Verlag London, 2015.
- [2] Genomes, T. A. Brown, 4th Edition, Garland Science: New York, 2017.

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Renata Grzywa, PhD, renata.grzywa@pwr.edu.pl

FACULTY of CHEMISTRY

SUBJECT CARD

Name of subject in Polish Krystalografia i struktura ciał stałych Name of subject in English Crystallography and structure of solids

Main field of study (if applicable): Biosciences

Specialization (if applicable): Medicinal Chemistry

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03BSS-SM2018W, W03BSS-SM2018C

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30	15			
Number of hours of total student workload (CNPS)	50	25			
Form of crediting (Examination / crediting with grade)	crediting with grade	crediting with grade			
For group of courses mark (X) final course					
Number of ECTS points	2	1			
including number of ECTS points for practical classes (P)		1			
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)		0.7			

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

1. General knowledge of mathematics, physics and chemistry.

SUBJECT OBJECTIVES

- C1 Knowledge of the structure, symmetry and diffraction of macro-, micro- and nanocrystals.
- C2 Knowledge of directions of development of crystallography.
- C3 Understanding data in crystallographic papers.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

A person who has completed the course:

PEU_W01 has knowledge of the structure and symmetry of crystals.

PEU_W02 understands the international symbols and graphical representation of space groups and the international symbols of crystal classes.

PEU_W03 knows the relationships between a diffraction pattern and crystal structure.

PEU W04 has knowledge of directions of development of crystallography.

relating to skills:

A person who has completed the course:

PEU_U01 is able to study scientific literature on crystal structures and evaluate crystal data.

related to social competences:

A person who has completed the course:

PEU_K01 is able to take part in discussions on crystallographic structural studies.

PEU_K02 understands the importance of crystallography in science and industry.

PROGRAMME CONTENT Number Lecture of hours 2 Lec 1 The historical and current definitions of crystals and crystallography. The internal structure of crystals. A crystal lattice, row lines, lattice planes, Miller symbols, a unit cell and cell types. The mosaic structure of real crystals, dislocations. The internal symmetry of crystals. Symmetry elements and operations. 4 Lec 2, Relationships between the internal and external symmetry of crystals. Lec 3 Crystal systems *vs* symmetry. Crystal systems and cell parameters. The conventional choice of unit cells. 2 Lec 4 The Bravais unit cells. Space groups: international symbols and graphical representations. An 2 Lec 5 asymmetric unit cell. Relationships between the symbol of a space group and the symbol of a 2 Lec 6 point group (crystal class). The types of point groups. Examples of crystal structures. Crystallographic databases. 2 Lec 7 X-rays: properties and sources. Synchrotron radiation: sources of the first, 2 Lec 8 second, third and fourth generations and properties. Synchrotron crystallographic studies. Lec 9. The directions and intensities of diffracted beams. Factors influencing the 4 Lec 10 directions and intensities. The phase problem. Diffraction pattern vs internal structure and symmetry of crystals. Lec 11 Neutronography and electronography vs roentgenography. Crystallographic 2 information files (cif). Nanocrystals. The quantitative and qualitative definition. The internal 4 Lec 12, structure of nanocrystals vs macrocrystals. Defects. External appearance. Lec 13 Diffraction in nanocrystals vs diffraction in microcrystalline materials. The broadening and shifting of peaks in powder diffraction patterns. Apparent lattice parameters: determination and influencing factors. Properties. Synchrotron crystallographic studies of nanocrystals. Lec 14 Quasi crystals: 1D, 2D and 3D-dimensional. Internal and external structure. 2 Diffraction. Properties. Lec 15 Crystallographic data in scientific papers. 2 30 Total hours

	Classes	Number of hours
Cl 1	The preliminary class.	1
C1 2	Lattice points, row lines, lattice planes.	1
Cl 3, Cl 4, Cl 5	Symmetry elements: an inversion center, a mirror plane, rotation axes, rotoinversion axes.	3
Cl 6, Cl 7	Screw axes and glide planes.	2
Cl 8	Bravais lattices.	1
Cl 9	Partial test I	1
Cl 10, Cl 11	Systematic absences.	2
Cl 12, Cl 13	Crystal classes: symbols and graphical representation.	2
Cl 14	Physical properties of crystals.	1
Cl 15	Partial test II	1
	Total hours	15
	THE ACTION OF THE ACTION	<u> </u>

TEACHING TOOLS USED

- N1. A multimedia presentation
- N2. Crystallographic models
- N3. A blackboard

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming	Learning outcomes	Way of evaluating learning outcomes
0 //	code	achievement
concluding (at semester		
end)		
F1 (lectures)	PEU_W01, PEU_W02,	partial test I
	PEU_K01-K02	
F2 (lectures)	PEU_K01-K02,	partial test II
	PEU_W03, PEU_W04	
F3 (classes)	PEU_W01 PEU_W02	partial test I
F4 (classes)	PEU_W03, PEU_U01	partial test II
P1=(F1+F2)/2		
P2=(F3+F4)/2		

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] P. Luger, Modern X-Ray Analysis on Single Crystals, de Gruyter, Berlin, 2014.
- [2] R. J. D. Tilley, Crystals and Crystal Structures, John Wiley & Sons Ltd, Chichester, 2006.

SECONDARY LITERATURE:

- [1] C. Giacovazzo, H. L. Monaco, G. Artioli, D. Viterbo, G. Ferraris, G. Gilli, G. Zanotti, M. Catti, Fundamentals of crystallography, C. Giacovazzo Ed., Oxford, 2011.
- [2] International Tables for Crystallography, Volume A, Springer, 2005; Willey 2016.

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

team

FACULTY of Chemistry

SUBJECT CARD

Name of subject in Polish – Eksploracja Danych Name of subject in English – Data Mining Main field of study (if applicable): Biosciences Specialization (if applicable): Bioinformatics

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory Subject code W03BSS-SM2009L

Group of courses NO

_	T .			T	1
	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)			15		
Number of hours of total student workload (CNPS)			25		
Form of crediting (Examination / crediting with grade)			crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points			1		
including number of ECTS points for practical classes (P)			1		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			0,7		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Computer skills
- 2. Knowledge of the programming basics

SUBJECT OBJECTIVES

- C1 Understand the applications of data mining methods to biological data.
- C2 Learn how to analyze the results of an experiment using learnt methods.

SUBJECT EDUCATIONAL EFFECTS

relating to skills:

PEU_U01 Proficiency in basic concepts of data exploration, data visualization, diverse data mining techniques, and the application of results in real-world contexts.

relating to social competences:

PEU_K01 Effective communication and the integration of data-driven insights into decision-making processes.

PROGRAMME CONTENT

Laboratory	Number of hours
La1 Introduction to Data Mining	1

La2	Data Preparation and Cleaning	2
La3	Data Exploration and Visualization	2
La4	Data Mining Techniques	2
La5	Classification Algorithms in Data Mining	2
La6	Clustering Algorithms in Data Mining	2
La7	Evaluation and Validation metrics	2
La8	Knowladge evalutaion – end semester project	2
	Total hours	15

TEACHING TOOLS USED

- N1. Computer Lab
- N2. Presentation with elements of live coding
- N3. Consultations
- N4. Independent additional studies

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

	Learning outcomes code	Way of evaluating learning outcomes achievement
F		Grade based on the assessment of the final project completed during the laboratory.
P = F		

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] [Python for data analysis (1st. ed.)., McKinney Wes. 2012., O'Reilly Media, Inc.
- [2] Introduction to Data Mining, Pang-Ning Tan, Michael Steinbach, Anuj Karpatne, Vipin Kumar, 2019, Pearson.
- [3] The StatQuest Illustrated Guide To Machine, Josh Starmer, 2022, StatQuest Publications

SECONDARY LITERATURE:

[1] Internet resources

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Dr inż. Wojciech Wojtowicz wojciech.wojtowicz@pwr.edu.pl

FACULTY of CHEMISTRY					
SUBJECT (Name of subject in PolishLeki nieorganiczne. Name of subject in EnglishInorganic drugs Main field of study (if applicable):Bioscien Specialization (if applicable):Medicinal Chen Profile: academic Level and form of studies: 2nd level, full-time Kind of subject: obligatory Subject code W03BSS-SM2025W Group of courses: NO	ces	•••••	••••••		••••
	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	15				
Number of hours of total student workload (CNPS)	25				
Form of crediting (Examination / crediting with grade)					
For group of courses mark (X) final course					
Number of ECTS points	1				
including number of ECTS points for practical classes (P)					
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)					
*delete as not necessary					
PREREQUISITES RELATING TO KNO	WLED	GE, SK	ILLS AND	OTHE	R

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

1. Principles of inorganic chemistry.

2.

3.

SUBJECT OBJECTIVES

- C1 To provide students with inorganic biologically active compounds and their influence on human metabolism.
- C2 To provide students with issues regarding the use of inorganic compounds in the field of medicine and pharmacy.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

PEU_W01 – the student has general knowledge about metal-based inorganic drugs and metal-based diagnostic agents and knows the basic concepts in the field of inorganic

medicinal chemistry.

PEU_W02 – the student knows the structure of commonly used inorganic drugs and their physicochemical properties, reactivity, and mechanism of their action.

PEU_W03 – has general knowledge about current development directions and the latest discoveries regarding the use of inorganic compounds in therapy and diagnostics, PEU_W04 – can distinguish particular groups of inorganic drugs and determine their use and therapeutic effect.

	PROGRAMME CONTENT	
	Lecture	Number of hours
Lec 1	Medicinal inorganic chemistry: state of the art. Classification of metal-based drugs according to their mechanisms of action (essential elements, therapeutic agents, radiopharmaceuticals, metallomics, chelation therapy, enzyme mimics, contrast agents, protein/enzyme regulators). Design of therapeutic and diagnostic agents.	2
Lec 2	The concept of bond theory in medicinal inorganic chemistry: nomenclature, coordination geometry, chelating ligands, isomerism, kinetic and thermodynamic stability.	2
Lec 3	Metal compounds as therapeutic agents. (antibacterial and antiviral agents, antiparasitic drugs, antiarthritic drugs, antimalarial drugs, treatment of diabetes and obesity, redox-active metal-based mediators).	3
Lec 4	Metal-related metabolic disorders. (controversial drugs, heavy-metal poisoning, chelation therapy).	1
Lec 5	Medical diagnostics with the use of inorganic complexes and radioisotopes (MRI, MRA, PET, SPECT).	2
Lec 6	Discovery of cisplatin, synthesis, its mechanism of anticancer activity and the path to obtaining next generations of drugs based on platinum.	2
Lec 7	Search for non-platinum anticancer drugs with the interesting biological properties (drugs based on: Pd, Ti, Ga, As, Ru, Bi, V, Au).	2
Lec 8	Final test.	1
	Total hours	15
	TEACHING TOOLS USED	

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F –	Learning outcomes code	Way of evaluating learning outcomes
forming during		achievement
semester), P –		
concluding (at		
semester end)		
P	PEU_W01-PEU_W04	Final test

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] E. Alessio (Ed.) Bioinorganic Medicinal Chemistry, Wiley-VCH, 2011
- [2] K. A. Strohfeldt, Essentials of Inorganic Chemistry for Students of Pharmacy, Pharmaceutical Sciences and Medicinal Chemistry, Wiley, 2015,
- [3] J.C. Dabrowiak Metals in Medicine. Wiley, 2009.

SECONDARY LITERATURE:

- [4] Nicholas P. Farrell, Uses of inorganic chemistry in medicine, RSC, 1999.
- [5] EudraLex, The Rules Governing Medicinal Products in the European Union, Volume 4, EU Guidelines for Good Manufacturing Practice for Medicinal Products for Human and Veterinary Use, European Commission, health and consumers directorate-general, Ref. Ares(2012)778531 - 28/06/2012.
- [6] J.L.Sessler, S.R.Doctrow, T.J.McMurry, S.J.Lippard, Medicinal Inorganic Chemistry 2005.
- [7] Metallopharmaceuticals I, DNA Interactions Eds. M.J. Clarke, P.J. Sadler (1999).
- [8] Metallopharmaceuticals II, Diagnosis and Therapy. Eds. M.J. Clarke, P.J. Sadler (1999).

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dr inż. Magdalena Malik, magdalena.malik@pwr.edu.pl

FACULTY of CHEMISTRY

SUBJECT CARD

Name of subject in Polish Wprowadzenie do Statystyki
Name of subject in English Introductory Statistics

Main field of study (if applicable): Biosciences

Specialization (if applicable): Medicinal Chemistry

Profile: academic / practical*

Level and form of studies: 1st/ 2nd level, uniform magister studies*, full-time / part-time*

Kind of subject: obligatory / optional / university-wide*

Subject code W03BSS-SM2017C

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)		15			
Number of hours of total student workload (CNPS)		50			
Form of crediting (Examination / crediting with grade)		crediting with grade			
For group of courses mark (X) final course					
Number of ECTS points		2			
including number of ECTS points for practical classes (P)		2			
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)		0.7			

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Basic knowledge of mathematical calculations, linear algebra.
- 2. Basic ability to use a spreadsheet software.

SUBJECT OBJECTIVES

- C1 Acquainting the student with the basics of descriptive statistics and possibilities of its practical usage.
- C2 Acquainting the student with possibilities of mathematical models utilization in analysis and interpretation of data.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

PEU_W01 Student has a basic knowledge in the area of descriptive statistics.

PEU_W02 Student has information about data analysis methods with aid of statistics.

relating to skills:

PEU_U01 Student is able to solve basic problems from the descriptive statistics field and is able to present experimental data sets in an appropriate way.

relating to social competences:

PEU_K01 Student is able to present and explain the results of the completed project.

	PROGRAMME CONTENT			
	Classes	Number of hours		
Cl 1	Introduction to the basic subjects of descriptive statistics. Types of data sets.	2		
Cl 2	Methods of experimental data processing and its analysis.	2		
C1 3	Numerical and graphical representation of the statistical data.	2		
Cl 4	Confidence intervals and statistical hypothesis testing. Student's t-test.	2		
Cl 5	Data distribution functions and its utilization.	2		
Cl 6	Correlation analysis of experimental data.	2		
Cl 7	Usage of ANOVA tests in data analysis.	2		
Cl 8	Analysis of common errors and application of improvements.	1		
	Total hours	15		
	TEACHING TOOLS USED			

- N1. Multimedia presentation.
- N2. Solving project tasks with mathematical and statistical calculations software.
- N3. Project with usage of *Design thinking* method.

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	Learning outcomes code	Way of evaluating learning outcomes achievement
F1	PEU_W01, PEU_U01	Sprawozdanie 1
F2	PEU_W01, PEU_W02, PEU_U01	Sprawozdanie 2
F3	PEU_W01, PEU_W02, PEU_U01, PEU_K01	Sprawozdanie 3
P = (F1 + F2 + F3) / 3	•	
	PRIMARY AND SECONDA	RV LITERATURE

PRIMARY LITERATURE:

- [1] A. Agresti, C. A. Franklin, Statistics: the art and science of learning from data, Pearson Prentice Hall, Upper Saddle River, 2007,
- [2] T. Hill. P. Lewicki, Statistic: methods and applications: a comprehensive reference for science, industry and data mining, StatSoft, Tulsa, 2006.

SECONDARY LITERATURE:

[1] L. Rogers, D. Willoughby, Numbers: data and statistics for the non-specialist, HarperCollins Publishers, London, 2013.

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

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FACULTY of CHEMISTRY

KARTA PRZEDMIOTU

Name of subject in Polish Izolacja i identyfikacja bioproduktów Name of subject in English Isolation and identification of bioproducts

Main field of study (if applicable): Biosciences

Specialization (if applicable): Medicinal Chemistry

Profile: academic

Level and form of studies: 2nd level

Kind of subject: obligatory Subject code W03BSS-SM2016L

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of					
organized classes in			30		
University (ZZU)					
Number of hours of					
total student workload			50		
(CNPS)					
Form of crediting			Craditina		
(Examination /			Crediting		
crediting with grade)			with grade		
For group of courses					
mark (X) final course					
Number of ECTS			2		
points					
including number of					
ECTS points for			2		
practical classes (P)					
including number of					
ECTS points					
corresponding to					
classes that require			1,4		
direct participation of					
lecturers and other					
academics (BU)					

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Knowledge of organic chemistry at the university level.
- 2. Knowledge of analytical chemistry at the university level.
- 3. Proficiency in practical work in an organic chemistry laboratory.
- 4. Familiarity with basic techniques for identifying chemical compounds in mixtures.

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SUBJECT OBJECTIVES

- C1 Familiarization with the classification of chromatographic methods.
- C2 Familiarization with the operation and software of gas chromatography.
- C3 Understanding the impact of chromatographic experiment parameters on the separation of organic compounds.
- C4 Familiarization with issues related to qualitative and quantitative analysis.
- C5 Learning methods for identifying compounds released into the environment.
- C6 Introduction to the basics of thin-layer chromatography.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

A person who has passed the subject:

PEU_W01 - knows the classification of chromatographic methods and the principles of chromatographic separation.

PEU_W02 - knows the types of applications of chromatographic techniques in various fields of science.

PEU_W03 - understands the operating principle of analytical equipment.

PEU_W04 - can plan a scientific experiment.

relating to skills:

A person who has passed the subject:

PEU_U01 - can perform analyses using analytical equipment.

PEU_U02 - can conduct a scientific experiment.

PEU_U03 - can determine the concentration of organic compounds in an unknown sample using analytical equipment.

PEU_U04 - can prepare a report from the experiment in the form of a scientific article.

Z zakresu kompetencji:

Osoba, która zaliczyła przedmiot:

PEU_K01 – uznaje znaczenie wiedzy w rozwiązaniu problemów w zakresie identyfikacji bioproduktów

	PROGRAMME CONTENT - laboratory	Number of hours
La1	Overview of the curriculum and assessment methods. Safe working conditions in a chemical laboratory. Description of basic working tools. Proposal for a scientific project topic.	4
La2	Gas chromatography. Preparation of a method for the initial qualitative analysis. Impact of temperature and flow on the separation of volatile organic compounds. Qualitative analysis of a natural compound solution. Quantitative analysis. Creation of a calibration curve for a natural compound. Determination of concentration in an unknown sample.	4
La3	Implementation of a scientific project. Independent work.	4
La4	Implementation of a scientific project. Independent work.	4
La5	Implementation of a scientific project. Independent work.	4
La6	Implementation of a scientific project. Independent work.	4

La7	Implementation of a scientific project. Independent work.	4
La8	Final assessment class. Presentation of project results in the form of an article in the format of an international scientific journal.	2
	an article in the format of an international scientific journal.	
	Total hours	30

TEACHING TOOLS USED

N1 Working with a computer using scientific and patent databases.

N2 Independent experimental work in the field of chromatographic techniques.

OCENA OSIĄGNIĘCIA PRZEDMIOTOWYCH EFEKTÓW UCZENIA SIĘ

Evaluation (F –	Learning outcomes	Way of evaluating learning outcomes
forming during	code	achievement
semester), P –		
concluding (at		
semester end)		
P (laboratory)	PEU_W01-	Written assessment paper
	PEU_W04	
	PEU_U01-	
	PEU_U04,	
	PEU_K01	

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- 1. J.L. Anderson *et al. Analytical Separation Science*, vol. 3 Wiley-VCH Verlag, Weinheim, **2015**;
- 2. Anonymous (University of California Davis) *Thin Layer Chromatography*, LibreTexts: https://chem.libretexts.org/Core/Analytical_Chemistry/Lab_Techniques/Thin_Layer_Chromatography; ostatnia modyfikacja: **16.02.2017**
- 3. Lecture 3 Thin layer chromatography | MIT 5.301 Chemistry Laboratory Techniques, IAP **2004**; access: Massachusetts Institute of Technology OpenCourseWare https://www.youtube.com/watch?v=EUn2skAAjHk
- 4. K. Thet, N. Woo, *Gas Chromarography*. LibreTexts; https://chem.libretexts.org/Core/Analytical_Chemistry/Instrumental_Analysis/Chromatography/Gas_Chromatography
 Last modification: **13.03.2015**
- 5. A. Wesołowska *et al.* Comparison of chemical compositions of essential oils isolated by hydrodistillation from wild thyme (*Thymus serpyllum* L.) with use of Deryng and Clevenger apparatus. *herba polonica*, **2014**, 60(2), DOI: 10.2478/hepo-2014-0006

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Dr inż. Daniel Strub, daniel.strub@pwr.edu.pl

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Uczenie maszynowe w chemii i biologii
Name of subject in English Machine Learning for Chemistry and Biology

Main field of study (if applicable): Bioscences

Specialization (if applicable): Profile: academic / practical*

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03BSS-SM2013W, W03BSS-SM2013L

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30		30		
Number of hours of total student workload (CNPS)	50		50		
Form of crediting	Examination		crediting with grade*		
For group of courses mark (X) final course					
Number of ECTS points	2		2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			1.4		

*delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Fundamentals of Physical Chemistry
- 2. Understanding the structure of bioorganic molecules
- 3. Fundamentals of mathematical analysis and linear algebra
- 4. Pre-intermediate experience with python scripting

SUBJECT OBJECTIVES

- C1 To familiarize the students with the fundamentals of machine learning and deep learning methods.
- C2 To familiarize the students with possible applications of machine learning models in chemistry and biology.
- C3 Acquiring the ability to identify and apply the most appropriate machine learning methods to solve a given research problem or analyze data.
- C4 Learning how to evaluate the trained models and interpret their results.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

PEU_W01 Knows the basic strategies and algorithms of supervised and unsupervised learning.

PEU_W02 Has knowledge of common applications of machine learning methods in chemistry and biology.

PEU_W03 Is able to assess the strengths, weaknesses and limitations of individual machine learning methods in applications to various problems in the field of computational biology.

PEU_W04 Has knowledge of good practices in training machine learning models to avoid overtraining and identify potential shortcomings in the training data set. PEU_W05 Knows various forms of representation of the structure of bioorganic molecules, including commonly used geometry formats (xyz, pdb, zmat, smiles, smarts, sdf) as well as representations dedicated to machine learning.

PEU_W06 Knows the formats and representations of data that can be used to train machine learning models.

relating to skills:

PEU_U01 Is able to effectively select and prepare a representative data set in the appropriate format for a given machine learning method.

PEU_U02 Can apply supervised learning models for data classification.

PEU_U03 Can apply unsupervised learning models for data clustering.

PEU_U04 Can conceptually/schematically describe an algorithm to solve a given research problem or data analysis problem.

PEU_U05 Can implement an algorithm to solve a given research problem or data analysis problem using the Python scripting language.

PEU_U06 Can evaluate machine learning models and interpret the results they offer.

relating to social competences:

PEU_K01 Students are able to work in a group, performing various roles, including group leader

PEU_K02 Students are aware of the social role of an MSc in Bioinformatics PEU_K03 Students are ready to critically evaluate his knowledge and the received content

PROGRAMME CONTENT		
	Lecture	Number of hours
Lec 1	Introduction to machine learning. Explanation of the term machine learning and its relation to the so-called artificial intelligence. To familiarize students with the general classification of supervised and unsupervised learning methods. An overview of the most popular applications of machine learning in science, engineering, and life sciences.	2
Lec 2	Machine learning datasets. Data sources and representative data formats that can be used for machine learning. Sources of data errors. Good practices in data selection.	2

Lec 3	Supervised learning - artificial neural networks I. A brief history of artificial neural	2
	networks and similarities to biological networks. Research directions and applications of neural networks. Linear networks.	
Lec 4	Supervised learning - artificial neural networks II. Training a neural network using the gradient descent method and back propagation. Rosenblatt perceptron. Multilayer and deep networks. Detailed application examples.	2
Lec 5	Supervised learning - other methods. Support vector machines, kernel ridge regression, decision trees, random forest.	2
Lec 6	Unsupervised learning. Description of the basic methods of unsupervised learning. Classification and grouping. Train the model to recognize features that characterize the data set.	2
Lec 7	Structural biology I. Introduction/review of selected issues in structural biology concerning the structure and dynamics of proteins and nucleic acids. Predicting the secondary structure of peptides from sequences.	2
Lec 8	Structural biology II. Predicting the structure of biomolecules - AlphaFold and nucleic acids.	2
Lec 9	Machine learning models in molecular simulations I. Introduction/review of the elements of computational chemistry. Potential and free energy surfaces. Classification of various methods in computational chemistry including machine learning potentials.	2
Lec 10	Machine learning models in molecular simulations II. Representation of the geometry/structure of molecules in machine learning. Training of models to reproduce the shape of the potential energy surface and selection of the data set. Advantages and disadvantages of neural networks and kernel ridge regression.	2
Lec 11	Machine learning models in molecular simulations III. Learning molecular properties. Non-bonding interactions, oxidation states and electron configurations.	2
Lec 12	Drug design. Interaction of the drug with the active site. Methods of estimating the free energy of active substance binding in the active site.	2
Lec 13	Prediction of synthetic pathways to organic molecules. Reaxys database. SMARTS and SMILES structure formats. Approaches to prediction of organic synthesis pathways using retrosynthesis.	2
Lec 14	Image analysis and medical applications. Examples and methods of analyzing diagnostic images using machine learning	2
Lec 15	Revision of the most important topics presented during the lectures. Preparation for the exam, discussion and questions.	2
	Total hours	30

	Laboratory	Number of hours
Lab 1	Organization of work in the computer laboratory and computing center. Discussion of the principles of occupational health and safety. Account distribution and basic information about available operating systems. Reminder of elements and selected commands of the LINUX operating system. Basic information about the operating system. Using Anaconda and Jupyter Notebooks.	2
Lab 2	Introduction to the basics of statistics using the Pandas module. Tasks: histograms, block plots, exploration of pseudo-random number generation, meshing histograms with Pandas; binomial, Poisson and normal distributions. Introduction to the SciKit-learn library in python.	4

Lab 3	Data visualization and dimensionality reduction - introduction and exercises. Tasks: use of block charts to visualize many variables simultaneously. Correlation analysis between data based on heat maps.	4
Lab 4	Data classification - introduction and exercises. Tasks: classification of white and red wines on the basis of physical and chemical properties. Assessment of the accuracy of the trained model.	4
Lab 5	Regression methods - introduction and exercises. Tasks: regularization.	4
Lab 6	Structural biology: Grouping of biomolecular structures using the DBSCAN algorithm. Sequence based peptide secondary structure prediction.	4
Lab 7	Training models for molecular simulations based on DFT calculations - models based on kernel ridge regression (AQML) and neural networks (ANI).	4
Lab 8	Work on individual projects. Presentation of reports on the implementation of individual projects.	4
	Total hours	30

TEACHING TOOLS USED

- N1. Presentation.
- N2. Problem solving in a small-group setting.

 N3. Implementation of solutions to problems and realization of tasks in a computer laboratory.

 EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	-	Way of evaluating learning outcomes achievement Grading mid-term reports (max 50 points)
1 1	PEU_K01-03	Grading find term reports (max 50 points)
P1	PEU_U01-PEU_U06	Grading the final report and project (max 50 points)
P2	PEU_W01-PEU_W06	Exam grade (max 100 points)
P (lab classes) 2.0 if (F1+P1) < 50 points 3.0 if (F1+P1) = 50 - 59 p 3.5 if (F1+P1) = 60 - 69 p 4.0 if (F1+P1) = 70 - 79 p 4.5 if (F1+P1) = 80 - 89 p 5.0 if (F1+P1) = 90 - 97 p 5.5 if (F1+P1) = 98 - 100	oints oints oints oints oints oints	
P (lecture) 2.0 if (P2) < 50 points 3.0 if (P2) = 50 - 59 points 3.5 if (P2) = 60 - 69 points 4.0 if (P2) = 70 - 79 points 4.5 if (P2) = 80 - 89 points	s s	

5.0 if (P2) = 90 - 97 points 5.5 if (P2) = 98 - 100 points

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] A. Géron, Hands-on Machine Learning with Scikit-Learn, Keras, and TensorFlow: Concepts, Tools, and Techniques to Build Intelligent Systems, O'Reilly Media, Sebastopol, CA, 2020.
- [2] B. Ramsunda, P. Eastman, P. Walters, V. Pande, Deep Learning for the Life Sciences, O'Reilly Media, Sebastopol, CA, 2019.

SECONDARY LITERATURE:

- [1] Lafuente D. et al., A Gentle Introduction to Machine Learning for Chemists: An Undergraduate Workshop Using Python Notebooks for Visualization, Data Processing, Analysis, and Modeling, J. Chem. Educ. 2021, 98, 2892-2898
- [2] Keith J.A. et al., Combining Machine Learning and Computational Chemistry for Predictive Insights Into Chemical Systems, Chem. Rev. 2021, 121, 9816-9872.
- [3] Artrith N. et al., Best practices in machine learning for chemistry, Nat. Chem. 2021, 13, 505-508.

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

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Attachment no. 4. to the Program of Studies

FACULTY of CHEMISTRY

SUBJECT CARD

Name of subject in Polish Naturalne produkty medyczne Name of subject in English Medicinal natural products

Main field of study (if applicable): Biosciences Specialization (if applicable): Medicinal chemistry

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code: W03BSS-SM2022W, W03BSS-SM2022L

Group of courses: NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	15		30		
Number of hours of total student workload (CNPS)	50		50		
Form of crediting (Examination / crediting with grade)	examination		passing with a grade		
For group of courses mark (X) final course					
Number of ECTS points	2		2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			1,4		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Knowledge in the field of general and organic chemistry.
- 2. Skills in basic laboratory techniques used in organic and analytical chemistry.

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SUBJECT OBJECTIVES

- C1 Acquiring knowledge about important groups of active compounds present in plant material their structures, properties, isolation and identification methods, mechanism of action, activity, and sources of occurrence.
- C2 Familiarizing students with methods of isolation and identification of biologically active compound products.
- C3 Developing skills for selecting isolation methods for specific plant raw materials.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

- PEU_W01 Familiarity with basic concepts in the field of phytochemistry and pharmacognosy.
- PEU_W02: Understanding the chemical groups determining the therapeutic properties of plant substances and products.
- PEU_W03: Knowledge of the basic chemical structures of coumarins, flavonoids, terpenoids, and alkaloids, including their actions and applications.
- PEU_W04: Understanding the main biogenetic pathways and building blocks of plant secondary metabolites.
- PEU_W05: Familiarity with methods for isolating biologically active compounds from plant material.

relating to skills:

- PEU_U01: Safely handling tasks in organic chemistry laboratories.
- PEU_U02: Properly conducting planned chemical experiments.
- PEU_U03: Isolating biologically active compounds from natural materials (e.g., plants).
- PEU_U04: Applying distillation and extraction techniques in the isolation of natural products.
- PEU_U05: Using chromatographic methods for the purification and identification of isolated compounds.
- PEU_U06: Writing a detailed report on conducted experiments, analyzing results, and drawing correct conclusions.

relating to social competences:

- PEU_K01: Ability to collaborate effectively in a group during laboratory sessions.
- PEU_K02: Willingness to organize one's work efficiently, critically assess acquired knowledge, and evaluate the progress of assigned tasks.

	PROGRAMME CONTENT	
	Lecture	Number of hours
Lec 1	Development of phytochemistry and natural product chemistry. Basic concepts, the role of compounds of natural origin in modern medicine and pharmacy. Secondary metabolites. Contemporary principles of classifying plant compounds. Preliminary information, assessment criteria.	
Lec 2	Biogenesis and Building Blocks. Main biogenetic pathways and building blocks of plant secondary metabolites. The information will encompass elements of their biogenesis.	
Lec 3	Coumarins. Characteristics of coumarins as a group of compounds with diverse pharmacological actions (e.g., anticoagulant, photosensitizing effects). Properties, structure, and mechanisms of action. Plant sources. Coumarin preparations available on the Polish market.	
Lec 4	Flavonoids and Stilbenes. Occurrence and characteristics of polyphenolic compounds, exemplified by flavonoids and stilbenes, natural antioxidants with diverse pharmacological significance (e.g., anti-inflammatory, vasodilatory, antimicrobial, vascular-sealing effects). Structure, classification, properties, and application of polyphenolic compounds (including glycosides) in medicinal products; plant sources rich in polyphenolic compounds. Polyphenolic preparations available on the Polish	

	market.	
	Terpenoids. Characteristics, structure, and properties of terpenoid compounds present in essential oils, used as medicinal products and dietary supplements (e.g., in digestive system disorders). Plant sources rich in essential oils. Preparations available on the Polish market.	
	Alkaloids and Their Glycosides. Structure, definition, properties, and classification of alkaloids, pharmacological properties of selected alkaloids and protoalkaloids, plant sources. Selected alkaloid preparations available on the Polish market.	
	Total hours	15
	Classes	Number of hours
Cl 1		
Cl 2		
Cl 3		
Cl 4		
	Total hours	
	Laboratory	Number of hours
Lab 1	Organizational activities, discussion of methods for isolating active substances from plant material, and occupational health and safety training.	2
Lab 2	Plant fats – isolation of trimyristin from nutmeg. Determination of the saponification value. Hydrolysis of trimyristin to myristic acid. Determination of the acid value.	
Lab 3	Terpenes – isolation of eugenol from clove oil.	4
5	The role of lycopene and β -carotene in the body – isolation of lycopene and β -carotene from tomatoes and carrots. Application of column chromatography for product separation.	
Lab 6	Steroids - isolation of cholesterol from egg yolk.	4
	Triterpene alcohols – isolation of betulin from birch bark. Continuous extraction. Passing grade colloquium.	8
	Total hours	30
	Project Project	Number
	·	of hours
Proj 1		
Proj 2		
Proj 3		
Proj 4		
•••	Total haves	
	Total hours	N T 3
	Seminar	Number of hours

Semin		
1		
Semin		
2		
Semin		
3		
		
	Total hours	
	TEACHING TOOLS USED	
N1. Le	cture with audiovisual aids.	
N2. Lal	boratory classes – conducting experiments.	

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F	Learning outcomes	Way of evaluating learning outcomes achievement
– forming	code	
during		
semester), P –		
concluding (at		
semester end)		
P (lecture)	PEU_W01-	examination
	PEU_W05	
F1	PEU_U01 -	passing with a grade
(laboratory)	PEU_U06, PEU_K01-	
_	K02	
F2	PEU_U01 -	Assessment of the correctness of experiment execution
(laboratory)	PEU_U06, PEU_K01-	and preparation of a report after completing laboratory
	K02	classes
	P (laboratory)) = F1 + F2; F1 - 60%; F2 - 40%

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

[1] J. Sołoducho, J. Cabaj, *Medicinal natural products*, http://zasobynauki.pl/

SECONDARY LITERATURE:

[1] P.M. Dewick, Medicinal natural products, Wiley 2009

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Joanna Cabaj, joanna.cabaj@pwr.edu.pl

Attachment no. 4. to the Program of Studies

FACULTY of CHEMISTRY

SUBJECT CARD

Name of subject in Polish METABOLOMIKA
Name of subject in English METABOLOMICS
Main field of study (if applicable): BIOSCIENCES

Specialization (if applicable): MEDICINAL CHEMISTRY

Profile: academic / practical*

Level and form of studies: 1st/2nd level, uniform magister studies*, full-time / part-time*

Kind of subject: obligatory / optional / university-wide* Subject code W03BSS-SM2021W, W03BSS-SM2021L

Group of courses NO

_					
	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	15		30		
Number of hours of total student workload (CNPS)	50		50		
Form of crediting (Examination / crediting with grade)	Crediting with grade		Crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points	2		2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)	, i		1,4		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Basic knowledge of biochemistry.
- 2. The ability to search for scientific information in magazines.
- 3. The ability to work in a group.
- 4. The ability to use remote work tools.
- 5. Knowledge of English.

SUBJECT OBJECTIVES

- C1. To familiarize students with metabolomics and the practical possibilities of its use in medicine and biotechnology.
- C2. To familiarize students with the use of modern chemical diagnostic methods in medicine and analytical methods of NMR spectroscopy and mass spectrometry.
- C3. To familiarize students with methods of preparing biological samples for analysis; safety rules.

- C4. Familiarizing students with scientific literature and the ability to interpret results and develop research protocols.
- C5. To familiarize students with the elements of chemometrics and statistics.
- C6. To familiarize students with metabolomics databases.
- C7. To familiarize students with ethical problems in science metabolomics.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

- PEU_W01 knows what metabolomics is and knows the scope of its applicability.
- PEU_W02 is able to interpret data on metabolites based on metabolomic pathways.
- PEU_W03 knows what chemometrics is and knows the basic methods of data analysis.
- PEU W04 knows how to use databases.
- PEU_W05 knows what NMR spectroscopy and MS spectrometry are and knows how they can be used in metabolomics research.
- PEU_W06 knows the procedures for preparing biological material for a specific measurement method.

relating to skills:

- PEU_U01 can read chemometric and statistical data.
- PEU_U02 is able to assign the appropriate sample preparation procedure to the appropriate measurement method.
- PEU_U03 is able to construct complex questions in factual databases and search for and analyze professional literature.
- PEU_U04 can look for relationships between biochemical pathways based on metabolomics data.
- PEU_U05 knows bioinformatics tools intended for the analysis of metabolomics data.
- PEU_U06 is able to work in the laboratory with biological material.
- PEU_U07 is able to use appropriate laboratory techniques for use in metabolomics.
- PEU_K01 jest gotów do krytycznej oceny posiadanej wiedzy

	PROGRAMME CONTENT	
	Lecture	Number of hours
Lec 1	General presentation of the subject's characteristics - assumptions, goals, possibilities General overview of the subject, definitions of assumptions and goals of metabolomics	1
Lec 2	Methods of sample preparation for metabolomics analysis. Discussion of the preparation of various types of samples for analysis. Discussion of the preparation of samples of biofluids, muscle tissue, feces, filamentous fungi and bacteria.	2

Lec 3	Application of MS mass spectrometry in metabolomics. Discussion of the basics and principles of operation of a mass spectrometer coupled with liquid chromatography.	2
Lec 4	Application of nuclear magnetic resonance (NMR) spectrometry in metabolomics Discussion of the basics and principles of operation of nuclear magnetic resonance spectroscopy.	2
Lec 5	Application of chemometric and statistical methods in metabolomics. Introduction to statistical and chemometric methods used in metabolomics, familiarization with the interpretation of results.	2
Lec 6	Bioinformatics tools Metabolomics analysis programs will be discussed, e.g. the MetPa program, along with the determination of disturbed metabolic pathways.	2
Lec 7	Application of metabolomics methods in medical diagnostics Discussion of the use of metabolomics methods in metabolomic, medical and biotechnological discrimination.	4
	Total hours	15

	Laboratory	Number of hours
Lab 1	Presentation of the general characteristics of the subject - literature review Discussion of basic concepts and definitions. Scope of applicability of metabolomics research. Methods used	1
Lab 2	Application of NMR spectroscopy in metabolomics – literature review Discussion of the principles of NMR spectroscopy, processing and interpretation of spectra, search for biomarkers	5
Lab 3	Application of MS mass spectrometry in metabolomics – a literature review Discussion of the principles of MS mass spectrometry, processing and interpretation of spectra, search for biomarkers	5
Lab 4	Application of statistical and chemometric methods in metabolomics – literature review Discussion of statistical and chemometric methods (PCA, PLS-DA, OPLS-DA) used in metabolomics, interpretation of the obtained data, search for a panel of biomarkers.	4
Lab 5	Discussion of the operation of the instruments, preparation of NMR and MS spectra Presentation of the NMR and MS instrument with a discussion of the measurements. Demonstration of important individual measurement steps.	2
Lab 6	Preparation of biofluid samples for analysis (e.g. blood and milk - commercial material of animal origin) with and without extraction of metabolites and NMR spectra Preparation of biofluids along with individual stages of metabolite extraction. Influence of sample preparation/extraction conditions on the results obtained. Sample preparation - with and without metabolite	3

	extraction. Differences in the sample preparation process	
Lab 7	Preparation of muscle and liver tissue for analysis (model purchased material - pork) along with preparation of NMR and MS spectra. Preparation of muscle and liver tissue along with individual stages of metabolite extraction. Influence of sample preparation/extraction conditions on the results obtained.	3
Lab 8	Analysis of the obtained spectra for the determination of selected metabolites. Presentation of spectra with discussion of metabolites and their interpretation. The use of computer programs for visualization of NMR and MS spectra along with their discussion	4
Lab 9	Application of statistical, chemometric and bioinformatic tools to analyze results, discriminant analysis Application of computer programs for statistical, chemometric and bioinformatic analysis of the obtained results - comparative and discriminatory studies.	3
	Total hours	30

TEACHING TOOLS USED

- N1. Multimedia presentations at lecture.
- N2. Film screenings.
- N3. Instruments of the metablomic laboratory (homogenizer, centrifuge, etc.)
- N4. Computer software

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	Learning outcomes code	Way of evaluating learning outcomes achievement
Lecture		
P	PEU_W01-PEU_W06, PEU_K01	kolokwium
Laboratory		
F1	PEU_U01- PEU_U07	Report on laboratory classes
F2		Activiti during classes
P		P = 70%F1 + 30%F2

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] Spectroscopic methods and their application to the identification of organic compounds, edited by Wojciech Zieliński and Andrzej Rajca; [author] Roman Mazurkiewicz [et al.]
- [2] Statistics and chemometrics in analytical chemistry, James Miller, Miller Jane
- [3] Materials from the lecture
- [4] scientific journals containing information related to the subject
- [5] knowledge found on websites.

SECONDARY LITERATURE:

- [1] William J Griffiths, NMR spectroscopy, Basic principles, concepts, and applications in chemistry, Second Edition, H Guenter, JOOHN WILEY & SONS
- [2] Metabolomics, Methods and Protocols, Wolfram Weckwerth, HUMANA PRESS;
- [3] Metabolomics, Metabonomics and Metabolite Profiling, William J. Griffiths, RSC Publishing
- [4] Mass Spectrometry, Juergen H Gross, Springer

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

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Attachment no. 4. to the Program of Studies

FACULTY of CHEMISTRY

SUBJECT CARD

Name of subject in Polish Nowoczesne Leki i Biofarmaceutyki

Name of subject in English Modern Pharmaceuticals and Biopharmaceuticals

Main field of study (if applicable): Biosciences

Specialization (if applicable): Medicinal Chemistry

Profile: academic / practical*

Level and form of studies: 1st/2nd level, uniform magister studies*, full-time / part-time*

Kind of subject: obligatory / optional / university-wide* Subject code W03BSS-SM2023W, W03BSS-SM2023L

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30		30		
Number of hours of total student workload (CNPS)	50		50		
Form of crediting (Examination / crediting with grade)	Exam		crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points	2		2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)	1,3		1,4		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Principles of organic chemistry, theoretical and practical.
- 2. Basic knowledge on biochemistry.
- 3. Knowledge in the field of basis of analytical chemistry is recommended.

SUBJECT OBJECTIVES

- C1 Acquaintance with the knowledge on the distribution of medicinal products and medical devices on basic groups, according to their mechanism of action on the human body.
- C2 Acquaintance with issues of the elementary production processes units in the area of pharmaceutical technology and biopharmacy.
- C3 Acquaintance with the generally applicable operating in the pharmaceutical industry and related sectors quality standards, concerning the manufacturing process and the final product, including the ways of managing waste and REACH requirements.

SUBJECT EDUCATIONAL EFFECTS

Relating to knowledge:

- PEU_W01 has knowledge on the distribution of medicines and medical products on the basic groups,
- PEU_W02 has knowledge on the methods of obtaining biologically active substances and the elementary production processes units in the area of pharmaceutical technology and biopharmacy,
- PEU_W03 can define the various forms of medicines and medical devices, and has knowledge on the technology of receiving them,
- PEU_W04 has knowledge on the generally applicable operating in the pharmaceutical industry and related sectors quality standards, concerning the manufacturing process and the final product, taking into account REACH directive.

Relating to skills:

- PEU_U01 has skills in the qualitative and quantitative analysis of a pharmaceutical formulation, due to the principles of proper samples preparation, precision and repetition in measurements and proper interpretation of the results,
- PEU_U02 has the ability to prepare simple biopharmaceutical preparation,
- PEU_U03 has skills in working in accordance with the principles of good laboratory practice (GLP), in the interpretation of the results of analyzes, error assessment, and the preparation of a laboratory report.

Relating to social competences:

- PEU_K01 Student is able to interact in a group and to plan an experiment.
- PEU K02 Student is able to discuss the quality of an experimental result.
- PEU_K03 Student works consciously and effectively in a sub-group to searches information and can subject them to critical analysis.

PROGRAMME CONTENT Number Lecture of hours The modern pharmaceutical industry: key assets to scientific and medical 2 Lec1 Drug targets – the idea of "golden bullet" for proteins, carbohydrates, lipids, 2 Lec2 DNA, and RNA. From discovery to clinical trials – the phases of pharmaceutical 2 Lec3 development. Good Clinical Practice rules (GCP) established by WHO. Quality assurance of pharmaceuticals and biopharmaceuticals. 2 Lec 4 2 Lec 5 Ways of obtaining active pharmaceutical ingredients (API). Lec 6 Biotechnology-derived drug product development. 2 Lec 7 Biopharmaceuticals – historical perspectives and future directions. Biopharmaceuticals of animal and microbial origin. Lec 8 2 Physical and physicochemical bases of pharmaceutical formulation. Lec 9 Lec 10 Pharmaceutical preformulation: types of naturally occurred excipients. 2 Purity problem. 2 Lec 11 Pharmaceutical preformulation: synthetic and semisynthetic excipients.

Lec 12	Tablets and capsules design. Modern solid dosage systems.	2
	Controlled release of API from solid and semisolid formulations — bioavailability problem.	2
Lec 14	The role of micro- and nanotechnology in pharmaceutical industry. Pharmaceutically accepted micro- and nanosystems.	2
Lec 15	Modern control mechanisms of the pharmaceutical industry. The influence of worldwide trends on the drug regulatory system.	2
	Total hours	30
	Laboratory	Number of hours
Lab 1	Safety rules in the laboratory of organic chemistry, good laboratory practice (GLP) and the rules of the reports preparation. Introduction to the separation and identification techniques of API.	2
Lab 2	Identification and qualitative analysis of drotaverine hydrochloride in NO-SPA tablet according to Pharmacopoeia regulations.	4
Lab 3	Suspension form of a drug for children containing ibuprophen – isolation and purification techniques of API. Analysis of the main compound.	4
Lab 4	Three compounds drug: Etopiryna (ethenzamide + acetylsalicylic acid + caffeine) – strategies of APIs separation from a tablet form.	4
Lab 5	Three compounds drug – analysis of the isolated APIs.	4
Lab 6	Polymeric nanocarriers for oral delivery of lipophilic vitamins – synthesis and characterization.	4
Lab 7	Kinetics of the release of clotrimazole from the ointment for epidermal application.	4
Lab 8	Electrophoresis as a tool for qualitative and quantitative analysis of high- protein dietary supplement.	4
	Total hours	30
	TEACHING TOOLS USED	

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- N1 Multimedial presentations.
- N2 Performing experiments with different laboratory equipment and instruments.
- N3 Preparation of report including analysis and interpretation of obtained results.

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	0	Way of evaluating learning outcomes achievement		
F1	PEU_W01-W04, PEU_U01 – PEU_U03	Exam - the grade for the final test of the lectures part		
F2	PEU_U01 – PEU_U04 PEU_K01 – PEU_K03	grades of the laboratory experiments (reports)		
P = the grade for the final test of the lectures part + average grade of the laboratory reports				

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] House of Commons Health Committee. The Influence of the Pharmaceutical Industry. HC 42-I [Incorporating HC 1030-i-iii], Published by authority of the House of Commons London: The Stationery Office Limited. 2005.
- [2] The European Federation of Pharmaceutical Industries and Associations. The Pharmaceutical Industry in Figures. 2022.
- [3] Quality assurance of pharmaceuticals: a compendium of guidelines and related materials. Vol. 2, Good manufacturing practices and inspection. 2nd ed. WHO Press. 2007.
- [4] Shayne Cox Gad, Pharmaceutical Manufacturing Handbook. Production and Processes. John Wiley & Sons, Inc. 2008.
- [5] Alfred Fahr, Voigt's Pharmaceutical Technology. John Willey & Sons Inc., 2018.
- [6] Introduction to Biopharmaceuticals. Mongomery County Community College, 2016.

SECONDARY LITERATURE:

- [7] EudraLex, The Rules Governing Medicinal Products in the European Union, Volume 4, EU Guidelines for Good Manufacturing Practice for Medicinal Products for Human and Veterinary Use, European Commission, health and consumers directorate-general, Ref. Ares(2012)778531 28/06/2012
- [8] Mark Gibson. Pharmaceutical Preformulation and Formulation Second Edition. A Practical Guide from Candidate Drug Selection to Commercial Dosage Form. Informa Healthcare USA, Inc. 2009.

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

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Attachment no. 4. to the Program of Studies

FACULTY of Chemistry

SUBJECT CARD

Name of subject in Polish: Dynamika Molekularna Name of subject in English: Molecular Dynamics

Main field of study (if applicable): Biosciences.....

Specialization (if applicable): Bioinformatics......

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03BSS-SM2002W, W03BSS-SM2002L

Group of courses NO

	Lectur e	Classe s	Laborator y	Projec t	Semina r
Number of hours of organized classes in University (ZZU)	30		30		
Number of hours of total student workload (CNPS)	100		50		
Form of crediting (Examination / crediting with grade)	Credit ing with grade		Creditin g with grade		
For group of courses mark (X) final course					
Number of ECTS points	4		2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)	1,3		1,4		

*delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. General Chemistry, Physics I and I
- 2. Algebra, Mathematical Analysis
- 3. Physical Chemistry

SUBJECT OBJECTIVES

- C1. Basic knowledge of statistical thermodynamics
- C2. Design of force fields and basics of molecular dynamics (MD)
- C3. Algorithms used in molecular dynamics
- C4. Preparation and running of molecular dynamics simulations

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

- PEU_W01 Basic concepts and laws of statistical thermodynamics
- PEU_W02 Potential energy form for a force field and understanding the physical meaning of each term
- PEU_W03 Methods to search for a global minimum in biological systems
- PEU_W04 How to choose suitable algorithms for molecular dynamics simulations
- PEU_W05 Algorithms to control temperature and pressure
- PEU_W06 Algorithms to calculate free energy within molecular dynamics framework
- PEU_W07 Analysis of MD results

Relating to skills:

- PEU_U01 Practical knowledge of Linux operating sytem
- PEU_U02 Practical knowledge of specific software to visualize and manipulate biomolecules
- PEU_U03 Practical knowledge of preparing input files and run and analyze simple minimization and MD simulations
- PEU_U04 Practical knowledge on how to prepare and run basic MD simulations for proteins

Relating to social competences:

PEU_K01 student is ready to critically evaluate his/her knowledge and received content

PROGRAMME CONTENT

	Lectures	Nu
Lec 1	Basic concepts. Molecular mechanics vs. quantum mechanics. Limitations of molecular mechanics. How good can MD be? - comparison with experimental results. The choice of a time step in MD simulations to describe various phenomena.	2
Lec 2	Introduction to statistical thermodynamics. Permutations and configurations. Probability theory in chemistry. Stirling approximation. Maxwell distribution. Partition function. Significance of Boltzmann distribution in chemistry. Statistical ensembles. Canonical ensemble. Canonical partition function: translational, rotational, vibrational and electronic terms.	2
Lec 3	Introduction to statistical thermodynamics – part 2. Internal energy and partition function: translational, rotational, vibrational and electronic contributions. Heat capacity and partition function. Entropy and partition function. Boltzmann equation and canonical partition function. Residual entropy. Free energy and equilibrium constant and partition function.	2
Lec 4	Quiz 1. Statistical thermodynamics	2
Lec 5	Force field – part 1. Definition of force field. Potential energy in force field. Bonding and non-bonding terms of potential. Harmonic and Morse potential. Mixed terms. Point charge model. RESP procedure. Buckingham and Lennarda-Jones potentials. Combination rules to create van der Waals parameters. Scaling of non-bonding potentials. Evaluation of cpu time in calculations of various potential energy terms.	2
Lec 6	Force field – part 2. All-atom and united-atom force fields. Transferability of force field parameters among different force fields. Accuracy of various force fields.	2
Lec 7	Preparation of input files for MD simulations. GROMACS options. How to choose an initial structure? A choice of a force field. Phases of MD procedure: minimization, heating, equilibration and production phase. Preparation of all required input files for MD simulations.	2

Lec 8	Methods of searching for global minimum in biomolecules. Methods for energy minimization. Levinthal paradox. Local and global minima in biosystems. Monte-Carlo method. Simulated annealing method. Genetic algorithm. Chain growth method. Homology modelling. Distance-geometry algoritm. Fragment-based algorithm.	2
Lec 9	MD algorithms – part 1. Determinism. Lyapunov instability. Newton's formalism. Lagrange's formalism. Hamilton's formalism. Integer algorithms: Euler, Verlet, velocity-Verlet, leap-frog, predictor-corrector. What are the features of a good algorithm? What are the criteria of choosing an optimal algorithm?	2
Lec 10	MD algorithms— part 2.Time step. Shake and rattle algorithms. Multiple time-step method. Liouville operator.	2
Lec 11	MD algorithms— part 3. Periodic boundary conditions. Minimum image convention. Cut-off technique. Switching i shifting functions. Neighbor list, cell list and Verlet list methods.	2
Lec 12	MD algorithms – part 4. Temperature and pressure in MD. Methods to control temperature in MD: stochastic, weak-coupling, strong-coupling, Nose-Hoover. Methods to control pressure in MD: volume scaling, Berendsen, Nose-Hoover and Andersen.	2
Lec 13	Free energy in MD. Algorithms to calculate free energy in MD: thermodynamic perturbation, thermodynamic integration and linear interaction energy. Free energy of solvation. Free energy binding of inhibitor to enzyme.	2
Lec 14	Analysis of MD results. Average quantities – temperature and pressure. Fluctuations: isobaric and isochoric heat capacity. Structural quantities: pair distribution function and static structure factor. Dynamic quantities: diffusion coefficient, velocity autocorrelation function, dynamic structure factor, MSD. Dipole autocorrelation function.	2
Lec 15	Quiz 2. MD algorithms	2
	Total hours	30

Computer laboratory	Nu

Lab 1	Requirements to pass a laboratory course.	2
Lab 2	Basic Linux commands	2
Lab 3	Basic commands ot 'vim' text editor.	2
Lab 4	Statistical thermodynamics - solving tasks.	2
Lab 5	Statistical thermodynamics - solving tasks.	2
Lab 6	VMD as a tool to analyze results of MD simulations.	2
Lab 7	VMD as a tool to analyze results of MD simulations.	2
Lab 8	Preparation of input files to simulate 216 water molecules using GROMACS. Calculations and analysis of results.	2
Lab 9	Preparation of input files to simulate 216 methanol molecules using GROMACS. Calculations and analysis of results.	2
Lab 10	Preparation of input files to simulate a ribonuclease S-peptide using MD.	2
Lab 11	Analysis of MD results for ribonuclease S-peptide in water.	2
Lab 12	Preparation of input files for minimization procedure of BPTI protein in water.	2
Lab 13	MD simulations of BPTI protein in water – heating, equilibration and production phases of MD.	2
Lab 14	Trajectory analysis of MD simulations of BPTI protein in water: RMSD, RMSF, kinetic energy, temperature, pressure, Ramachandran plot, hydrogen bonds and salt bridges, density of protein and water.	2
Lab 15	How does the change in time step, force field, deviation in Cartesian coordinates, the choice of an algorithm and van der Waals cut-off affect the physical properties of S-peptide? Analysis of the results.	2
	Total hours	30

	TEACHING TOOLS USED
N1	Lecture with multimedia presentation

N2	Solving pratice problem sets
N3	Usage of software
N4	Preparation of reports

EVALUATION OF SUBJECT EDUCATIONAL EFFECTS ACHIEVEMENT						
Evaluation F – forming (during semester), C – concluding (at semester end)	Educational effect number	Way of evaluating educational effect achievement				
F1	PEU_W01, PEU_K01	Quiz 1				
F2	PEU_W02 - PEU_W07, PEU_K01	Quiz 2				
P (laboratory)	PEU_U01 – PEU_U05	Report+obligatory presence at all laboratory classes				
P (lecture) = 3.0 if $(F1 + F2) = 50-60\%$ max. no of poins 3.5 if $(F1 + F2) = 61-70\%$ max. no of poins 4.0 if $(F1 + F2) = 71-80\%$ max. no of poins 4.5 if $(F1 + F2) = 81-90\%$ max. no of poins 5.0 if $(F1 + F2) = 91-99\%$ max. no of poins 5.5 if $(F1 + F2) = 100\%$ max. no of poins						

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

[1]. D. Frenkel, B. Smith "Understanding Molecular Simulation", Academic Press, 2001.

[2] J.M. Haile "Molecular Dynamics Simulation: Elementary Methods", Wiley-Interscience, 1997.

SECONDARY LITERATURE:

[1] M. P. Allen, D. J. Tildesley "Computer Simulation of Liquids", Oxford University Press, 1989.

SUBJECT SUPERVISOR

(NAME AND SURNAME, E-MAIL ADDRESS)

prof. dr hab. Tadeusz Andruniów, tadeusz.andruniów@pwr.edu.pl

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Inżynieria molekularna w analizach genomowych Name of subject in English Molecular engineering in genomic analyses

Main field of study (if applicable): Biosciences
Specialization (if applicable): Bioinformatics

Profile: academic / practical*

Level and form of studies: 1st/ 2nd level, uniform magister studies*, full-time / part-time*

Kind of subject: obligatory / optional / university-wide*

Subject code W03BSS-SM2015L Group of courses YES / NO*

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)			45		
Number of hours of total student workload (CNPS)			50		
Form of crediting (Examination / crediting with grade)			crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points			2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			2.1		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. The knowledge of basic molecular biology and genetic engineering.
- 2. The knowledge of the basic laboratory skills.
- 3. The ability of the basic laboratory calculations including calculations of mass and molar concentrations

SUBJECT OBJECTIVES

- Familiarization with analytical DNA techniques used in biotechnology, medicine, agriculture, archaeology and others.
- C2 Ability to isolate genetic material.
- C3 Familiarization with techniques used for detection of polymorphisms within genomic sequences.
- C4 Familiarization with techniques used for editing of nucleotide sequence.
- C5 Familiarization with techniques used for gene/genomes structure analysis.
- C6 Familiarization with analysis of genes expression and their function.

SUBJECT EDUCATIONAL EFFECTS

Relating to knowledge:

A student who has completed the course:

- PEU_W01 knows basic molecular tools and techniques used for obtainment and analysis of DNA
- PEU_W02 knows basic techniques of isolation, amplification and biochemical/biophysical description of DNA
- PEU_W03 knows techniques used for analysis of gene and genomes sequences
- PEU_W04 knows techniques used for analysis of gene expression and function
- PEU_W05 know the possible applications of genetic engineering in biotechnology, medicine, agriculture and others
- PEU_W06 know techniques of DNA sequence editing

Relating to skills:

- PEU_U01 can isolate genetic material from various sources
- PEU_U02 can plan restriction reaction and perform
- PEU_U03 can perform agarose gel electrophoresis and can interpret obtained results
- PEU_U04 can design primers and PCR program for enhancement of desired genome fragment
- PEU_U05 can use bioinformatics tools to compare genomic sequences

	PROGRAMME CONTENT					
	Laboratory					
Lab 1	Introduction, Health and Safety training, discussion on form of crediting of the course and the general introduction of the objective of this course.	6				
Lab 2	Isolation of the genetic material from the chick epithelium.	6				
Lab 3	Polymorphism of the gene coding for alcohol dehydrogenase ADH3	6				
Lab 4	Analysis of the insertion-deletion polymorphism of the gene coding for angiotensin convertase ACE.	6				
Lab 5	The use of a single-nucleotide polymorphism to predict bitter-tasting ability	6				
Lab 6	Analysis of the meat product authenticity.	6				
Lab 7	Detection of the transgenic soya beans in the food products /Analysis of the polymorphism of insertion of Alu element.	6				
Lab 8	Test	3				
	Total hours	45				
	TEACHING TOOLS USED					
N1.	Short introduction					
N2.	Multimedia presentation					
N3.	Realisation of the laboratory protocol					
N4.	Calculations, problem solving					
N5.	Preparation of the final assessment					
N6	Bioinformatics software					

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation	Learning	Way of evaluating learning outcomes achievement
(F – forming	outcomes code	
during		

semester), P – concluding		
(at semester		
end)		
		Written end-course examination and/or short question quiz at the beginning of the laboratory (according to teacher instructions presented during introduction laboratory)
F2	PEK_U01-	Written assessment from the performer work
(laboratory)	PEK_U05	
F3	PEK_U01-	Activity and involvement during classes
(laboratory)	PEK_U05	
P (laboratory	$(-1) = 0.8 \cdot F1 + 0.15$	6·F2 +0,05·F3
Attendance e	very class and s	submission of all the assessment is necessary to pass the course.
P (laboratory		$F_1+0,15\cdot F_2+0,05\cdot F_3) = 60,0-70,0 \text{ points}$
	, , ,	$F1+0,15\cdot F2+0,05\cdot F3) = 70,1-75,0$ points
	, , ,	$F1+0,15\cdot F2+0,05\cdot F3 = 75,1-80,0 \text{ points}$
	4,5 if (0,8·1	$F1+0,15\cdot F2+0,05\cdot F3 = 80,1-85,0$ points

PRIMARY AND SECONDARY LITERATURE

 $5.0 \text{ if } (0.8 \cdot \text{F1} + 0.15 \cdot \text{F2} + 0.05 \cdot \text{F3}) = 85.1 - 90.0 \text{ points}$

PRIMARY LITERATURE:

- [1] Brown, T.A. Gene Cloning and DNA Analysis: An Introduction. John Wiley & Sons, 7th edition
- [2] Experiment manuals available on the course-specific website only to qualified students

SECONDARY LITERATURE:

- [1] Voet, D., Voet, J.G. *Biochemistry* Wiley & Sons, Inc., 4th edition
- [2] Brown, T.A. Genomy PWN 2018
- [3] Węgleński, P. Genetyka molekularna PWN 2012

 $5.5 \text{ if } (0.8 \cdot \text{F1} + 0.15 \cdot \text{F2} + 0.05 \cdot \text{F3}) = 90.1 - 100.0 \text{ points}$

- [4] Berg, J.M., Tymoczko, J.L., Stryer, L. *Biochemia* PWN 2018
- [5] Berg, J.M., Tymoczko, J.L., Stryer, L. *Biochemistry* W.H. Freeman and Co., New York 9th edition
- [6] http://www.blackwellpublishing.com/genecloning/

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Prof. Andrzej Ożyhar, DSc, PhD, Eng andrzej.ozyhar@pwr.edu.pl

^{*}delete if not necessary

Attachment no. 4. to the Program of Studies

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Modelowanie molekularne

Name of subject in English Molecular modeling

Main field of study (if applicable): Biosciences

Specialization (if applicable):

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03BSS-SM2007W, W03BSS-SM2007L, W03BSS-SM2007S

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	15		30		15
Number of hours of total student workload (CNPS)	50		50		25
Form of crediting (Examination / crediting with grade)	Examination		crediting with grade		crediting with grade
For group of courses mark (X) final course					
Number of ECTS points	2		2		1
including number of ECTS points for practical classes (P)					
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)	1,3		1,4		0,7

*delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Basic knowledge of atomic and molecular structure concepts
- 2. Basic knowledge of analytic geometry
- 3. Basic knowledge of computer science
- 4. Basic knowledge of organic chemistry

SUBJECT OBJECTIVES

- C1 Teaching construction of 3-D molecular models
- C2 Teaching applications of quantum chemistry methods
- C3 Teaching elementary concepts of the theory of intermolecular interactions
- C4 Teaching modeling techniques of molecular aggregates
- C5 Teaching modeling chemical reactions

SUBJECT EDUCATIONAL EFFECTS

Relating to knowledge:

PEU_W01 – knowledge of construction of 3-dimensional molecular models and their transformations

PEU_W02 – knowledge of elementary molecular modeling methods and limits of their applications.

PEU_W03 – knowledge of major components of intermolecular interaction energy

PEU_W04 – knowledge of modeling drugs and biocatalysts

Relating to skills:

PEU_U01 – ability of construction of 3-D molecular model starting from assumed hybridization type

PEU_U02 – ability to predict molecular structure and properties

PEU_U03 – ability to predict possibile structures of molecular aggregates

PEU_U04 – ability to analyse protein-ligand interactions

PEU_U05 – ability to model dynamic properties of molecular aggregates

PROGRAMME CONTENT						
	Lecture	Number of hours				
Lec 1	Basic concepts. Interdisciplinary charakter of molecular modeling. Typical molecular modeling tasks. Molecular structure sources. Algorithms used in construction of 3-D molecular models with examples. Hybridization. Coordinate transformations. Basic concepts of molecular graphics. Visualization techniques. Literature review.	2				
Lec 2	Basic concepts of quantum chemistry. Review of quantum chemistry computational methods. Hueckel Molecular Orbitals and <i>ab intio</i> methods. Theoretical prediction of physical properties and structures.	2				
Lec 3	Construction of molecular models – exercises and test	2				
Lec 4	Basic concepts of the theory of intermolecular interactions. Perturbation theory. Characteristics of major components of intermolecular interaction components.	2				
Lec 5	Hydrogen bonding. Molecular charge distribution and electrostatic models. Force fields.	2				
Lec 6	Predicting properties and structure of molecular aggregates – exercises and test.	2				
Lec 7	Modeling interactions in receptors and enzyme active centers. Drug design techniques. Molecular dynamice. Homology modeling.	2				
Lec 8	Analysis of enzyme catalytic activity and biocatalyst design.	1				
	Total hours	15				
	Laboratory	Number of hours				
Lab 1	Introduction and lab organization. Editing of molecular structures.	2				
Lab 2	Force field parametrization of arbitral organic molecules: initial topology, atom types and non-bonding parameters	2				
Lab 3	Force field parametrization of arbitral organic molecules: optimization of atomic	2				
Euo 3	charges					
Lab 4	Force field parametrization of arbitral organic molecules: bonding parameters	2				
	Force field parametrization of arbitral organic molecules: bonding parameters Computational task #1.	2 2				
Lab 4 Lab 5 Lab 6	Force field parametrization of arbitral organic molecules: bonding parameters Computational task #1. Preparing molecular dynamics simulations	2 2				
Lab 4 Lab 5 Lab 6 Lab 7	Force field parametrization of arbitral organic molecules: bonding parameters Computational task #1. Preparing molecular dynamics simulations Preparing molecular dynamics simulations	2 2 2				
Lab 4 Lab 5	Force field parametrization of arbitral organic molecules: bonding parameters Computational task #1. Preparing molecular dynamics simulations	2 2				

Lab 1	1 Modeling energy profile of a reaction using QM/MM methods	2
Lab 1	2 Computational task #3.	2
Lab 1	Receptor-ligand docking and virtual screening	2
Lab 1	4 Quantum mechanical calculation of interaction energies	2
Lab 1	5 Computational task #4	2
	Total hours	30
	Seminar	Number of hours
Se1		1
Se2		2
Se3	7	2
Se4	7	2
Se5	Student's presentations of selected topics	2
Se6	7	2
Se7	7	2
Se8	7	2
	Total hours	15
	TEACHING TOOLS USED	
N1	Lecture with multimedia presentation	
N2	Solving problems	
N3	Use of software	
N4	Student multimedia presentation	
N5	Preparing report	

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming	Learning outcomes	Way of evaluat	ing learning outcomes achievement
during semester), P –	code		
concluding (at semester			
end)			
F_Lec1	PEU_W01, PEU_W02,	Test with proble	em solving
	PEU_U01		
F_Lec2	PEU_W02, PEU_W03,	Test with proble	em solving
	PEU_W04, PEU_U01,	•	· ·
	PEU_U03		
F_Lab1	PEU_W04, PEU_U05	Computational	task #1
F_Lab2	PEU_W01, PEU_W04,	Computational	task #2
	PEU_U01, PEU_U04		
F_Lab3	PEU_W04, PEU_U03,	Computational	task #3
	PEU_U04		
F_Lab4	PEU_W04, PEU_U02	Computational	task #4
$P_{\text{lecture}} = F_{\text{Lec}}1 + F_{\text{Lec}}$	2 or final exam	Score	Grade
$P_lab = F_Lab1 + F_Lab2 + F_$	Lab3+F Lab4	50-59,99%	3,0
		60-69,99%	3,5
		70-79,99%	4,0
		80-89,99%	4,5
		90-100%	5,0
P_seminar		Preparation and	l presentation of seminar on
		individual topic	···

Active participation in discussion of presentations of other students

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] L. Piela, Quantum Chemistry Ideas, Elsevier, 2010
- [2] A.R. Leach, Molecular Modeling: Principles and Applications, (2-nd Ed), Prentice Hall, 2001
- [3] H.D. Hotje, Molecular modeling. Basic principles and applications, (3-rd Ed), Wiley, 2008
- [4] T. Schlick, Molecular modeling and simulation, Springer, 2002.

SECONDARY LITERATURE:

- [1] F. Jensen, Introduction to computational chemistry, Wiley, 2006 (2-nd Ed)
- [2] J.M. Goodman, Chemical Applications of Molecular Modeling, RSC, 1999.
- [3] J.P. Doucet, J. Weber, Computer-Aided Molecular Design, 1996, Academic Press, 1996
- [4] G.H. Grant, W.G. Richards, Computational chemistry, Oxford Sci. Publ., 1995

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Paweł Kędzierski, <u>Pawel.Kedzierski@pwr.edu.pl</u>

FACULTY of CHEMISTRY

SUBJECT CARD

Name of subject in Polish ...Wieloetapowa synteza organiczna
Name of subject in EnglishMultistep organic synthesis....

Main field of study (if applicable): ...BIOSCIENCES....
Specialization (if applicable): ...Medicinal chemistry

Profile: academic

Level and form of studies: 2nd level,

Kind of subject: obligatory

Subject code ... W03BSS-SM2024L Group of courses YES / NO*

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)			60		
Number of hours of total student workload (CNPS)			75		
Form of crediting (Examination / crediting with grade)			crediting with grade)		
For group of courses mark (X) final course					
Number of ECTS points			3		
including number of ECTS points for practical classes (P)			3		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			2,8		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Knowledge and skills at the level of completing the "Fundamentals of organic chemistry laboratory" course or equivalent
- 2. Basic knowledge of English at a communicative level

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SUBJECT OBJECTIVES

- C1 Acquires students' proficiency in laboratory work using advanced experimental techniques of organic synthesis.
- C2 Ability to practically use various transformation methods in multi-stage synthesis creating new C-C bonds, transformations on functional groups
- C3 Ability to perform a complex synthetic sequence based on literature data.

SUBJECT EDUCATIONAL EFFECTS

relating to skills:

- PEU_U01 is able to carry out a multi-stage synthesis of an organic compound,
- PEU_U02 knows how to use scientific literature and chemical databases
- PEU_U03 is able to select the conditions for various transformations and plan methods of isolating and purifying products,
- PEU_U04 is able to independently interpret the results, measure basic physicochemical constants, interpret spectroscopic spectra of organic compounds relating to social competences:
- PEU_K01— knows English at a communicative level, is able to keep a laboratory journal in English

	PROGRAMME CONTENT	
	Lecture	Number of hours
Lec 1		
Lec 2		
Lec 3		
Lec 4		
Lec 5		
	Total hours	
	Classes	Number of hours
Cl 1		
Cl 2		
Cl 3		
Cl 4		
	Total hours	
	Laboratory	Number of hours
Lab 1	Information on how to conduct and pass exercises and keep a laboratory journal. Basic equipment (glass and metal) and laboratory operations. Work safety in the laboratory: harmful, flammable substances, etc. Synthesis planning - using literature and databases.	4
Lab 2	Carrying out one-step syntheses requiring selective reduction of the C=O	4
Lab 3	and C=C bonds - procedures	4
Lab 4	to be selected by the lecturer (from a prepared script)	4
Lab 5		4
Lab 6	Carrying out a one-step synthesis requiring selective oxidation - procedure	4
Lab 7	to be selected by the instructor (from a prepared script)	4
Lab 8	Conducting a 3- and 4-step synthesis of a compound with known biological	4

	transformations on various functional groups. Purification, identification and characterization of products - measurement of physico-chemical				
Lab 10					
Lab 11					
Lab 12	Interpretation of the results.				
Lab 13					
Lab 14		4			
Lab15	Settlement of laboratory equipment and laboratory notes.	4			
	Total hours				
	Project	Number of hours			
Proj 1					
Proj 2					
Proj 3					
Proj 4					
•••					
	Total hours				
Seminar					
Semin 1					
Semin 2					
Semin 3					
	Total hours				
	TEACHING TOOLS USED				
	scussion of the experiment: planning the equipment, techniques used and sub	sequent			

- N2. Carrying out experiments independently
 N3. Preparing a report in a laboratory journal (in English)

 EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F	Learning	Way of evaluating learning outcomes achievement					
forming	outcomes code						
during							
semester), P –							
concluding (at							
semester end)							
P	PEU_U01-	Independent synthesis of given products, measurement of					
	PEU_U04	physical and chemical constants for product characterization,					
	PEU_K01	preparing a report in the laboratory book in English.					
PRIMARY AND SECONDARY LITERATURE							

PRIMARY LITERATURE:

- [1] R. Siedlecka, Multistep organic synthesis. Laboratory course for students of medicinal chemistry, Wrocław, 2020;
- [2] A. Mucha, R. Siedlecka, Multistep organic synthesis. Practical course, Wrocław, 2010;
- [3] A. I. Vogel, Preparatyka organiczna, WNT, Warszawa, 2006;
- [4] Bazy danych: Beilstein, Chemical Abstracts, Current Contents.

SECONDARY LITERATURE:

- [1] J. Gawroński, K. Gawrońska, K. Kacprzak, M. Kwit, Współczesna synteza organiczna, PWN, Warszawa, 2004
- [2] L.-T. Ho, Tactics of Organic Synthesis, J. Wiley, New York, 1994

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Dr hab. inż. Renata Siedlecka, renata.siedlecka@pwr.edu.pl

FACULTY Chemistry

SUBJECT CARD

Name of subject in Polish: Sieci i stacje robocze z systemem unix

Name of subject in English: Networks and workstations with unix system

Main field of study (if applicable): Biosciences Specialization (if applicable): Bioinformatics

Profile: academic / practical*

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory Subject code W03BSS-SM2003L

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)			30		
Number of hours of total student workload (CNPS)			50		
Form of crediting (Examination / crediting with grade)			crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points			2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			1,4		

*delete as applicable

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. English basic level
- 2. Basic computer skills

SUBJECT OBJECTIVES

- C1 Learning the mechanisms of unix system, and rules of computer network based on the internet protocol
- C2 Developing skills for using unix systems at unassisted administration level

SUBJECT EDUCATIONAL EFFECTS

relating to skills:

- PEU_U01 Student can run programs from the command line, to perform various file operations, and use a text editor
- PEU_U02 Student can use documentation for programs, available in unix system
- PEU_U03 Student can write an inittab file and simple scripts responsible for initial system configuration, check the consistency of a filesystem and attach it to the directory tree
- PEU_U04 Student can add and remove user accounts, change passwords and assign users to groups, can write session scripts (bash shell)
- PEU_U05 Student can assign the network address to network interface, build the routing table, create local list of address-name relationships and prepare the system for using the DNS service

PEU_U06 Student can use network services of remote terminal, copying files between systems and electronic mail, can make them available for remote users and limit this remote access to specific addresses.

PEU_U07 Student can run local and remote graphical applications in the X window system PROGRAMME CONTENT Laboratory Number of hours Lab 1 Programs and processes. Parent and child processes, system mechanisms for 2 running programs and process termination. Signals. User's and group's identifiers - introduction of mechanisms regulating access rights to various system resources. Lab 2 Files and file types: normal, directories, special (character and block devices), files representing communication channels (sockets and named pipes). Normal pipes and their similarity to files. The notion of a filesystem, hard and symbolic links. Review of programs for various file operations, including short introduction to the vi editor. Lab 3 Running the linux kernel under control of the QEMU emulator. Creation of a file representing hard disk, partitioning and creation of filesystem. Archives created with the tar program. Installation of minimal set of programs, needed for running the system. Lab 4 Duties of the program running with process identifier equal to 1. Configuration of the init program (implementation: sysvinit) - the inittab file. Review of tasks performed at the system's initialization stage. Lab 5 Checking of filesystems' consistency and attaching filesystems to the directory tree. Mount and umount programs, the /etc/fstab file. Shared libraries. Lab 6 User accounts - entries in the /etc/passwd file, relationship of names with user identifiers, home directories, encryption and storing of passwords. System and personal session scripts. Creation of groups (the /etc/group file). Programs: su and newgrp. IP address, address' class, structure of an address within given network Lab 7 segment (network mask). Assignement of IP address to the network interface, with the ifconfig program. The loopback interface. Creation of the routing table with the route program. Internet names, name-address relationship. Methods of translating names to Lab 8 addresses and addresses to names: local list in the /etc/hosts file and the DNS network service. Lab 9 TCP and UDP transport protocols. The notion of network socket. Assignement of network services to port numbers (/etc/services file). Rules of making services available by the inetd program. Lab 10 Limiting remote access to network services – mechanisms and configuration 2 of the TCP wrappers software (tcpd program and library code) by access control lists in /etc/hosts.allow and /etc/hosts.deny files. Lab 11 Working in a remote system - services of remote terminal (telnet and ssh) and file transfer (ftp, scp, sftp). Reasons for using encrypted communication

channels.

	Total hours	30
Lab 15	Crediting	2
	The X window system - graphical environment with client-server architecture.	2
Lab 13	The WWW server - basic configuration of the boa program, creation of simplest WWW pages in the HTML language. Text WWW browser - lynx.	2
	Electronic mail - MTA and MUA programs, running an MTA program (smail) and using the mutt mail client (MUA). Basic rules for securing the mail server (MTA).	2

TEACHING TOOLS USED

- N1. Demonstration
- N2. Practical exercises, under teacher's control
- N3. Practical exercises, with a simple problem to be solved single-handedly by the student

EVALUATION OF SUBJECT EDUCATIONAL EFFECTS ACHIEVEMENT

Evaluation (F – forming (during semester), P – concluding (at semester end)	Educational effect number	Way of evaluating educational effect achievement
F1	PEU_U01-U07	practical exercises (up to 25 points)
P1	PEU_U02-U06	written test (up to 75 points)
F2	PEU_U02-U06	outstanding knowledge or skills (up to 10 points)
C=F1+P1+F2		
$50 \le C \le 603.0$		
$60 \le C < 703.5$		
70 . 0 .00 40		

 $70 \le C \le 804.0$

80 <= C < 90 4.5 90 <= C < 100 5.0

 $C >= 100 \quad 5.5$

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

[1] Aeleen Frisch, UNIX: administracja systemu, O'Reilly & Associates, wydawnictwo RM, Warszawa 1997

SECONDARY LITERATURE:

[1] Craig Hunt, TCP/IP: administracja sieci. wydawnictwo RM, Warszawa 2003

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS) dr hab. inżź. Krzysztof Strasburger, e-mail: krzysztof.strasburger@pwr.edu.pl, strasbur@chkw386.ch.pwr.wroc.pl

FACULTY of Chemistry

SUBJECT CARD

Name of subject in Polish: Praca dyplomowa I
Name of subject in English: Graduate laboratory I

Main field of study (if applicable): Specialization (if applicable):

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03W03-SM1054D, W03W03-SM2054D

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)			60		
Number of hours of total student workload (CNPS)			150		
Form of crediting (Examination / crediting with grade)			crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points			6		
including number of ECTS points for practical classes (P)			6		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			3		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

1.

SUBJECT OBJECTIVES

- C1 Developing the ability to select and analyze sources of knowledge, including scientific literature
- C2 Developing the ability to create a written study on the topic of the diploma thesis
- C3 Expanding the skills of planning and conducting scientific work

SUBJECT EDUCATIONAL EFFECTS

In relation to knowledge:

PEU_W01 – knows the types of sources of scientific and professional knowledge,

PEU W02 – has in-depth knowledge of the topic of the diploma thesis

In relation to skills:

PEU_U01 – is able to collect information useful for learning about a specific issue and preparing for the completion of a diploma thesis

PEU_U02 – is able to critically analyze the collected information in a form written on a selected scientific or practical issue.

PEU_U03 – (optional) is able to plan and carry out experiments/design work as well as develop the results and draw conclusions from their achievements and plan further work. In relation to social competences:

PEU_K01 – is ready to critically evaluate knowledge obtained from various sources PEU_K02 – is ready to comply with the principles of professional ethics and respect copyrights

	PROGRAMME CONTENT					
	, and the state of	Number of hours				
	Lab 1- Individual student work on a selected topic according to the schedule agreed Lab15 with the diploma thesis supervisor					
	Total hours					
	TEACHING TOOLS USED					
N1. coi	nsultations					

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	<u>o</u>	Way of evaluating learning outcomes achievement
		assessment of student work based on progress in completing the diploma thesis

PRIMARY AND SECONDARY LITERATURE

Scientific and professional literature indicated by the course tutor and/or found by the student.

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Supervisors of individual diploma thesis topics

Subject card preparation:

Piotr Rutkowski, piotr.rutkowski@pwr.edu.pl

FACULTY of Chemistry

SUBJECT CARD

Name of subject in Polish: Praca dyplomowa II
Name of subject in English: Graduate laboratory II

Main field of study (if applicable): Specialization (if applicable):

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03W03-SM1055D, W03W03-SM2055D

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University			210		
(ZZU)					
Number of hours of total student workload (CNPS)			500		
Form of crediting (Examination / crediting with			crediting		
grade)			with grade		
For group of courses mark (X) final course					
Number of ECTS points			20		
including number of ECTS points for practical			20		
classes (P)					
including number of ECTS points corresponding to			9,5		
classes that require direct participation of lecturers					
and other academics (BU)					

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

1.

SUBJECT OBJECTIVES

C1 implementation of a research project

C2 written preparation of the diploma thesis

SUBJECT EDUCATIONAL EFFECTS

In relation to knowledge:

PEU_W01 – knows the types of sources of scientific and professional knowledge

PEU W02 – has advanced knowledge of the topic of the diploma thesis

In relation to skills:

PEU_U01— is able to carry out experiments / develop a project in accordance with the developed work plan

- PEU_U02 is able to compare information obtained from sources of knowledge with the results of research, verify the results of own research, draw conclusions and plan further work
- PEU_U03 is able to develop the results of his/her work on a selected topic and present them in the form of a diploma thesis

In relation to social competences:

PEU_K01 – is ready to critically evaluate the obtained results of research work on a selected topic

PEU_K02 – is ready to comply with the principles of professional ethics and respect copyrights

	PROGRAMME CONTENT					
	v	Number of hours				
	Individual student work on a selected topic according to the schedule agreed with the diploma thesis supervisor	210				
	Total hours	210				
	TEACHING TOOLS USED					
N1. cor	nsultations					

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F –	Learning	Way of evaluating learning outcomes achievement
forming during	outcomes code	
semester), P –		
concluding (at		
semester end)		
P	PEU_W01 -	assessment of student work based on progress in
	PEU_W02	completing the diploma thesis
	PEU_U01 -	
	PEU_U03	
	PEU_K01 –	
	PEU_K02	

PRIMARY AND SECONDARY LITERATURE

Scientific and professional literature indicated by the course tutor and/or found by the student.

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Supervisors of individual diploma thesis topics

Subject card preparation:

Piotr Rutkowski, piotr.rutkowski@pwr.edu.pl

FACULTY of Chemistry

SUBJECT CARD

Name of subject in Polish: Proseminarium

Name of subject in English: Graduation proseminar

Main field of study (if applicable): Specialization (if applicable):

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03W03-SM1053S, W03W03-SM2053S

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)					15
Number of hours of total student workload (CNPS)					25
Form of crediting (Examination / crediting with grade)					crediting with grade
For group of courses mark (X) final course					
Number of ECTS points					1
including number of ECTS points for practical classes (P)					1
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)					0,7

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

l.

2.

3.

SUBJECT OBJECTIVES

C1

C2

SUBJECT EDUCATIONAL EFFECTS

In relation to knowledge:

PEU_W01 – has knowledge of research topics related to the studied field of study conducted in organizational units of the Faculty of Chemistry

In relation to skills:

PEU_U01 – can take an active part in discussions on scientific topics

In relation to social competences:

PEU_K01 – is aware of the need to improve their competences in the field of research within the field of study

	PROGRAMME CONTENT				
	Seminar	Number of hours			
Semin 1- 15	Discussion of the topics of diploma theses by employees of the Faculty's units conducting research related to the field of study; Presentation of research and analytical laboratories in the Faculty's units; Discussion of the rules for selecting the topics of diploma theses and the rules for implementing/passing the "Diploma Thesis" courses	15			
	Total hours	15			

TEACHING TOOLS USED

- N1. Presentation
- N2. Discussion
- N3. Consultations

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F –	Learning outcomes	Way of evaluating learning outcomes achievement
forming during	code	
semester), P –		
concluding (at		
semester end)		
P	PEU_W01	Attendance at classes, participation in discussions -
	PEU_U01	assessed by the people conducting the classes
	PEU_K01	

PRIMARY AND SECONDARY LITERATURE

N/A

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Chairman of the study program committee

SUBJECT CARD

Name of subject in Polish: Przedmiot kierunkowy wybieralny

Name of subject in English: Elective course

Main field of study (if applicable):
Specialization (if applicable):

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: elective

Subject code

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30				
Number of hours of total student workload (CNPS)	25				
Form of crediting (Examination / crediting with grade)	Zaliczenie na ocenę				
For group of courses mark (X) final course					
Number of ECTS points	2				
including number of ECTS points for practical classes (P)					
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)					

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

1.

2

3.

SUBJECT OBJECTIVES

C1 Familiarizing the student with advanced issues in the field of chemical sciences (including biotechnology) and/or materials engineering and/or chemical engineering (including chemical technology)

SUBJECT EDUCATIONAL EFFECTS

In relation to knowledge:

PEU_W01 – Knows and is able to describe the basic phenomena and processes occurring in the life cycle of devices, objects and technical systems.

PEU_W02 – has in-depth knowledge of development trends and new achievements in the field of chemical engineering and technology and related sciences

In relation to social competences:

PEU_K01 – is ready to critically evaluate the knowledge acquired and the content received

PEU_K02 – is aware of the role of a technical university graduate and the need to maintain the ethos of the engineering profession

	PROGRAMME CONTENT				
Lecture	Number of hours				
To familiarize students with advanced concepts, theories describing phenomena, operations and processes occurring in living and inanimate systems, as well as with the latest trends in chemical sciences, chemical engineering and related sciences. Issues presented in an elective subject, depending on the field studied, may include, among others: - adsorbents in environmental protection and industry - alternative and renewable energy sources, renewable raw materials in industry, recycling technologies - technical security - medical and pharmaceutical chemistry - chemistry of coordination compounds - chemistry of fragrance compounds - physical chemistry of chemical processes and products - chemistry, engineering and technology of materials (polymer, carbon, ceramic, metallic) and composites - technologies of dispersed systems - catalysts and catalysis in industry - instrumental methods in chemistry - physicochemical description of simple and complex systems - from the borderline of biology and medicine, describing the biological and biochemical basis of the functioning of organisms, including chemical and biochemical processes at the cellular and molecular level - industrial aspects of biotechnology - recycling of precious metals - issues of technological process and quality management, principles of investing and operating chemical technologies - modern chemical technologies - modern chemical technologies - biotechnology development trends - basics of spectroscopic methods, - bioelectrochemical systems - issues related to sustainable development - characteristics of the biotechnology and chemical industry in Poland and in the world	30				
	50				

N1. Presentation

N2. Discussion

<i>U V V V V V V V V V V</i>	_	Way of evaluating learning outcomes achievement					
concluding (at semester end)							
Р	PEU_W01- PEU_W02 PEU_K01- PEU_K02	Writing test (to pass minimum 50% of points)					
PR	PRIMARY AND SECONDARY LITERATURE						
[1] Literature is provided during the first classes by the teachers of the elective subject							

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Chairman of study program committee

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Racjonalne projektowanie leków

Name of subject in English Rational drug design Main field of study (if applicable): Biosciences

Specialization (if applicable):

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03BSS-SM2006W

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30				
Number of hours of total student workload (CNPS)	75				
Form of crediting (Examination / crediting with grade)	crediting with grade				
For group of courses mark (X) final course					
Number of ECTS points	3				
including number of ECTS points for practical classes (P)					
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)	, , -				

*delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Knowledge of the basics of organic chemistry
- 2. Knowledge of the basics of biology

SUBJECT OBJECTIVES

- C1 To familiarize students with the basics of drug design.
- C2 Understanding the economic aspects of drug design.
- C3 Learning about targeted therapy methods.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

Person who passed the subject:

PEU_W01 – knows the basic principles of drug design,

PEU_W02 – is able to select the appropriate drug design technique depending on the level of knowledge about the physiological process,

PEU_W03 – has basic knowledge about the costs and time horizon of drug design,

PEU_W04 – understands the physiological and economic effects of using drugs.

relating to:

Person who passed the subject:

PEU_K01 – recognizes the importance of non-technical aspects of scientific activity

	PROGRAMME CONTENT				
	Lecture	Number of hours			
Lec 1	Economics of drug design and development. Cost and time required to introduce new drug to the market. Generic drugs. Globalization.	2			
Lec 2	Randomized screening. Historical perspective. Illustration of the opinion of Louyis Pasteur "Fortune favors prepared minds". Case studies.	2			
Lec 3	Natural products as a source of drugs. History of the discovery of aspirin, morphine, artemisinin, quinine, penicillin and taxol. Current trends in natural drug research.	2			
Lec 4	Choice of the target. HIV as an example for choice of the target for drug design.	2			
Lec 5	Theory of structural analogy. Historical perspective (sulfonamides). Direct similarity versus topological one with analogs of morphine and anti-influenza drugs as examples.	2			
Lec 6	Theory of structural analogy. Chemical outlook, trics and "magic methods". Peptidomimetics.	2			
Lec 7	Covalent drugs. Overview of functional groups able for irreversivble bonding with proteins. Techniques of design of covalent drugs. Case studies.	2			
Lec 8	Transition-state analogues. Techniques used for the identification of transition state. Pauling's theory of the course of enzymatic reaction. Construction of transition-state analogues. Computer-aided techniques.	2			
Lec 9	Topological conformity. Antagonists and agonists. Natural peptides as scaffolds.	2			
Lec 10	QSAR models. Analysis of inhibitory activity using Hansh and Wilson models.	2			
Lec 11	Three-dimensional structure of receptors as a basis for drug design. Construction of pharmacophore. Computer-aided methods for drug design — QSAR and molecular modeling. Receptor flexability.	2			
Lec 12	Selective complexation enzyme inhibitors. The analysis of forces governing the ligand-protein binding.	2			
Lec 13	Structure-based drug design. The use of protein crystal structure and molecular modelling tools for drug design.	2			
Lec 14	Drug targeting and delivery. Prodrugs. Engineered metabolic activation. Targeted enzyme prodrug therapy.	2			
Lec 15	Final Test	2			
	Total hours	30			

TEACHING TOOLS USED				
N1. lecture with multimedia presentation				
N2 own work				

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	•	Way of evaluating learning outcomes achievement
	PEU_W01 - PEU_W04 PEU_K01	Test

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] K. M. Merz, Drug Design, structure and Ligand-Based Approaches, Cambridge University Press, 2010
- [2] Medicinal Chemistry and Drug Design, Intech (open access), 2012

SECONDARY LITERATURE:

- [1] Design of Drugs: Basic Principles and applications, ed. J. H. Poupaert, Marcel Dekker, 2002
- [2] The Organic Chemistry of Drug Design and Drug Action, Academic Press, 2004
- [3] Virtual Screening. ed. M. O. Taha, Intech (open access), 2012
- [4] Drug Development A Case study Based Insight intor Modern Startegies, Intech (open access), 2011

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Prof. dr hab. Łukasz Berlicki, lukasz.berlicki@pwr.edu.pl

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Informacja naukowa i techniczna

Name of subject in English Retrieval of scientific and technical information

Main field of study (if applicable): Biosciences

Specialization (if applicable): Bioinformatics, Medicinal Chemistry

Profile: academic / practical*

Level and form of studies: 1st/ 2nd level, uniform magister studies*, full-time / part-time*

Kind of subject: obligatory / optional / university-wide*

Subject code W03BSS-SM2008L

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)			15		
Number of hours of total student workload (CNPS)			25		
Form of crediting (Examination / crediting with grade)			Crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points			1		
including number of ECTS points for practical classes (P)			1		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			0,7		

*delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

1. Basic knowledge of information technology

SUBJECT OBJECTIVES

- C1 Acquainting with the selected topics regarding the scientific literature
- C2 Acquainting with the literature databases
- C3 Acquainting with the factographic databases in the fields of chemistry and biotechnology
- C4 Acquainting with the research funding
- C5 Acquainting with the selected topics of ethics in science

SUBJECT EDUCATIONAL EFFECTS

relating to skills:

PEU U01 Student is able to develop the complex search queries for literature databases

PEU_U02 Student is able to develop the complex search queries for factographic databases

PEU_U03 Student is able to find job and internship calls

PEU U04 Student is able to find active grants regarding the selected topic

PEU_U05 Student is able to detect the plagiarism

relating to social competences:

PEU_K01 Student appreciates the necessity of the assessment of the quality and credibility of

the scientific information

PEU_K02 Student is able to follow the code of ethics in science and to respect the copyright policies

	PROGRAMME CONTENT	
	Lecture	Number of hours
Lec 1		
Lec 2		
Lec 3		
Lec 4		
Lec 5		
	Total hours	
	Classes	Number of hours
Cl 1		
C1 2		
Cl 3		
Cl 4		
••		
	Total hours	
	Laboratory	Number of hours
Lab 1	Structure and preparation of scientific articles	2
Lab 2	Current Contents literature database and building of search queries	2
Lab 3	Web of Science literature database and Journal Citation Reports	2
Lab 4	Preparation of grant proposals and searching for grants, internships and patents	2
Lab 5	Analysis of structural data from Cambridge Structural Database	2
Lab 6	Reaxys-Beilstein and Scifinder-Chemical Abstracts databases	2
Lab 7	Searching for job offers and preparation of academic resume	2
Lab 8	Code of ethics in science	1
	Total hours	15
	Project	Number of hours
Proj 1		
Proj 2		
Proj 3		
Proj 4		
•••	T-4-11	
	Total hours	Ni1 · · · · · · · ·
	Seminar	Number of

		hours
Semin 1		
Semin 2		
Semin 3		
	Total hours	
	TEACHING TOOLS USED	

- N1. Lecture with multimedia presentation
- N2. Problem solving
- N3. Problem solving with the computer software

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming during semester), P – concluding (at semester end)	Learning outcomes code	Way of evaluating learning outcomes achievement
F	PEU_U01-PEU_U05 PEU_K01, PEU_K02	Final report (max 100 points)
P = 3,0 (F=50-60 points) 3,5 (F=61-70 points) 4,0 (F=71-80 points) 4,5 (F=81-90 points) 5,0 (F=91-95 points) 5,5 (F=96-100 points)		

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] D. Ridley, Finding scientific information information retrieval, Wiley, 2002
- [2] D. Lindsay, Scientific writing = thinking in words, CSIRO Publishing, 2011
- [3] M. Carter, Designing Science Presentations. A Visual Guide to Figures, Papers, Slides, Posters, and More, Academic Press 2013

SECONDARY LITERATURE:

[1] On Being a Scientist: A Guide to Responsible Conduct in Research: Third Edition, 2009, The National Academies Press

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

Dr inż. Edyta Dyguda-Kazimierowicz, Edyta.Dyguda@pwr.edu.pl

FACULTY of Chemistry

SUBJECT CARD

Name of subject in Polish: Seminarium dyplomowe
Name of subject in English: Graduation seminar

Main field of study (if applicable): Specialization (if applicable):

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code: W03W03-SM1056S, W03W03-SM2056S

Group of courses: NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)					15
Number of hours of total student workload (CNPS)					50
Form of crediting (Examination / crediting with grade)					crediting with grade
For group of courses mark (X) final course					
Number of ECTS points					2
including number of ECTS points for practical classes (P)					2
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)					0,7

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

n/a

SUBJECT OBJECTIVES

C1 development of students' social competences in presenting the results of their diploma thesis, initiating discussions and actively participating in them

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

PEU_W01 – has in-depth knowledge of the topic of the diploma thesis

relating to skills:

PEU_U01 – is able to collect and verify information necessary to learn about the selected research topic

PEU_U02 – is able to draw conclusions from the results of one's own research in relation to literature sources

PEU_U03 – is able to publicly present the results of his research and defend them during public discussion

PEU_U04 – is able to transfer knowledge to others

relating to social competences:

PEU_K01 – is aware of the importance of knowledge, including its critical analysis

PEU_K02 – is ready to deepen knowledge and skills, and, if necessary, use the help of experts

PROGRAMME CONTENT			
		Number of hours	
Se 1	Discussion of the diploma process in the field of study	1	
Se 2 - Se 15	Presenting a multimedia presentation and participating in the discussion	14	
	Total hours	15	

	TEACHING TOOLS USED	
N1. Presentation		
N2. Discussion		
N3. Consultations		ļ

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

` `	code	Way of evaluating learning outcomes achievement
		assessment based on the presentation and activity in discussions

PRIMARY AND SECONDARY LITERATURE
N/A
SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)
Chairman of the program committee for the relevant field of study
Card preparation:
Piotr Rutkowski, piotr.rutkowski@pwr.edu.pl

FACULTY of CHEMISTRY

SUBJECT CARD

Name of subject in Polish Metody spoektroskopowe w chemii medycznej Name of subject in English Spectroscopic methods in medicinal chemistry

Main field of study (if applicable): Biosciences Specialization (if applicable): Medicinal chemistry

Profile: academic

Level and form of studies: 2nd level,

Kind of subject: obligatory

Subject code W03BSS-SM2020W, W03BSS-SM2020L

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30		30		
Number of hours of total student workload (CNPS)	50		50		
Form of crediting (Examination / crediting with grade)	Ex		crediting		
For group of courses mark (X) final course					
Number of ECTS points	2		2		
including number of ECTS points for practical classes (P)			2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)			1,4		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. Knowledge of general chemistry.
- 2. Basic knowledge of physical chemistry.
- 3. Basic knowledge of organic chemistry.
- 4. Basic skills in the field of physicochemical and mathematical calculations.
- 5. Basic knowledge of spectroscopic techniques used in structural analysis.

SUBJECT OBJECTIVES

- C1 Acquiring knowledge about spectroscopic analysis methods
- C2 Acquiring knowledge about methods of interpreting one-dimensional magnetic resonance spectra.
- C3 Acquiring knowledge about methods of interpreting two-dimensional magnetic resonance spectra.
- C4 Acquiring knowledge on how to interpret FT-IR spectra and mass spectrometry spectra.
- C5 Practical knowledge of selected applications of mass spectrometry and magnetic resonance.

SUBJECT EDUCATIONAL EFFECTS

relating to knowledge:

PEU_W01 Has knowledge of spectroscopic analysis methods

PEU_W02 Has knowledge of the use of spectroscopy in analysis

PEU_W03 Has knowledge of the use of spectroscopic methods and medical diagnostics relating to skills:

PEU_U01 Is able to prepare material for analysis

PEU_U02 Is able to assess what methods to solve the current problem

PEU_U03 Is able to analyze spectroscopic data

relating to social competences:

Z zakresu kompetencji społecznych:

PEU_K01 Ma świadomość powiązań pomiędzy różnymi obszarami nauk chemicznych i/lub

technicznych oraz ich aspekty praktyczne.

PROGRAMME CONTENT Number Lecture of hours 2 Introduction in spectroscopic methods Lec 1 IR spectroscopy – theoretical background and applications 2 Lec 2 Lec 3 Raman spectroscopy - introduction 2 Raman spectroscopy - applications 2 Lec 4 2 Mass spectrommetry - introduction Lec 5 Mass spectrommetry – types of ionization 2 Lec 6 Mass spectrommetry - analizators Lec 7 2 Mass spectrommetry – fragmentation and interpretation of spectra Lec 8 UV-Vis and CD spectroscopy 2 Lec 9 2 Lec 10 NMR spectroscopy – theoretical background Lec 11 NMR spectroscopy – chemical shift Lec 12 NMR spectroscopy – coupling constant 2 Lec 13 2D NMR spectroscopy Lec 14 2D NMR spectroscopy Lec 15 EPR spectroscopy 30 Total hours Laboratory Number of hours Introduction in spectroscopic methods Lab 1 Lab 2 IR spectroscopy – interpretation of spectra 2 Lab 3 Raman spectroscopy Lab 4 2 Raman spectroscopy Lab 5 Mass spectrommetry – introduction 2

Lob 6	Mass spectrommetry – fragmentation	2
Lab o	Mass spectronimen y – fragmentation	Z
Lab 7	Mass spectrommetry - interpretation of spectra	2
Lab 8	Mass spectrommetry – interpretation of spectra	2
Lab 9	NMR spectroscopy – the principles	2
Lab 10	NMR spectroscopy – interpretation of 1D spectra	2
Lab 11	NMR spectroscopy – interpretation of 2D spectra	2
Lab 12	NMR spectroscopy – spectra simulations	2
Lab 13	NMR spectroscopy – spectra simulations	2
Lab 14	UV-Vis spectroscopy - applications	2
Lab 15	CD spectroscopy – interporetation and simulations	2
	Total hours	30

TEACHING TOOLS USED

- N1. Problem lectures multimedia presentations
- N2. Laboratory problematic issues (multimedia presentations)
- N3. Laboratory solving practical examples, drawing structures and spectra and performing calculations on a multimedia board
- N4. Own work preparation for partial tests
- N5. Own work consultations with the teacher

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F –	Learning outcomes code	Way of evaluating learning outcomes
forming during		achievement
semester), P –		
concluding (at semester		
end)		
P1 (lecture)	PEU_W01-W03,	examination
	PEU_K01	
F2 (laboratory)	PEU_UO1-UO2	test
D (1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	C 1	

P (1 laboratory) = arithmetic mean of test grades

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] K. Pigoń, Z. Ruziewicz, Chemia fizyczna t 2 Fizykochemia molekularna, Wyd. PWN, Warszawa 2007
- [2] P.W. Atkins, Chemia fizyczna, PWN 2001
- [3] R. M. Silverstein, F. X. Webster, D. J. Kiemle, Spektroskopowe metody identyfikacji związków organicznych PWN, Wraszawa 2007.
- [4] D. A. Skoog, D. M. West, F. J. Holler, S. R. Crouch, Podstawy chemii analitycznej 2, PWN, Warszawa 2007.
- [5] Z. Kęcki, Podstawy spektroskopii molekularnej, Wyd. PWN, Warszawa 1992.

SECONDARY LITERATURE:

- [1] A. Cygański, Metody spektroskopowe w chemii analitycznej. WNT Warszawa, 2009
- [2] J. Demichowicz-Pigoniowa, Chemia fizyczna t 3, Obliczenia fizykochemiczne, PWN, Warszawa 2010
- [3] J. Najbar, A. Turek, Fotochemia i spektroskopia optyczna, PWN, Warszawa 2009.

 [4] P. Suppan, Chemia i światło, PWN, Warszawa 1997. [5] W. Zieliński, A. Rajca, Metody spektroskopowe i ich zastosowanie do identyfikacji związków organicznych, WNT, Warszawa 2000
Związkow organicznych, wrwi, warszawa 2000
SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)
Prof Rafal Latajka, rafal.latajka@pwr.edu.pl

FACULTY OF CHEMISTRY

SUBJECT CARD

Name of subject in Polish Chemia Teoretyczna
Name of subject in English Theoretical Chemistry

Main field of study (if applicable): Biosciences

Specialization (if applicable):

Profile: academic

Level and form of studies: 2nd level, full-time

Kind of subject: obligatory

Subject code W03BSS-SM2001W, W03BSS-SM2001C, W03BSS-SM2001L

Group of courses NO

	Lecture	Classes	Laboratory	Project	Seminar
Number of hours of organized classes in University (ZZU)	30	15	30		
Number of hours of total student workload (CNPS)	100	50	50		
Form of crediting (Examination / crediting with grade)	Exam	crediting with grade	crediting with grade		
For group of courses mark (X) final course					
Number of ECTS points	3	2	2		
including number of ECTS points for practical classes (P))	2	2		
including number of ECTS points corresponding to classes that require direct participation of lecturers and other academics (BU)	1,3	0,7	1,4		

^{*}delete as not necessary

PREREQUISITES RELATING TO KNOWLEDGE, SKILLS AND OTHER COMPETENCES

- 1. General Chemistry and Physics
- 2. Linear algebra and mathematical analysis
- 3. Fundamentals of physical and quantum chemistry

SUBJECT OBJECTIVES

- C1 To acquaint students with fundamentals of molecular quantum mechanics.
- C2 To acquaint students with modern methods of theoretical description of the electronic structure of atoms and molecules and to acquire the ability to apply these methods to determine the electronic structure and properties of molecular systems.
- C3. Acquiring the ability to apply methods of theoretical chemistry to prediction and interpretation of selected spectral and thermodynamical properties of molecular systems.

SUBJECT EDUCATIONAL EFFECTS

related to knowledge:

Upon finishing the course, a student:

- PEU_W01 understands the problems and shortcomings of classical physics in the microscopic description,
- PEU_W02 knows the postulates of quantum mechanics and elements of the operator calculus,
- PEU W03 can write the Schrödinger equation (SE) for model systems and for any molecular system,
- PEU_W04 knows the solutions of SE for the free particle, particle in model systems and for the hydrogen atom; understands the interpretation of these solutions,

- PEU_W05 knows the basic approximations used in the description of the electronic structure of molecular systems: the Born-Oppenheimer approximation and the basics of the variational and perturbational methods,
- PEU_W06 knows the basics of the theory of molecular orbitals,
- PEU_W07 has a basic knowledge of the solutions of the Hartree-Fock and Hartree-Fock-Roothan equations,
- PEU_W08 has basic knowledge of the theory of electron correlation and methods of its calculation (configuration interaction method, many-body perturbation theory, density functional theory),
- PEU_W09 has a basic knowledge of the theory of intermolecular interactions.

related to skills:

Upon finishing the course, a student:

- PEU_U01 can plan, carry out and interpret the results of calculations of the electronic structure of molecules within HF approximation and using selected methods taking into account electron correlation,
- PEU_U02 can predict the equilibrium structure of molecules,
- PEU_U03 can predict and interpret spectra of electronic states of molecules,
- PEU U04 can interpret spectroscopic measurements based on quantum-chemical calculations,
- PEU_U05 can analyze the mechanisms of chemical reactions based on the results of quantum-chemical calculations.

related to social competences:

PEU_K01 student is ready to critically evaluate his/her knowledge and received content

PROGRAMME CONTENT		
	Lecture	Number of hours
Lec 1	Introduction to molecular quantum mechanics. Discussion of postulates of non-relativistic quantum mechanics. Definition of a wave function and its probabilistic interpretation. Definition of operators representing mechanical observables and elements of operator algebras. Time-dependent and time-independent Schrödinger's equation.	2
Lec 2	Free particle and particle in model potentials. Solving the Schrödinger equation for a free particle, particle in a box and in a harmonic potential.	2
Lec 3	Hydrogen atom. Solving the Schrödinger equation for a rigid rotator and hydrogen-like atoms.	2
Lec 4	Molecular Hamiltonian. Separation of the electronic and nuclear degrees of freedom. The adiabatic approximation and the Born-Oppenheimer approximation. The harmonic approximation. Normal modes analysis and interpretation of absorption spectra in the infrared range.	2
Lec 5	Approximate methods of solving the Schrödinger equation I. Variation calculus and its applications to model problems. Rayleigh-Ritz method. Molecular orbitals theory. Hückel method and its illustrative applications.	2
Lec 6	Approximate methods of solving the Schrödinger equation II. A time-independent perturbation theory. Perturbation in two-state and multi-state systems. Perturbation theory for degenerate reference states.	2
Lec 7	Wave functions for many-electron systems. Symmetry of the wave function. A determinantal wave function. The Slater-Condon rules. General expressions for matrix elements between Slater's determinants.	2
Lec 8	The Hartree-Fock method. The self-consistent field method. The Hartree-Fock-Roothan method. The charge density and matrix elements of the Fock operator.	2

Lec 9	Molecular orbitals. Elements of point group theory. Symmetry and nomenclature of molecular orbitals. Molecular orbitals diagrams for diatomic and polyatomic molecules. Walsh diagrams.	2
Lec 10	Electronic correlation I. Limitations of the Hartree-Fock method. Definition and methods for determining the electron correlation. The configuration interaction method.	2
Lec 11	Electronic correlation II. The Møller-Plesset perturbation theory. Elements of the coupled clusters method.	2
Lec 12	The density functional theory. One-particle density matrix and pair-density matrix. The Hohenberg-Kohn theorems. The Kohn-Sham method.	2
Lec 13	The interaction of matter with electromagnetic radiation. The fate of molecules in electronically excited states. Photochemical and photophysical processes in molecular systems. Jabłoński diagram. Absorption and fluorescence spectra in the UV and visible range. Fine structure of absorption and fluorescence spectra	2
Lec 14	Processes of nonradiative deactivation of excited states. Fermi's golden rule. Selection rules. Internal conversion. Conical intersections. Intersystem crossings. Excitation energy transfer - Förster's and Dexter's mechanisms. Natural and artificial light-harvesting systems. Photosynthesis.	2
Lec 15	Intermolecular interactions. The theory of intermolecular interactions. Hydrogen bond. Secondary structure of molecular systems, conformational analysis.	2
	Total hours	30
	Classes	Number of hours
Cl 1	Syllabus. Operator calculus . Elements of linear algebra. Examining the properties of operators, operator eigenproblem.	2
Cl 2	Solutions to the Schrödinger equation for model problems.	2
Cl 3	Simple applications of the variational principle to model problems.	2
Cl 4	Simple applications of the Rayleigh-Schrödinger perturbation theory to model problems.	2
Cl 5	Calculations of the electronic structure in the Hückel method for selected molecules I. The π -electronic approximation and basic assumptions of the Hückel method for unsaturated hydrocarbons. Eigenproblem solution. Determination of molecular orbital coefficients for simple molecules.	2
Cl 6	Calculations of the electronic structure in the Hückel model for selected molecules II. Own problem in matrix form. Hamiltonian diagonalization and interpretation of eigenvalue and eigenvector spectra. Bond density and order matrix and population analysis.	2
Cl 7	Hartree-Fock method I. Slater-Condon rules. Solving problems within the Hartree-Fock method.	2
Cl 8	Review and Test.	1
	Total hours	15
	Laboratory	Number of hours
Lab 1	Work organization in a computer lab and a computing center. Discussing the principles of health and safety at work. Distribution of accounts and basic information about available operating systems.	2
Lab 2	Elements of the LINUX system I. Basic information about the operating system. Selected BASH shell commands.	2

Lab 3	Elements of the LINUX system II. Support for selected text editors. Simple BASH shell scripts.	2
Lab 4	Selected electronic structure calculation packages. Preparation of batch files. Calculations of the electronic structure of atoms using the restricted and unrestricted Hartree-Fock method (HF). Structure of output files and interpretation of the results of calculations.	2
Lab 5	Representation of the structure of molecular systems. Orthogonal coordinates and internal coordinates on the example of Z-matrix.	2
Lab 6	Accuracy of computational chemistry methods. Selection of the basis functions. Comparison of the accuracy of selected ab initio methods and density functional theory methods. Validation of electronic structure calculation methods.	2
Lab 7	Optimization of equilibrium geometry of molecules and analysis of normal-mode vibrations. Discussion of gradient geometry optimization algorithms. Calculations of the harmonic frequencies' spectrum. Analysis of normal coordinates. Prediction and interpretation of infrared spectra.	2
Lab 8	Molecular orbital theory. Determination of potential energy curves for diatomic molecules in the HF method. Determination and interpretation of molecular orbital and Walsh diagrams. Charge-density population analysis.	2
Lab 9	Configuration interaction method. Calculation of electronic states' spectra using the configuration interaction method with single (CIS) and double excitations (CISD). Size-extensivity and size-consistency of the CI method. Project I. Calculations of the electronic states spectra and their interpretation for selected polyatomic molecules.	2
Lab 10	Project I. Calculations of the molecular structure and thermodynamical properties	2
Lab 11	Mechanisms of chemical reactions. Location of transition state geometry.	2
Lab 12	Project II – Calculations of electronic states spectra and their interpretation for selected polyatomic molecules.	2
Lab 13	Work on individual projects I.	2
Lab 14	Work on individual projects II.	2
Lab 15	Work on individual projects III.	2
	Total hours	30
	TEACHING TOOLS USED	
	ture at the blackboard	

- N2. Multimedia presentation
 N3. Implementation of tasks / projects in the computer lab
 N4. Personal computers / resources of the computing center / specialized software

EVALUATION OF SUBJECT LEARNING OUTCOMES ACHIEVEMENT

Evaluation (F – forming	Learning outcomes code	Way of evaluating learning outcomes achievement
during semester), P –		
concluding (at semester		
end)		
P	PEU_Lec01-	Final exam

	PEU_Lec15, PEU_K01	
F1	PEU_Cl01- PEU_Cl08, PEU_K01	Home assignments and test.
F2	PEU_La1- PEU_La15, PEU_K01	Individual projects
Р		

PRIMARY AND SECONDARY LITERATURE

PRIMARY LITERATURE:

- [1] Engel, T., Reid, P., Quantum Chemistry and Spectroscopy, 3rd ed. ed. Pearson, Boston, 2013
- [2] L. Piela, "Ideas of Quantum Chemistry" 3rd Edition, Elsevier, 2019
- [3] D. O. Hayward, "Quantum Mechanics for Chemists", RSC, 2002

SECONDARY LITERATURE:

[1] R. W. Góra, teaching materials for the course: "Theoretical chemistry", 2019

SUBJECT SUPERVISOR (NAME AND SURNAME, E-MAIL ADDRESS)

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